

# Rice increases phosphorus uptake in strongly sorbing soils by intra-root facilitation

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Upland rice (*Oryza sativa*) is adapted to strongly phosphorus (P) sorbing soils. The mechanisms underlying P acquisition, however, are not well understood, and models typically underestimate uptake. This complicates root ideotype development and trait-based selection for further improvement. We present a novel model, which correctly simulates the P uptake by a P-efficient rice genotype measured over 48 days of growth. The model represents root morphology at the local rhizosphere scale, including root hairs and fine S-type laterals. It simulates fast- and slowly reacting soil P and the P-solubilizing effect of root-induced pH changes in the soil. Simulations predict that the zone of pH changes and P solubilization around a root spreads further into the soil than the zone of P depletion. A root needs to place laterals outside its depletion- but inside its solubilization zone to maximize P uptake. S-type laterals, which are short but hairy, appear to be the key root structures to achieve that. Thus, thicker roots facilitate the P uptake by fine lateral roots. Uptake can be enhanced through longer root hairs and greater root length density but was less sensitive to total root length and root class proportions.

## KEYWORDS

Rhizosphere pH; slow sorption; coupled transport model; phosphorus uptake efficiency

# 1 | INTRODUCTION

2 Deficiency of phosphorus (P) is one of the main constraints to crop production on highly weathered soils of  
3 the humid tropics (Alewell et al., 2020). Owing to the high content of iron and aluminum oxides in the clay  
4 fractions of such soils, added P becomes strongly sorbed on soil surfaces and is therefore largely unavailable  
5 to plant roots. However, certain plant species – exemplified by upland rice (*Oryza sativa*) – are efficient in  
6 extracting this strongly sorbed P. There are large differences in tolerance of low P availability in the upland rice  
7 germplasm (Rose et al., 2013; Vandamme et al., 2016) and there has been progress in mapping the associated  
8 genes (Gamuyao et al., 2012; Schatz et al., 2014; Mori et al., 2016). However, the mechanisms involved are  
9 poorly understood (Nestler and Wissuwa, 2016; Wissuwa et al., 2020). Genotypes differ in total P uptake, P  
10 uptake per unit root surface area, and biomass per amount of P taken up – i. e. the internal P use efficiency  
11 (Wissuwa et al., 2015, 2020). Mathematical models are needed to describe the underlying mechanisms. In  
12 this paper, we develop such a model, focusing on the processes involved in P uptake.

13 Four factors potentially influence plant P uptake and need to be considered: (1) root length, diameter,  
14 architecture, and coverage with hairs, as these affect the surface area available for P uptake; (2) the P uptake  
15 per unit surface area of roots and hairs; (3) root-induced changes in the soil that affect the solubility of P; and  
16 (4) the effects of rhizosphere microbes and mycorrhizal fungi. Current understanding of these factors for  
17 upland rice is as follows.

18 First, root architecture and morphology. The upland rice root system comprises large numbers of roots  
19 which are classified into crown, seminal, L-type and S-type lateral roots. S-types are short and fine lateral  
20 roots (<1 cm length; <80  $\mu\text{m}$  diameter) unique to rice (Yamauchi et al., 1987) and typically double the total  
21 length of a root system (Kant et al., 2018; Wissuwa et al., 2020). The crown and lateral roots, including  
22 the fine S-types, produce hairs. Recent 3D modeling, allowing for soil transport, uptake limitations, S-type  
23 lateral roots, and hairs, shows that the P cost of those fine laterals is recovered within a day of their formation  
24 (Gonzalez et al., 2021). They are therefore efficient in the use of P.

25 Second, the P uptake per unit root surface area. Conventional models of P uptake treat individual roots as

26 'sinks' to which P is delivered by transport through the soil solution (Kuppe et al., 2021b). The sink strength is  
27 modeled as function of P concentration in solution at the root surface and the activity of P uptake transporters  
28 in the root membrane. Such models greatly under-predict P uptake by upland rice grown in strongly P-sorbing  
29 soils, both when root architecture is (Gonzalez et al., 2021; De Bauw et al., 2020) and is not considered (Kirk  
30 et al., 1999). In strongly sorbing soils, P concentrations in the soil solution can be so low that even high-affinity  
31 transporters may still be slow in taking it up. Rates of transport through the soil by diffusion are, however,  
32 even slower (note rates of mass flow in the transpiration stream are negligible at these P concentrations). The  
33 conventional models underestimate P uptake even when it is assumed that the root reduces the P concentration  
34 at its surface to zero. This suggests that other processes, not allowed for in the conventional models, are  
35 important.

36 Third, root-induced solubilization. Solubilization of P through the secretion of low molecular weight  
37 organic acid anions, such as citrate and malate, is widely discussed (Hinsinger et al., 2011; Oburger et al.,  
38 2009; Hoffland, 1992). For rice, Kirk et al. (1999) used a model developed by Kirk (1999) and calculated the  
39 rates of citrate secretion from roots required to explain the measured rates of P uptake. The P-solubilizing  
40 effect of citrate and the rate of citrate decomposition were determined experimentally. However, the measured  
41 effluxes of citrate and other P-solubilizing organic exudates from rice roots (Wissuwa, 2005) are much smaller  
42 than required. Likewise, there is no evidence for organic P solubilization by release of phosphate enzymes  
43 from upland rice roots (Hedley et al., 1994) or genotypic differences in phosphatase secretion (Rakotoson  
44 et al., 2020). We, therefore, exclude the role of organic acids or phosphatases in P uptake.

45 A further potential solubilization mechanism is by root-induced pH changes in the soil. Roots that take  
46 up nitrogen as nitrate ( $\text{NO}_3^-$ ) tend to take up an excess of anions over cations, balanced by the release of  
47 bicarbonate ( $\text{HCO}_3^-$ ) anions, so causing an increase in rhizosphere pH (Nye, 1981; Dijkshoorn et al., 1968).  
48 In soils with pH-dependent surface charge, an increase in pH tends to make the surface charge more negative  
49 and less P-sorbing (Barrow, 2017; Penn and Camberato, 2019). Hence, P becomes more plant available.

50 Fourth, mycorrhizal fungi and effects of the rhizosphere microbiome. Wissuwa et al. (2020) assessed  
51 the effects of mycorrhizal colonization on P uptake by P-efficient and -inefficient genotypes of upland rice in

52 strongly P-sorbing soil. Differences in mycorrhizal colonization – and expression of a mycorrhiza-induced P  
53 transporter (OsPT11) confirming that the symbiosis was functional – could not explain the genotypic variation  
54 in uptake per unit root surface area. Efficient genotypes always out-performed inefficient ones, independent of  
55 soil sterilization and inoculation treatments. This indicates that plant-specific factors rather than mycorrhizal  
56 fungi or other soil microbiome effects were responsible for the greater root efficiency. Though free-living  
57 microbes may solubilize P to some extent, there is little known to build a model, and any P solubilized at a  
58 distance from a root must get to a root surface.

59 To quantify the importance of these various factors for P acquisition by upland rice, we developed a  
60 new rhizosphere model and a method for up-scaling to the whole plant, and tested it against experimental  
61 data. This is the first published model to allow for interaction between root morphology and root-induced  
62 solubilization processes. We use it to ask (1) if solubilization by pH changes can quantitatively explain the  
63 measured P uptake of genotype DJ123; (2) how sensitive P uptake would be to variations in root and soil  
64 parameters, and thereby how root morphology contributes to P uptake; and (3) how much the individual root  
65 classes contribute to uptake, particularly the metabolically cheap S-types.

## 66 **2 | MATERIALS AND METHODS**

67 We simulate uptake of P and efflux of  $\text{HCO}_3^-$  over root surfaces based on diffusion-reaction equations radially  
68 symmetrical to the root axis (Figure 1a). The  $\text{HCO}_3^-$  balances excess intake of nutrient anions over cations  
69 due to plant nutrient demand. Reaction of  $\text{HCO}_3^-$  with the soil increases the soil pH and the concentration of  
70 P in the soil solution.

71 The total P in the system is the sum of P in the soil solution, P in the soil solid, and P taken up. The soil  
72 solid P is conceptually divided into pools that differ in their sorption and solubilization rates. Increasing pH  
73 leads to a faster desorption and a fast solubilization of P (Figure 1b).

74 We distinguish three root classes characteristic of rice: the crown roots, (long) L-type laterals, and

75 (short) S-type laterals, all with hairs (Figure 1a). We include seminal roots with the crown roots because the  
 76 distinction is experimentally difficult in older plants. The uptake by unit length of L-type laterals and crown  
 77 roots, both including their S-type lateral branches, are scaled-up to the whole root system using a fitted root  
 78 growth function (section 2.3). The S-type laterals are treated as sink and source terms in the rhizosphere of a  
 79 crown root and L-type lateral.

80 We assume P has a negligible influence on  $\text{HCO}_3^-$  because the amount of  $\text{HCO}_3^-$  reacting is much larger  
 81 than the amount of P in solution. Therefore, we first describe the equation for movement of  $\text{HCO}_3^-$  and  
 82 resulting pH changes because it is formulated independently of P. Secondly, we describe the equations for  
 83 pH-dependent P diffusion and reaction, including P uptake by root hairs, S-type laterals, and hairs of S-type  
 84 laterals. Thirdly, we describe the upscaling of the rhizosphere model to the total root system. Main symbols  
 85 are defined in Table 1.

86 [Figure 1 about here]

87 [Table 1 about here]

## 88 2.1 | $\text{HCO}_3^-$ efflux and diffusion

89 The pH increase, resulting from the reaction of  $\text{HCO}_3^-$ , is propagated away from the root surface by acid-base  
 90 transfer (Nye, 1981). The dominant acid-base pairs are  $\text{H}_3\text{O}^+ - \text{H}_2\text{O}$  and  $\text{H}_2\text{CO}_3 - \text{HCO}_3^-$ , the latter formed by  
 91 equilibration with  $\text{CO}_2$  in the soil air. The corresponding continuity equation for the concentration of titratable  
 92 soil acidity,  $HS$ , is (after Nye, 1981)

$$\frac{\partial HS}{\partial t} = \frac{1}{r} \frac{\partial}{\partial r} \left( r \theta f \left( D_H \frac{\partial H_\ell}{\partial r} - D_B \frac{\partial B_\ell}{\partial r} \right) \right) + E_R, \quad (1)$$

93 where  $H_\ell$ ,  $B_\ell$  are concentrations in soil solution,  $D_H$ ,  $D_B$  are diffusion constants in water for  $\text{H}_3\text{O}^+$  and  
 94  $\text{HCO}_3^-$ , respectively,  $\theta$  is the volumetric water content, and  $f$  is the diffusion impedance factor. The reaction  
 95 term  $E_R(r, t)$  is the efflux out of root hairs, S-types, and S-type hairs into the rhizosphere. The efflux of  
 96  $\text{HCO}_3^-$  over the whole root system balances the excess intake of anions ( $\text{NO}_3^-$ ,  $\text{H}_2\text{PO}_4^-$ ,  $\text{SO}_4^{2-}$ ) over cations  
 97 ( $\text{NH}_4^+$ ,  $\text{K}^+$ ,  $\text{Ca}^{2+}$ ,  $\text{Mg}_2^+$ ). We assume a constant efflux ( $E$ ) per unit root surface area over all root classes. We  
 98 express  $HS$  in terms of  $B_\ell$  as follows. Changes in  $HS$  are related to changes in pH by the soil pH buffer  
 99 power,  $b_{\text{HS}}$ , defined as  $b_{\text{HS}} = -dHS/d\text{pH}$ . From  $\text{pH} = -\ln H_\ell / \ln 10$  and  $dH_\ell/H_\ell = -dB_\ell/B_\ell$ , we get  
 100  $dHS = -b_{\text{HS}}dB_\ell/(B_\ell \ln 10)$ . Also  $dH_\ell = -K/B_\ell^2 dB_\ell$ , with  $B_\ell = K/H_\ell$ , and abbreviation defined as  
 101  $K = K_1 K_s p_{\text{CO}_2}$ , where  $K_1$  is the first apparent dissociation constant of  $\text{H}_2\text{CO}_3$ ,  $K_s$  is the solubility of  $\text{CO}_2$  in  
 102 water, and  $p_{\text{CO}_2}$  the  $\text{CO}_2$  pressure in the soil air. Thereby we obtain

$$\frac{\partial B_\ell}{\partial t} = \frac{B_\ell \ln 10}{b_{\text{HS}}} \left[ \frac{1}{r} \frac{\partial}{\partial r} \left( r \theta f \left( \frac{D_H K}{B_\ell^2} + D_B \right) \frac{\partial B_\ell}{\partial r} \right) + E_R \right]. \quad (2)$$

103 Note that  $b_{\text{HS}}/(B_\ell \ln 10)$  in Eqn 2 is the soil buffer power for  $\text{HCO}_3^-$ , which varies inversely with  $B_\ell$ .

104 The boundary condition at  $r = r_0$  is

$$\theta f \left( \frac{D_H K}{(B_\ell(r_0, t))^2} + D_B \right) \frac{\partial B_\ell(r_0, t)}{\partial r} = -E. \quad (3)$$

105 The zero-flux boundary condition at the rhizosphere outer radius,  $r = r_1$ , is

$$\frac{\partial B_\ell(r_1, t)}{\partial r} = 0. \quad (4)$$

106 We assume the initial pH throughout the soil around the root is constant at  $t = 0$ ; initial condition  $B_\ell(r, 0) =$   
 107  $B_{\ell, \text{init}}$ .

## 108 2.2 | P diffusion, reaction, and uptake

109 The rate of change in P concentration in the soil is

$$\frac{\partial P}{\partial t} = \theta \frac{\partial P_\ell}{\partial t} + \rho \left( \frac{\partial P_{s, \text{fast}}}{\partial t} + \frac{\partial P_{s, \text{fast-sol}}}{\partial t} + \frac{\partial P_{s, \text{slow}}}{\partial t} \right), \quad (5)$$

110 where  $P_\ell$  is the P concentration in the soil solution,  $P_{s, \text{fast}}$  is the concentration in rapid equilibrium with the  
 111 soil solution at the initial pH,  $P_{s, \text{fast-sol}}$  is the concentration of P rapidly solubilized as the pH is increased, and  
 112  $P_{s, \text{slow}}$  is the concentration in slow equilibrium. The relation between  $P_{s, \text{fast}}$  and  $P_\ell$  is defined by the soil P  
 113 buffer power  $b_p = \theta + \rho \left( \partial P_{s, \text{fast}} / \partial P_\ell \right)_{B_\ell}$ , and the additional P solubilized as the the pH increases is obtained  
 114 by introducing a P- $\text{HCO}_3^-$  interaction coefficient,  $\lambda = \left( \partial P_\ell / \partial B_\ell \right)_P$  (after Nye, 1983). Note,  $b_p$  and  $\lambda$  are  
 115 defined at constant concentrations of  $\text{HCO}_3^-$  and P in the whole soil, respectively. The equations for slow P  
 116 sorption and its dependence on  $P_\ell$  and  $B_\ell$  are given in section 2.2.1. Substitution and rearranging of Eqn 5  
 117 gives

$$\frac{\partial P}{\partial t} = b_p \left( \frac{\partial P_\ell}{\partial t} - \lambda \frac{\partial B_\ell}{\partial t} \right) + g \quad (6)$$

118 where  $g = \rho \partial P_{s, \text{slow}} / \partial t$  (see Supporting Information).

119 Now consider the change of P concentration around a crown root or L-type lateral as continuity equation

120 for P diffusion

$$\frac{\partial P}{\partial t} = \frac{1}{r} \frac{\partial}{\partial r} \left( r D_P \theta f \frac{\partial P_\ell}{\partial r} \right) - I_R \quad (7)$$

121 where  $I_R$  is the rate of uptake inside the rhizosphere domain by the S-type lateral surfaces ( $I_S$ ) and the hairs  
 122 on the S-types ( $I_{hS}$ ) and parent root ( $I_h$ ), i. e.  $I_R = I_h + I_S + I_{hS}$ . Combining Eqns 6 and 7 and rearranging  
 123 gives

$$\frac{\partial P_\ell}{\partial t} = \frac{1}{b_p} \left[ \frac{1}{r} \frac{\partial}{\partial r} \left( r D_P \theta f \frac{\partial P_\ell}{\partial r} \right) - I_R - g \right] + \lambda \frac{\partial B_\ell}{\partial t} \quad (8)$$

124 The influx into the crown root or L-type lateral at  $r = r_0$  is given by Michaelis-Menten kinetics with the  
 125 same  $V_{\max}$  and  $K_m$  for each root class. There is no flux of P across the boundary  $r = r_1$  between the zones of  
 126 influences of neighboring roots, and the initial P concentration is uniform throughout the soil. Hence,

$$D_P \theta f \frac{\partial P_\ell(r_0, t)}{\partial r} = \frac{V_{\max} P_\ell(r_0, t)}{K_m + P_\ell(r_0, t)}, \quad (9)$$

$$\frac{\partial P_\ell(r_1, t)}{\partial r} = 0, \quad (10)$$

$$P_\ell(r, 0) = P_{\ell, \text{init}}. \quad (11)$$

### 127 2.2.1 | Slow P sorption

128 In measurements of P desorption, there is generally an initial rapid release followed by a much slower release,  
 129 which may continue for days or longer (Barrow, 2008). We describe this slow reaction and its dependence on

130 pH as

$$\rho \frac{\partial P_{s,\text{slow}}}{\partial t} = \kappa_a \theta P_\ell - \kappa_d \rho P_{s,\text{slow}}, \quad (12)$$

131 where  $\kappa_a$  and  $\kappa_d$  are functions of  $B_\ell$ . Note, the notation for the  $r, t$ -dependency of variables is dropped to  
132 improve readability. We allow for the effect of pH on the slow sorption rates with

$$\kappa_a = k_1 - k_1^* (B_\ell - B_{\ell,\text{init}}), \quad (13)$$

$$\kappa_d = k_2 + k_2^* (B_\ell - B_{\ell,\text{init}}), \quad (14)$$

133 where  $k_1^*, k_2^*$  are additional adsorption and desorption rate constants. Hence, an increase in  $B_\ell$  results in  
134 increased P release from slowly reacting P.

### 135 2.2.2 | Root hairs and S-type laterals

136 The root hairs and S-type laterals are inside the rhizosphere of their parent roots (Figure 1a) and are therefore  
137 sink terms for P in Eqn 8 and source terms for  $\text{HCO}_3^-$  in Eqn 2. Starting with uptake by hairs and S-types,  
138 we calculate influx by assuming steady-state diffusion across their depletion zones following Baldwin et al.  
139 (1973), Bhat et al. (1976), Itoh and Barber (1983a) as justified by Leitner et al. (2010), and formulated for  
140 numerical discretizations by Kuppe et al. (2021a).

141 The equations for root hairs are as follows (S-types and their hairs are in the Supporting Information).

142 The rate of uptake by hairs per unit soil volume at distance  $r$  from the parent root is

$$I_h = \frac{V_{\max} P_{th}}{K_m + P_{th}} A_h, \quad (15)$$

143 where  $P_{th}$  is the concentration in solution at the root hair surface,  $A_h$  is the surface area of hairs per unit soil  
 144 volume, and  $V_{\max}$  and  $K_m$  are Michaelis-Menten constants, which, we assume, have the same values for root  
 145 and hairs. The value of  $A_h$  over an interval  $dr$  along the root hair length is given by (Kuppe et al., 2021a, Eqn  
 146 5)

$$A_h = \begin{cases} \frac{2r_h N_h dr}{(r+dr)^2 - r^2} & \text{for } r < l_h + r_0 \\ 0 & \text{for } r \geq l_h + r_0. \end{cases} \quad (16)$$

147 where  $r_h$  and  $l_h$  are the radius and length of root hairs, respectively, and  $N_h$  is the number of hairs per unit  
 148 hair length.

149 The mean P concentration in solution in the depletion zone around a hair is (after Baldwin et al., 1973,  
 150 Eqn iii)

$$\overline{P_\phi} = P_{th} + \frac{r_h \cdot V}{D_p \theta f} \left( \frac{r_{h1}^2}{r_{h1}^2 - r_h^2} \ln \left( \frac{r_{h1}}{r_h} \right) - \frac{1}{2} \right), \quad (17)$$

151 where the uptake  $V$  is a function of  $P_{th}$ . Substituting  $V(P_{th}) = V_{\max} P_{th} / (K_m + P_{th})$ , rearranging and solving

152 the resulting quadratic equation gives

$$P_{rh} = \frac{1}{2} \left( \overline{P}_\ell - K_m - \varphi + \sqrt{\varphi^2 + 2\varphi(K_m - \overline{P}_\ell) + (K_m + \overline{P}_\ell)^2} \right) \quad (18)$$

153 where  $\varphi = \frac{r_h V_{\max}}{D_p \theta f} \left( \frac{r_{h1}^2}{r_{h1}^2 - r_h^2} \ln \left( \frac{r_{h1}}{r_h} \right) - \frac{1}{2} \right)$ . This estimating of  $P_{rh}$  accounts for root hair depletion, remem-  
 154 bering  $P_{rh} < P_\ell$ . The mean concentration in the depletion zone around the hair ( $\overline{P}_\ell$ ) at distance  $r$  from the  
 155 parent root is equal to  $P_\ell$  in the numerical solution of Eqn 8. Hence,  $P_\ell$  in Eqn 18, the steady-state, is updated  
 156 at each time-point  $t$ . The mid-point between adjacent hairs at distance  $r$  is  $r_{h1} = \sqrt{r\pi/(2N_h)}$ .

157 The radius of S-type laterals is so small that we can treat the S-types as sink terms in Eqn 8 in the same  
 158 way as we do for root hairs. The corresponding equations are in the Supporting Information. The length of  
 159 S-types per unit length of crown root or L-type is

$$L_{SR} = \frac{\omega_S}{\omega_L + \omega_C}. \quad (19)$$

160 We assume that  $L_{SR}$  [ $\text{cm cm}^{-1}$ ] is constant over time and the S-type density is the same for crown roots and  
 161 L-type laterals. The actual S-type length in the rhizosphere is  $L_{S,act} = r_1 - r_0$ , which increases the number of  
 162 S-types per cm root according to  $L_{SR}$ .

163 Given the surface areas densities  $A_h$  [ $\text{cm}^2 \text{cm}^{-3}$ ], Eqn 16,  $A_{S,surf}$ , and  $A_{S,hairs}$  (see Supporting Informa-  
 164 tion), the reaction term in Eqn 2 for efflux into the rhizosphere is  $E_R(r, t) = -E(A_h + A_{S,surf} + A_{S,hairs})$ .

## 165 2.3 | Uptake rate per segment and upscaling to the whole root system

166 The cumulative uptake per unit root length of each class (L-type or crown root) at time  $t$ ,

$$U(t) = \int_0^t \dot{U}(s) ds,$$

167 is the sum of the uptake by the root and its hairs,

$$U = U_{\text{hairs}} + U_{\text{surf}} + U_{\text{S,surf}} + U_{\text{S,hairs}} \quad (20)$$

168 with corresponding uptake rates of P over time per unit root length, for root hairs

$$\dot{U}_{\text{hairs}}(t) = 2\pi r_h N_h \int_{r_0}^{r_0+l_h} \frac{V_{\text{max}} P_{\text{rh}}}{K_m + P_{\text{rh}}} dr \quad (21)$$

169 which we calculate by summing the volumetric  $J_h$  of Eqn 15 corrected to area, and the root surface

$$\dot{U}_{\text{surf}}(t) = 2\pi r_0 \frac{V_{\text{max}} P_{\ell}(r_0, t)}{K_m + P_{\ell}(r_0, t)}. \quad (22)$$

170 The rates of uptake by the surfaces and hairs of S-types are found with analogous equations.

171 We are interested in uptake by whole root systems, not just root segments. Scaling up to the whole root

172 system requires that we take the age distribution of the roots into account. Therefore, we define the time

173 derivative  $\dot{L}(t)$  of a function of the total root length  $L(t)$ . The total P uptake is

$$\mathcal{P}(t) = \int_0^t \dot{L}(s)U(t-s)ds, \quad (23)$$

174 where  $dL = \dot{L}(s)ds$ , here,  $t \leq t_{\text{end}} = 48$  d. The root system P uptake is the sum of uptake for each class  
175 (Supporting Information).

176 We assume that the crown and lateral roots experience the same initial soil conditions everywhere in the  
177 soil. For the default simulation, we assume proportions of root class length to be constant over time. For the  
178 sensitivity analysis, we allow for change in length of root classes and their proportions, following Wissuwa  
179 et al. (2020), however, constant over time.

## 180 **2.4 | Numerical solution and mass balance**

181 Eqns 2, 8, and 12 were solved subject to their initial and boundary conditions using a higher-order numerical  
182 method (details in Supporting Information). The relative mass balance error of P in the system was in the  
183 order of  $10^{-5}$ . The sum of the amount of P taken up by the plant and that remaining in the soil per unit root  
184 length at any time is given by

$$M_{\text{sum}} = M_{\ell} + M_{s,\text{fast}} + M_{s,\text{fast-soil}} + M_{s,\text{slow}} + U \quad (24)$$

185 where  $U$  is the plant uptake per unit root length (Eqn 20) and  $M_{\ell}$ ,  $M_{s,\text{fast}}$ ,  $M_{s,\text{fast-soil}}$ ,  $M_{s,\text{slow}}$  are the amounts of  
186 P in the soil volume influenced by the root per unit root length at time  $t$ , subscripted as for the P concentrations  
187 in the soil (Eqn 8).

## 188 **2.5 | Plant parameters**

### 189 **2.5.1 | Measurements**

190 Wissuwa et al. (2020) measured time courses of root growth and P uptake by rice genotype DJ123 (which is  
191 considered efficient in P uptake, Mori et al., 2016) grown on a strongly P-sorbing volcanic ash soil with low  
192 available P (Andosol) in Tsukuba, Japan. Plants were grown in a field under aerobic upland conditions or  
193 in large 72-L boxes filled with the low-P Andosol from that same field. Entire root systems were collected  
194 at 7, 14, 21, and 28 days after emergence (DAE) from the box, and at 16, 28, and 48 DAE from the field.  
195 Root system excavation in the field was limited to a depth of 30 cm, which typically contains >90% of entire  
196 root biomass in rice (Mori et al., 2016). Roots were scanned and analyzed using the WinRHIZO software,  
197 with diameter settings of 0–0.009 cm, 0.009–0.025 cm and 0.025–0.1 cm for S-type, L-type, and crown roots,  
198 respectively. By 28 DAE the roots in boxes were too large to be scanned and total root length was estimated  
199 based on root dry weight and conversion factors determined from previous scans. Shoot samples were taken  
200 at the same times as roots, and P concentrations in shoot and root were determined to calculate the quantity of  
201 P taken up. The box experiment was repeated in June 2019.

202 Nestler and Wissuwa (2016) measured the length and densities of the root classes and their coverage  
203 with hairs in genotype DJ123 under the conditions as the Wissuwa et al. (2020) field experiment (Table 2).  
204 Plants were grown in the field and, at 50 days after seeding, a 1 m deep trench was dug perpendicular to the  
205 plant rows across the field to sample the roots. Intact roots were removed by washing and stored at 4 °C in  
206 50% ethanol prior to root hair evaluation. Using the longest crown root of a plant, 1 cm long sections at 1, 10,  
207 15, 20, and 25 cm from the root tip were imaged to determine root hair length and density on the crown root  
208 and its laterals as described by Nestler et al. (2016).

## 209 2.5.2 | Root length, S-type length, and RLD

210 To the measured root length, we fitted the function

$$L(t) = \frac{a \exp(t)}{(c + \exp(t))^b} \quad (25)$$

211 with parameters  $a$ ,  $b$ , and  $c$  using least square means, as shown in Figure 2a. This function is able to fit the  
212 experimentally observed growth jump after day 7.

213 Half the total length of the root system was S-type laterals ( $\omega_S = 0.5$ , Table 2), and we assume an equal  
214 proportion of S-types for crown roots and L-type laterals over time. Therefore, the S-type length per segment  
215 from Eqn 19 is  $L_{SR} = 1 \text{ cm cm}^{-1}$  for both classes of parent root.

216 The root length density (RLD) is calculated by dividing the total root system length of roots ( $2.5 \times$   
217  $10^4 \text{ cm plant}^{-1}$ ) by a nominal rooted soil volume in the experiments based on the plant spacing ( $10 \text{ cm} \times$   
218  $15 \text{ cm plant}^{-1}$ ) and rooting depth (30 cm), giving  $\text{RLD} = 5.55 \text{ cm cm}^{-3}$ . The mean radial mid-point between  
219 adjacent crown roots or L-type laterals is therefore

$$r_1 = \sqrt{\frac{1 + L_{SR}}{\text{RLD} \cdot \pi}} = 0.34 \text{ cm}. \quad (26)$$

## 220 2.5.3 | Root P uptake kinetics and $\text{HCO}_3^-$ efflux

221 We set the values of the Michaelis-Menten coefficients for P uptake such that they are not limiting, i.e.  
222 the concentration in solution at the root surfaces was reduced to near zero and uptake was not sensitive to  
223 increases in  $V_{\max}$  or decreases in  $K_m$ . The values for all root classes were  $V_{\max} = 8 \times 10^{-12} \text{ mol cm}^{-2} \text{ s}^{-1}$   
224 and  $K_m = 1 \times 10^{-9} \text{ mol cm}^{-3}$ .

225 Given that the P concentrations in soil solution are very low, the influx depends on the initial slope of  
226 the function  $V(P_\ell) = V_{\max} P_\ell / (K_m + P_\ell)$ . For  $K_m \gg P_\ell$  or  $P_\ell \rightarrow 0$ , the ratio of  $V_{\max}$  and  $K_m$  gives,  
227 approximately, the relevant slope. We discuss root P uptake properties further in Section 4.3.

228 In principle, the  $\text{HCO}_3^-$  efflux per unit root surface area,  $E$ , may vary with differences in cation and anion  
229 intake over the root system. However, to avoid undue complexity, we used a constant value across the root  
230 system based on the measured plant growth and nutrient intakes. The upscaled  $\text{HCO}_3^-$  efflux (as described in  
231 section 2.3 for uptake) fitted the data well ( $R^2 = 0.93$ ). Consider the increase in shoot and root dry weight  
232 of approximately  $4.5 \text{ g plant}^{-1}$  and the mean root surface area of  $1.25 \times 10^4 \text{ cm}^2$  between 28 and 48 DAE  
233 (Wissuwa et al., 2020). We used the following typical nutrient contents for rice during vegetative growth  
234 (Dobermann and Fairhurst, 2000): N, K, Ca, Mg, S were 30, 20, 3, 2, 2  $\text{mg g}^{-1}$ , respectively; and measured P  
235 content of  $1 \text{ mg g}^{-1}$ . Assuming all the N was taken up as  $\text{NO}_3^-$ , which is reasonable because nitrification is  
236 rapid in this soil under the conditions of the experiment (Nardi et al., 2013), the uptake of cations ( $\text{K}^+$ ,  $\text{Ca}^{2+}$ ,  
237  $\text{Mg}^{2+}$ ) was  $3.96 \text{ mol plant}^{-1}$  and that of anions ( $\text{NO}_3^-$ ,  $\text{H}_2\text{PO}_4^-$ ,  $\text{SO}_4^{2-}$ ) was  $10.66 \text{ mmol plant}^{-1}$ , which is an  
238 anion excess of  $6.7 \text{ mmol plant}^{-1}$ . Dividing this by the time-integral of root surface area gives the efflux rate  
239  $E = 1.83 \times 10^{-12} \text{ mol cm}^{-2} \text{ s}^{-1}$ .

240 [Table 2 about here]

## 241 2.6 | Soil parameters

### 242 2.6.1 | Measurements

243 The basic soil properties were clay = 17 %, silt = 43 %, organic C = 4.7 % and pH (1:5  $\text{H}_2\text{O}$ ) = 5.8. From the  
244 experimental set up:  $\rho = 0.87 \text{ g cm}^{-3}$  and  $\theta = 0.3$ . The diffusion impedance factor at the given bulk density  
245 and moisture content,  $f = 0.24$ , was measured by following the diffusion of a trace amount of non-adsorbed  
246  $\text{Br}^-$  applied on one end of a block of the soil by the method of Darmovzalova et al. (2020).

247 We measured the initial P concentration in the soil solution,  $P_{\ell, \text{init}}$ , as follows. Portions (120 g) of

248 air-dry soil were moistened to 50 % of field capacity with deionized water and incubated in sealed containers  
249 at approximately 20 °C for 21 d. The moist soil was then centrifuged and the supernatant filtered through  
250 0.45 µm filters (Whatman). Filtered supernatant (approximately 30 cm<sup>3</sup>) was shaken with a DGT device (DGT  
251 Research Ltd, UK) for 24 h, and the P adsorbed on the DGT ferrihydrite gel was desorbed into 1 cm<sup>3</sup> of  
252 1 M HCl for 24 h, and measured colorimetrically.

253 We estimated the other soil P fractions using the Hedley et al. (1982) sequential fractionation scheme  
254 (Table 3). Samples (1.0 g) of oven-dry soil were extracted sequentially for 16 h each with (1) a strip of  
255 HCO<sub>3</sub><sup>-</sup>-form anion exchange resin membrane (adsorption capacity 26 µmol per strip) in H<sub>2</sub>O, recovering P  
256 from the resin by extraction in 0.5 M HCl; (2) 30 cm<sup>3</sup> of 0.5 M NaHCO<sub>3</sub>, followed by digestion in H<sub>2</sub>SO<sub>4</sub> and  
257 H<sub>2</sub>O<sub>2</sub>; (3) 30 cm<sup>3</sup> of 0.1 M NaOH, followed by digestion of the extract in H<sub>2</sub>SO<sub>4</sub> and H<sub>2</sub>O<sub>2</sub>; and (d) 30 cm<sup>3</sup> of  
258 1 M HCl; and finally (4) by digestion of the residual soil in H<sub>2</sub>SO<sub>4</sub> and H<sub>2</sub>O<sub>2</sub>. Total P in the NaHCO<sub>3</sub> and  
259 NaOH extracts was obtained from P in the digested extract and organic P (P<sub>o</sub>) was inferred from the total P  
260 less the inorganic P (P<sub>i</sub>) measured in the undigested extracts. P concentrations in the extracts were measured  
261 colorimetrically by the molybdenum blue method (Murphy and Riley, 1962).

262 The pH buffer capacity (soil pH buffer power),  $b_{\text{HS}}$ , was measured by shaking 2 g portions of the soil in  
263 10 cm<sup>3</sup> of 10 mM CaCl<sub>2</sub> overnight with sufficient NaOH to raise the final pH by approximately 1 pH-unit. This  
264 gives  $b_{\text{HS}} = 1 \times 10^{-5} \text{ mol cm}^{-3} (\text{pH unit})^{-1}$  (see also Nye, 1981). We take a typical value for CO<sub>2</sub> pressure in  
265 the soil air,  $p_{\text{CO}_2} = 4 \times 10^{-3} \text{ atm}$  (Figure 1 in Nye, 1981).

266 [Table 2 about here]

267 **2.6.2 | Soil P pools**

268 Following Hedley et al. (1994), who measured depletion of NaOH-P<sub>i</sub> by upland rice, we define the total  
 269 concentration of P that is ultimately available for uptake by a root as

$$P = \theta P_{\ell} + \rho ([\text{Resin-P}] + [\text{NaHCO}_3\text{-P}_i] + [\text{NaOH-P}_i]) \quad (27)$$

270 We assume [Resin-P] and [NaHCO<sub>3</sub>-P<sub>i</sub>] are in rapid equilibrium with the soil solution, and an additional  
 271 proportion  $\alpha$  of [NaOH-P<sub>i</sub>] is rapidly released if the pH is raised by one unit. Hence, the initial values of  
 272  $P_{s,\text{fast}}$  and  $P_{s,\text{fast-sol}}$  are

$$P_{s,\text{fast}} = [\text{Resin-P}] + [\text{NaHCO}_3\text{-P}_i] \quad (28)$$

$$P_{s,\text{fast-sol}} = \alpha [\text{NaOH-P}_i] \quad (29)$$

273 The component of [NaOH-P<sub>i</sub>] that slowly equilibrates at the initial pH is

$$P_{s,\text{slow}} = (1 - \alpha) [\text{NaOH-P}_i]. \quad (30)$$

274 If the soil pH is raised by one unit and allowed to equilibrate, the new mass balance of P is

$$P = b_p P_{\ell}^* + \rho P_{s,\text{slow}}^* \quad (31)$$

275 where

$$\beta = \frac{P_{s,slow}^*}{P_{s,slow}} \quad (32)$$

276 and  $P_{s,slow}^*$ ,  $P_{\ell}^*$  are the equilibrium concentrations at  $\text{pH} = \text{pH}_{\text{initial}} + 1$ . The coefficients  $\alpha$  and  $\beta$  are both  
 277 less than one and pre-set. We used values  $\alpha = 0.33$  and  $\beta = 0.4$ . Substituting the measured Hedley fractions  
 278 (Table 3) and  $P_{\ell} = 8 \times 10^{-11} \text{ mol cm}^{-3}$  at the initial pH in Eqn 27 and 31 gives  $P_{\ell}^* = 1.4 \times 10^{-9} \text{ mol cm}^{-3}$ .

### 279 2.6.3 | P soil buffer power and P-HCO<sub>3</sub><sup>-</sup> interaction coefficient

280 We estimate the soil buffer power,  $b_p = (\partial P_e / \partial P_{\ell})_{B_{\ell}}$ , from  $b_p = \theta + \rho (P_{s,fast} / P_{\ell})_{\text{initial}}$ , note  $\theta \ll b_p$  and  
 281  $P_e = \theta P_{\ell} + \rho (P_{s,fast} + P_{s,fast-sol})$ ,  $\partial P_{s,fast-sol} / \partial P_{\ell} = 0$ . Substituting  $P_{\ell, \text{init}} = 0.08 \mu\text{M}$ ,  $P_{s,fast} = 0.5 \mu\text{mol g}^{-1}$ ,  
 282 and  $\rho = 0.87 \text{ g cm}^{-3}$  gives  $b_p \approx 5440$ .

283 We estimate the P-HCO<sub>3</sub><sup>-</sup> interaction coefficient,  $\lambda$ , as follows. If  $\Delta P_e$  is the change required to leave  $P_{\ell}$   
 284 unchanged following unit increase in pH above the initial pH, then (following Nye, 1983):

$$\Delta P_{\ell} = \left( \frac{\partial P_{\ell}}{\partial P_e} \right)_{B_{\ell}} \Delta P_e + \left( \frac{\partial P_{\ell}}{\partial B_{\ell}} \right)_P \Delta B_{\ell}^* = 0. \quad (33)$$

285 Substituting  $b_p = (\partial P_e / \partial P_{\ell})_{B_{\ell}}$  and  $\lambda = (\partial P_{\ell} / \partial B_{\ell})_P$  and rearranging gives

$$\lambda = \frac{\rho P_{s,fast-sol}}{b_p \Delta B_{\ell}^*} \quad (34)$$

286 where  $\Delta B_\ell^*$  is the change in  $B_\ell$  for unit increase in pH above the initial value. This gives  $\lambda = 1.49 \times 10^{-3}$ .

## 287 2.6.4 | Adsorption-desorption rates

288 The half-times of slow equilibration, obtained by integrating Eqn 12 at constant  $P_\ell$ , are

$$t_{1/2} = \frac{\ln 2}{k_2} \quad (35)$$

$$t_{1/2}^* = \frac{\ln 2}{k_2 + k_2^* \Delta B_\ell^*}. \quad (36)$$

289 The half-times for strongly P-sorbing soils are in the order of weeks (Barrow, 2008). Hence, we obtain  
290  $k_2 = 2.87 \times 10^{-7} \text{ s}^{-1}$  and  $k_2^* = 9.31 \text{ cm}^3 \text{ mol}^{-1} \text{ s}^{-1}$  for trial values  $t_{1/2} = 28 \text{ d}$ ,  $t_{1/2}^* = 2 \text{ d}$  after unit pH change,  
291 respectively. We then compute  $k_1$  and  $k_1^*$  from the following equations. At equilibrium,  $\partial P_{s,\text{slow}}/\partial t = 0$ , and  
292 so from Eqns 12–14,

$$k_1 \theta P_{\ell,\text{init}} = k_2 \rho P_{s,\text{slow}} \quad (37)$$

$$(k_1 - k_1^* \Delta B_\ell^*) \theta P_\ell^* = (k_2 + k_2^* \Delta B_\ell^*) \rho \beta P_{s,\text{slow}}, \quad (38)$$

293 where  $\Delta B_\ell^*$  is the change in  $B_\ell$  for unit increase in pH above the initial value. Substituting the above values  
294 in these equations gives  $k_1 = 7.86 \times 10^{-2} \text{ s}^{-1}$  and  $k_1^* = 1.34 \times 10^5 \text{ cm}^3 \text{ mol}^{-1} \text{ s}^{-1}$ .

295 [Table 3 about here]

## 296 2.7 | Simulations and sensitivity analysis

297 Uptake from the rhizospheres of unit length crown roots and L-type laterals are simulated over a 48 d growth  
298 period and upscaled with the method described in section 2.3. The default parameters, Table 4, are used as a  
299 reference parameter set in the sensitivity analysis. The parameters are varied relative to their default value  
300 over a four-fold change.

### 301 **Scaling of the rates for efflux**

302 We assume the same total amount of  $\text{HCO}_3^-$  released for varying root morphological traits. The amount  
303 of  $\text{HCO}_3^-$  released depends on the surface area of the root system and the efflux integrated over time. For  
304 parameters that change the root surface area, we scale the efflux per unit root surface area ( $E$ ), so that the  
305  $\text{HCO}_3^-$  released by the whole root system remains unchanged (Supporting Information, Figure S1: upscaled  
306 segment efflux). Also, the efflux,  $E$ , is varied in a range that does not increase pH more than one unit locally  
307 (as the model is limited by  $\Delta B_\ell = 4 \times 10^{-7} \text{ mol cm}^{-3}$ , which is approx.  $\Delta\text{pH} = 1.06$ ). Here, the variation  
308 of efflux with fixed default root size can be interpreted as a change in total nutrient uptake, i. e. the balance  
309 between cation and anion uptake.

### 310 **Distances to neighboring roots**

311 Changing the outer radius,  $r_1$ , from half to double, other parameters fixed (in particular  $L_{\text{SR}} = 1$ ), the RLD  
312 changes in a range from 1.39 to 22.2  $\text{cm cm}^{-3}$  (eight-fold for RLD). Placing main root segments closer together  
313 by reducing the outer radius also decreases the distances among S-type laterals. The mid-distance to the  
314 neighboring root,  $r_1$ , does not influence the efflux, but the pH change can drastically increase when the distance  
315 to the zero-flux outer boundary decreases. Thus  $\text{HCO}_3^-$  accumulates when the volume of the rhizosphere  
316 decreases and the RLD increases. The variation of the outer radius (measured from the center of the root) is  
317 limited to unit pH change and, thereby, uptake was not simulated for  $r_1 < 0.3386$  for crown roots.

### 318 **Root length**

319 S-type length is changed via  $L_{SR}$  (Eqn 19, change in  $\omega_S$  fixed  $\omega_L$  and  $\omega_C$ ) and with this, RLD changes (max.  
320 8.33 for  $2 \cdot L_{SR}$ ). Note that  $L_{SR}$  determines the S-type number on a segment. The root length of crown roots  
321 and L-types is varied while keeping the default one cm S-type length per cm parent root and the number of  
322 root hairs per surface root area unchanged. Hence, a change in the length  $L_C$ , and  $L_L$  implies a change in  
323 total S-type length.

### 324 **S-type proportion**

325 We vary  $\omega_S$  in Eqn 19 in order to vary  $L_{SR}$  while keeping either the root system length, the root surface area,  
326 or the root volume fixed. Note,  $L_{SR}$  changes  $A_{S,surf}$  and  $A_{S,hairs}$ . We assume a fixed RLD and a fixed ratio of  
327 L-type by crown root proportion,  $\omega_L/\omega_C$ .

### 328 **Solubilization fractions and rates**

329 The total P in soil is kept constant for all simulations. We vary  $\alpha$ , which partitioned the measured NaOH-P<sub>i</sub>  
330 over  $P_{s,slow}$  and  $P_{s,fast-soil}$ . We also vary the ad- and desorption rates and their responsiveness to the pH changes  
331 by varying the half-times of slow equilibration,  $t_{1/2}$ ,  $t_{1/2}^*$ , for pH 5.8, 6.86, respectively, and  $\beta$ , Eqn 32. The  
332 half-times determine the desorption rates,  $k_2$ ,  $k_2^*$ , whereas  $\beta$  determines the pH effect on the adsorption rate  
333 through a change in  $k_1^*$ .

## 334 **3 | RESULTS**

### 335 **3.1 | Whole root system P uptake**

336 Using the default parameter set (Table 4), we found that only 10% of the observed uptake could be explained  
337 without allowing for solubilization, slow sorption, and S-types (Figure 2b). Including the S-type laterals and  
338 their root hairs as additional root-morphological features increased uptake significantly (2.6-times more than

339 without S-type laterals), but not sufficiently, the uptake was 26.4 % of that of the full model with solubilization,  
340 which fits the data. Allowing all of the NaOH-P<sub>i</sub> to contribute to slow sorption without solubilization (i. e.  
341 making  $\alpha$  in Eqns 29 and 30 equal to zero) increased P uptake, but only to 41 % of that with solubilization.

342 To simulate the measured uptake without solubilization and without slow sorption required either a  
343 3.6-fold increase in  $P_{\ell,init}$  or a 224-fold increase in  $P_{s,fast}$  (Supporting Information, Figure S2). However, such  
344 a fast pool would lead to a soil-solution equilibrium constant, an order too high even when compared to whole  
345 inorganic solid phase P:  $1.4 \times 10^6 \gg (P_{s,fast}(r, 0) + \text{NaOH-P}_i) / P_{\ell,init}$ . Neither of these is plausible within the  
346 uncertainty of the parameter estimates.

### 347 3.2 | Importance of different root classes

348 Including hairs on respective root classes, crown roots took up 48.8 % of the total P compared to 30.6 % for  
349 S-type laterals and 20.6 % for L-type laterals (Figure 2c). The root hairs on S-types, L-types, and crown roots  
350 in sum were responsible for almost 77 % of the total uptake (black dashed line), where those on crown root  
351 and S-types mattered most with a combined uptake of approximately 64 % of the total. The results seem  
352 surprising, as it is generally thought that most P is taken up by lateral roots. This is still the case, but the lateral  
353 roots are here split over two classes: S- and L-types. To understand these results for the root system, we have  
354 to take a look at the cumulative uptake per unit length of crown and L-type lateral roots (Figure 3a vs. 3b).

355 [Figure 2 about here]

356 Cumulative uptake of P per unit root length by a crown root was 3.1-times that of an L-type lateral. This  
357 can be understood by considering that a crown root has three times the surface area per unit length with about  
358 double the number of root hairs. For crown roots, most of the uptake was achieved by the hairs of the crown  
359 root itself, whereas in the rhizosphere of L-type laterals, most of the uptake was by the root hairs on their  
360 S-type laterals. Very little was taken up over the S-type root surface. Hence, the root hair contribution to  
361 uptake increased when the main root radius decreased.

362 In models without solubilization, the P concentration is slowly depleted and the cumulative uptake levels  
363 off or becomes approximately linear since it only depends on the uptake from a buffered P solution. In  
364 our simulations, however, more P dissolved and the P uptake by S-type hairs increased exponentially over  
365 time (compare convex shape of dashed blue line versus concave shape of dashed cyan line in Figure 3a).  
366 Consequently, the cumulative uptake did not level off. The uptake rate even accelerated somewhat at later  
367 time points for S-type hairs while it slowed down slightly for crown root hairs. S-types by themselves did not  
368 achieve the necessary pH change (data not shown, separate uptake of S-types shown in Supporting Information,  
369 Figure S3), just like L-type laterals achieved a smaller pH change in their rhizosphere compared to crown  
370 roots. P uptake by S-type laterals depended on the efflux zone of the crown and L-type lateral roots, thereby  
371 on the parent root and on the proximity to other roots.

372 [Figure 3 about here]

### 373 3.3 | Rhizosphere P concentration profiles

374 Of importance to P-uptake is the concentration in the soil solution,  $P_\ell$ , which is very small, but its dynamics  
375 follows  $P_{s,fast}$  (Figures 3c and 3d). The spatial-temporal variation in  $P_\ell$  (Figures 4a and 4b) shows that  $P_\ell$  was  
376 quickly depleted close to the parent root surface. This depletion zone, however, was rather narrow ( $< 0.5$  mm).  
377 For Figure 4, note that the root center is located at  $r = 0$  and the root surface at  $r = r_0$  (for crown roots  
378  $r_0 = 0.03$  and for L-type laterals  $r_0 = 0.01$  cm, Table 2).  $P_\ell$  was greatly increased one mm away from the  
379 root surface. The increase of up to ten times the initial  $P_\ell$  concentration seems dramatic, but the resulting  
380  $0.8 \mu\text{M}$  P is still a very low concentration, normally considered highly growth limiting (Tinker and Nye, 2000).  
381 The root hairs increase plant P uptake most when placed where the  $P_\ell$  increase is the greatest. The measured  
382 length of the hairs on the crown and lateral roots were not large enough to reach that zone (Table 2), but the  
383 hairs on the S-type laterals were placed across the rhizosphere.

384 The root and the sorbing soil pools competed for P in solution, as shown in the P balances in Figures 3c  
385 and 3d. Note that the uptake plotted in Figures 3a and 3b formed only a very small fraction of the total

386 P concentration in the crown- and L-type lateral rhizospheres. The pools of  $P_{s,slow}$  (red line) and  $P_{s,fast-sol}$   
387 (purple line) were depleted at a much greater rate than the rate at which P was taken up because the pool of  
388  $P_{s,fast}$  (green line) adsorbed a large fraction of the released P. The movement between P fractions was largely  
389 driven by the pH change.

390 Crown roots achieved a greater pH change in the rhizosphere: at the root surface up to +1.05 pH units  
391 and at the outer boundary up to +0.76 units (Figures 4c and 4d). The greater pH change in the rhizosphere of  
392 crown roots was associated with their larger surface area (i. e. diameter and number of root hairs).

393 [Figure 4 about here]

### 394 **3.4 | Sensitivity analysis**

395 We analyzed the sensitivity of the P uptake to a relative change in the default parameters (Figures 5 and 6,  
396 note the log-log scale). Sensitive parameters have steep slopes and need to be determined accurately. They  
397 also present opportunities for increasing P uptake if the parameter values can be changed through breeding or  
398 agricultural management.

#### 399 **3.4.1 | Plant parameters**

400 The radius  $r_1$  is the radial distance to the mid-point between neighboring roots in the radial model. Reducing  $r_1$   
401 (which corresponds with an increase in RLD) increased the local pH change and the solubilization, especially  
402 in the case of crown roots. Hence, crown roots closer together take up more P. We excluded simulations with  
403 a change in pH of more than one unit as this would be outside the validity of the parameter ranges. Therefore,  
404  $r_1$  could be reduced by 40 % to approximately  $r_1 = 0.2$  cm for the rhizosphere of L-type segments, obtaining  
405 ten times the initial  $P_e$  locally (or here maximally by 45 % to  $r_1 = 0.186$  cm with 17.5 times  $P_{e,init}$ ). The  
406 sensitivity to  $r_1$  thus demonstrates that intra-root facilitation increases with increasing RLD (Figure 5b).

407 Among the root parameters, morphological traits influence P uptake most (root hair length  $l_h$ , root hair

408 number  $N_h$ , root hair radius  $r_h$ , and S-type length  $L_{SR}$ , Figures 5a and 5b). Uptake by the root system was  
409 highly sensitive to root hair morphology ( $l_h$ ,  $N_h$ , and  $r_h$ ), with root hair length and number equally important  
410 (Figure 6). Uptake by L-type laterals was more sensitive than uptake by crown roots concerning change in  
411 root hair number, root hair radius, and S-type length (Figures 5b and 5a). The results suggest that there is  
412 more potential to increase the benefit of L-types. However, these sensitivity differences diminish at the whole  
413 root system level, because the contribution of uptake by L-type laterals to total uptake was less than that by  
414 thicker crown roots (see section 3.4.3).

415 Besides root morphology, the root-induced  $\text{HCO}_3^-$  efflux was very important in the model. The sensitivity  
416 to the efflux was in the range of the root morphological parameters for L-types (pink line in Figures 5a and  
417 5b). Additional S-types in the rhizosphere were beneficial (brown line).

### 418 3.4.2 | Soil parameters

419 We recap that  $\alpha$  denotes the fraction of  $\text{NaOH-P}_i$ , which dissolves when the pH increases, and  $\beta$  the ratio of  
420 the slow pools in steady-state at initial pH and unit pH increase. The parameters  $\alpha$  and  $\beta$  are estimated, not  
421 measured. However, the simulated uptake was not sensitive to them (Figures 5c and 5d). Insensitivity to  $\alpha$   
422 means that the pools of  $P_{s,\text{fast-sol}}$  and  $P_{s,\text{slow}}$  can replace each other. For example, the extreme case  $\alpha = 1$   
423 (no slow sorption) could compensate for the lack of a slow pH-dependent sorption (Supporting Information,  
424 Figure S2).

425 For the variation in slow sorption, we varied the reaction half-times with and without unit pH change ( $t_{1/2}$   
426 and  $t_{1/2}^*$  in Eqns 35 and 36). The uptake increased when the starting half-time ( $t_{1/2}$ ) of the slow pool increased or  
427 when the end half-time ( $t_{1/2}^*$ ) decreased. A smaller  $t_{1/2}^*$  led to greater depletion of the slow pool. The accelerated  
428 release in response to increased pH (i. e. the discrepancy between  $t_{1/2}$  and  $t_{1/2}^*$ ) influenced uptake strongly.  
429 Hence,  $t_{1/2}$  and  $t_{1/2}^*$  should be determined carefully to obtain accurate predictions of uptake. For example, an  
430 accelerated slow pH-dependent desorption could compensate for the lack of rapid solubilization (Supporting  
431 Information, Figure S2). Decreasing  $t_{1/2}^*$  to 1 d resulted in approximately 130 % of the reference uptake (while

432  $\alpha = 0$ ), and decreasing  $t_{1/2}$  to 14 d with  $t_{1/2}^* = 2$  d resulted in approximately 90 % of the reference uptake  
433 ( $\alpha = 0$ ).

434 We found that the fast and slow sorbing P pools (i. e.  $P_{s,fast}$  and  $P_{s,slow}$ ) compete for the P in solution, and  
435 the extent of this competition depends on the relative rates of the fast and slow adsorption-desorption reactions.  
436 Therefore, we specified a relatively large pH-dependent desorption rate to overcome the competition of slow  
437 and fast P pools by depletion of the slow P pool.

### 438 3.4.3 | Whole root system uptake

439 The total P uptake by the root system was less sensitive to the lengths of S-types, L-types, and crown roots  
440 ( $L_{SR}$ ,  $L_L$ , and  $L_C$ ) than to the root hair morphology (Figure 6). While the root length increased, the released  
441  $\text{HCO}_3^-$  per segment decreased, and thereby the uptake. Hence, the resulting sensitivity was less than the  
442 1:1-line. The sensitivity of uptake to  $L_C$  was greater than that to  $L_L$  and about the same as to  $L_{SR}$  in the  
443 rhizospheres of both parent roots. Note that the change in  $L_C$ , and  $L_L$  implied change in total S-type length.

444 Changing the relative proportions of S-type to L-type to crown root (volume, surface area, or length)  
445 did not affect uptake strongly (Figure 6). Increasing the proportion of the length of S-type laterals ( $\omega_S$ )  
446 relative to the other two root classes affected uptake the least and slightly negative (solid black line, condition:  
447  $\omega_S + \omega_L + \omega_C = 1$ ). Because an increase in S-type length at the cost of L-types ( $\omega_L$ ) and crown roots ( $\omega_C$ )  
448 reduces the total surface area and volume of the whole root system. The direct effect of smaller surface area on  
449 P uptake is moderated by the effect on  $\text{HCO}_3^-$  efflux, which is increased since the total  $\text{HCO}_3^-$  release per plant  
450 is fixed. Usually, trade-offs for investment into root classes are presented on a metabolic equivalence basis,  
451 either in terms of carbon or P. We do not know these costs in our system, but we may compare root system  
452 compositions with varying S-type lengths at a fixed total root surface area (dashed black line in Figure 6) or  
453 volume (gray line). Note that keeping surface area or volume fixed, an increase in  $\omega_S$  increases the total root  
454 length ( $\omega_S + \omega_L + \omega_C \neq 1$ ). Also, we kept the RLD fixed by varying  $r_1$ . An increase in S-type proportion but  
455 fixed total root surface area or volume resulted in only slightly more P uptake. Hence, the investment between

456 S-types and thicker root classes seems balanced in DJ123.

457 [Figure 5 about here]

458 [Figure 6 about here]

## 459 **4 | DISCUSSION**

### 460 **4.1 | The importance of solubilization and its interaction with root morphology**

461 Conventional models of P uptake in which roots are treated as sinks for P, but not otherwise influencing the  
462 solubility of P in the soil, work reasonably well for soils with large P concentrations in the soil solution. But  
463 for strongly P-sorbing soils, such as highly weathered soils of the humid tropics and volcanic ash soils, such  
464 models tend to greatly underestimate uptake. Hence, the model without solubilization could only account  
465 for 41 % of the observed uptake (Figure 2b). The full model allows for solubilization of P and interactions  
466 between solubilization and root morphology. The full model successfully predicts the total P uptake by  
467 the P-efficient upland rice genotype DJ123 growing in a strongly sorbing soil using input parameter values  
468 measured independently.

469 As P is removed from the soil by root uptake, a zone of depletion develops around the root (Figure 4).  
470 The depletion zone is narrow because P diffusion in the soil is very slow due to the strong sorption. However,  
471 simultaneously the root releases  $\text{HCO}_3^-$  ions into the soil to balance higher anion than cation intake, and  
472 the reaction of  $\text{HCO}_3^-$  with the soil tends to increase the P concentration in solution. In our simulations,  
473  $\text{HCO}_3^-$  diffused faster than P and, consequently, a zone developed in which the P concentration in solution  
474 rose above that in the soil bulk. Thus peaks in P concentration in solution formed at approximately 0.1 cm  
475 from the crown root and 0.06 cm from the L-type (Figures 4a and 4b). Phosphate, therefore, diffuses from the  
476 solubilization zone, albeit very slowly (Supporting Information, Figure S4). This is where the interaction with  
477 fine S-type lateral roots branching off the parent root becomes important because the laterals and their hairs

478 may intercept solubilized P that would otherwise not be available to the plant. The increase in P uptake due to  
479 solubilization is thereby reinforced.

## 480 **4.2 | Root-induced pH changes**

481 Efflux of  $\text{HCO}_3^-$  from the roots depends on the excess influx of anions over cations, which depends on the rate  
482 of increase in total plant biomass, the nutrient composition of the plant, the root-shoot ratio, and root surface  
483 area. That means, for a fixed root surface area, the total  $\text{HCO}_3^-$  release increases as the rate of plant growth  
484 increases (Figures 5a and 5b, pink lines). Hence, a genotype with better internal P use efficiency (greater  
485 growth rate per unit root surface) can have better P uptake efficiency due to an increased  $\text{HCO}_3^-$  efflux and P  
486 solubilization.

487 We estimated  $\text{HCO}_3^-$  efflux from the total biomass with a typical nutrient composition of rice and used  
488 a constant efflux per unit root surface area. Therefore, crown roots induced greater pH change than L-type  
489 laterals (Figure 4). For a given efflux, root-induced pH changes will be greater with more compact root  
490 systems. Hence, at greater root length density, the distance between neighboring roots is smaller, and therefore  
491 the  $\text{HCO}_3^-$  accumulates in a smaller volume of soil, resulting in greater solubilization and uptake. However,  
492 this effect will be offset if the uptake of other nutrients, and hence the  $\text{HCO}_3^-$  efflux, is impaired at the greater  
493 rooting density.

494 Soil factors also influence the pH change away from the roots but were not studied in detail. The initial  
495 soil pH, the  $\text{CO}_2$  pressure in the soil air, and the soil pH buffer power (Eqn 2) will therefore also influence P  
496 solubilization and uptake to some extent.

## 497 **4.3 | Root P uptake kinetics**

498 We assigned values for the root P uptake parameters such that they did not limit uptake, and there was no  
499 minimum concentration below which uptake ceased. In reality, there must be some minimum concentration

500 for P uptake transporters to operate. Rice plants can reduce solution P concentrations to 16 nM (Mori et al.,  
501 2016), however, reliable minimum concentrations are rarely documented (Griffiths and York, 2020). If the  
502 P transporter activity limited uptake to some minimum concentration, additional solubilization would be  
503 required to account for the observed rates of P uptake. There is a need to measure genotypic variation in root  
504 uptake parameters in the sub-micro-molar range as it may explain (among other traits) why some genotypes  
505 have lower uptake per unit root length than more P-efficient genotypes.

#### 506 **4.4 | Contributions of different root classes to uptake**

507 S-type laterals increase net P uptake dramatically because they are positioned inside the solubilization zones  
508 but outside the P depletion zones of L-types and crown roots. Although the S-types have a small diameter, their  
509 effective surface area for P uptake is greatly increased by their hairs. They are therefore efficient in intercepting  
510 P solubilized by their parent root. Over time, as the zone of pH-change and P solubilization around the parent  
511 root spreads further into the soil, the importance of S-types 'reaching out' increases. Treating S-type laterals  
512 independent of parent roots reduced their contribution to total P-uptake from 30 % to 11 % (Supplemental  
513 Material, Figure S3). Likewise, using a model that did not explicitly allow for solubilization, Gonzalez et al.  
514 (2021) found that S-types contributed little to total P uptake by the root system. Nonetheless, they showed that  
515 the low P cost of forming S-type laterals can be recovered by their P uptake within a day, much faster than  
516 other root types. With a model allowing for solubilization, this period would be even shorter. This facilitation  
517 of P uptake among the root classes suggests that there needs to be a balanced investment into the different root  
518 classes to optimize uptake.

519 Crown roots contribute less to the total root length than other root types but much to the total root surface  
520 area because of their larger diameter (Table 2). That large surface area furthermore allows more root hairs  
521 to grow per unit root length. This explains why crown root hairs contribute most to total plant P uptake. In  
522 addition to crown root uptake, their role as a parental root in enhancing P uptake of S-type laterals is larger  
523 than for parental L-type laterals due to the greater pH change and P solubilization they cause. We conclude

524 that crown roots have an important role in P uptake, and this may explain why genotypes with many crown  
525 roots have high performance under low P, despite crown roots being metabolically expensive (Wissuwa et al.,  
526 2020).

527 The total uptake of the whole root system was not sensitive to changes in the partitioning of S-type length  
528 (Figure 6, gray and brown line). This suggests that the P-efficient genotype DJ123 achieves a good balance  
529 among the root classes.

#### 530 **4.5 | Influence of root hair morphology on uptake**

531 In the model, root hairs compete for P in the soil solution with the perpendicular root surface, and most  
532 of the P is taken up by hairs (Figure 3). Overall, the simulated uptake is strongly increased in response to  
533 increased root hair length, number, and diameter (Figures 5 and 6). Zygalkakis et al. (2011) found – with a  
534 model allowing for root hairs but not solubilization or S-type laterals – that hair density had a much smaller  
535 effect on the uptake than length. This was because the hairs rapidly depleted P in the soil volume between  
536 them so that increasing the hair density had little effect. In our simulations, most hairs on S-type laterals  
537 are outside of the depletion zone of the parent root but within the zone where solubilization increases the P  
538 concentration in the soil solution. Therefore more densely spaced hairs had a similar impact to longer hairs.

539 Nestler and Wissuwa (2016) found that differences among upland rice genotypes were mostly due to  
540 the ability to increase hair length in response to P deficiency, and all the measured hair parameters showed  
541 consistent variation across root types. However, compared to the length of root hairs reported for other  
542 graminaceous species (Marzec et al., 2015; Itoh and Barber, 1983b; Hill et al., 2010), rice root hairs seem  
543 short and those on the parent roots do not reach as far as the peak of solubilized P concentration. Presumably,  
544 the P returns per unit P invested are greater for hairy S-types (length  $\leq 1$  cm) than for longer hairs alone.

545 While less important than root hair length and density, we found hair radius did influence uptake (Figures 5  
546 and 6). We are not aware of any published data on variation in root hair diameter in rice, but two-fold variation  
547 can be seen in images by Kim et al. (2007). Data from arabidopsis showed variation among genotypes

548 exceeding variations within a given genotype (Parker et al., 2000), suggesting that variations in root hair  
549 diameter should be explored as a potential way to increase P uptake.

#### 550 **4.6 | Time dynamics of whole root system P uptake**

551 The upscaled P uptake curve follows the exponential increase in measured P uptake quite nicely (Figure 2).  
552 Our simulations, however, overestimate the uptake during earlier time points and are below the measured  
553 average at the last time point. The shape of the P uptake curve for the whole root system is largely determined  
554 by the shape of the root growth curve with little influence of temporal uptake dynamics per unit root segment  
555 (Figures 3a and 3b). In upscaling, we account for the cumulative uptake by younger and older roots. Since the  
556 root system is a combination of older and younger roots at any time, the time dynamics of solubilization at  
557 the rhizosphere scale averages out at the whole root system scale. It proved, therefore, difficult to improve  
558 the fit of the whole root system P uptake curve by changing the model parameters. Possibly, root length was  
559 underestimated in the field experiment, where collecting whole root systems is challenging. Also, additional  
560 processes may have contributed in the field, such as an increase in the contribution of mycorrhizal fungi.

#### 561 **4.7 | Trade-offs and feedbacks at the whole plant level**

562 At the whole plant level, there may be feedbacks that enhance growth under P-limited conditions causing  
563 small rhizosphere effects to have an increasing impact over time (Wissuwa, 2003). Greater P uptake is likely  
564 to increase growth and also the uptake of other nutrients resulting in increased  $\text{HCO}_3^-$  efflux, solubilization,  
565 and eventually P uptake. This positive feedback would benefit from greater internal P use efficiency. It  
566 would also benefit from an optimal investment into the different root classes and root hairs to capture the  
567 solubilized P. We concluded earlier that DJ123 seems balanced in this respect. However, the uptake of other  
568 nutrients, particularly nitrogen, also influences the relative investment into root classes. The best root system  
569 for capturing  $\text{NO}_3^-$  may have long crown roots exploiting the whole soil volume (Dathe et al., 2016). Such

570 a diffuse and deep root system is likely to be less effective for P uptake. But since P uptake depends on  
571 solubilization,  $\text{NO}_3^-$  uptake is also important. Therefore there may be trade-offs for optimizing root architecture  
572 and morphology. Further model studies are needed to explore this.

## 573 **5 | CONCLUSIONS**

574 Our mechanistic model successfully simulates P uptake by the P-efficient genotype DJ123 growing on strongly  
575 sorbing soil. The model allows the interchange of P between the soil solid and solution through fast and slow,  
576 pH-dependent reactions and for the P-solubilizing effect of root-induced pH changes. The modeled root system  
577 contains three root classes, all hairy: crown roots, L-type laterals, and S-type laterals. The simulated uptake is  
578 a result of an important interaction between solubilization and the morphology of the upland rice root system.  
579 In the absence of solubilization or S-type laterals, the model could not account for the measured uptake. P  
580 uptake is less sensitive to total root length and root class proportions. But S-types greatly enhance overall P  
581 uptake because they extend across the solubilization zone around the parent and beyond its P depletion zone.  
582 Hence, S-type laterals, longer root hairs, and greater root length density can improve uptake. However, these  
583 S-types are particularly beneficial in the solubilization zone. Solubilization over time is greater around thicker  
584 roots and when root length density is high. The greater surface area that comes with thicker roots increases  
585 the release of  $\text{HCO}_3^-$ , which relates to the plant nutrient composition and relative growth rates. Root hairs and  
586 S-types do not achieve strong solubilization themselves, but their P uptake is facilitated by the near presence  
587 of thicker roots, especially nodal roots. We thus found a pronounced facilitative interaction between very fine  
588 laterals and their thicker parent root and this needs to be considered in formulating target traits for selecting  
589 P-efficient rice cultivars.

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## 592 DATA AVAILABILITY

593 Contact the corresponding author for experimental data or source code.

## 594 CONFLICT OF INTEREST

595 The authors declare that they have no competing interests.

## 596 AUTHOR'S CONTRIBUTIONS

597 The root data and model challenge was presented by MW. The model was developed by CK, GK, and JP. Code  
598 by CK. GK provided soil data, MW the plant data. Figures made by CK. CK wrote the manuscript with input  
599 from all authors. All authors approved the final version of the manuscript.

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**TABLE 1** List of main symbols

Symbol	Definition	Unit
$P$	concentration of P in the whole soil	$\text{mol cm}^{-3}$ (soil)
$P_\ell$	concentration of P in soil solution	$\text{mol cm}^{-3}$ (solution)
$P_{s,\text{fast}}$	concentration of P in rapid equilibrium with the soil solution at the initial pH	$\text{mol g}^{-1}$ (soil)
$P_{s,\text{fast-sol}}$	concentration of P rapidly solubilized as the pH is increased	$\text{mol g}^{-1}$ (soil)
$P_{s,\text{slow}}$	concentration of P in slow equilibrium with the solution	$\text{mol g}^{-1}$ (soil)
$B_\ell$	concentration of $\text{HCO}_3^-$ in soil solution	$\text{mol cm}^{-3}$ (solution)
$H_\ell$	concentration of $\text{H}_3\text{O}^+$ in soil solution	$\text{mol cm}^{-3}$ (solution)
$b_p$	soil buffer power at initial $B_\ell$ , defined as $\theta + \rho (\partial P_{s,\text{fast}} / \partial B_\ell)_{B_\ell}$	$\text{cm}^3$ (solution) $\text{cm}^{-3}$ (soil)
$b_{\text{HS}}$	soil pH buffer power, defined as $-\partial HS / \partial \text{pH}$ where $HS$ is the concentration of titratable acidity in the soil	$\text{mol cm}^{-3}$ (soil) (pH unit) $^{-1}$
$\lambda$	soil P- $\text{HCO}_3^-$ interaction coefficient, defined as $(\partial P_\ell / \partial B_\ell)_P$	
$D_H, D_B, D_P$	diffusion coefficient in free solution for $\text{H}_3\text{O}^+$ , $\text{HCO}_3^-$ , and $\text{H}_2\text{PO}_4^-$ , respectively	$\text{cm}^2 \text{s}^{-1}$
$\theta$	soil volumetric water content	$\text{cm}^3$ (solution) $\text{cm}^{-3}$ (soil)
$\rho$	soil bulk density	$\text{g cm}^{-3}$ (soil)
$f$	diffusion impedance factor	
$r_0$	root radius	cm
$r_1$	outer radius (from root center)	cm
$I_R$	uptake rate inside the rhizosphere including root hairs $I_h$ , S-type laterals $I_s$ , and S-type hairs $I_{\text{hS}}$	$\text{mol cm}^{-3}$ (soil) $\text{s}^{-1}$
$E_R$	efflux into the rhizosphere including root hairs $E_h$ , S-type laterals $E_s$ , and S-type hairs $E_{\text{hS}}$	$\text{mol cm}^{-3}$ (soil) $\text{s}^{-1}$
$E$	efflux of base $\text{HCO}_3^-$	$\text{mol cm}^{-2}$ (root) $\text{s}^{-1}$
$t$	time dimension	d
$r$	space dimension (symmetrical radial coordinates)	cm

**TABLE 2** S-type, L-type, and crown root default parameters of rice genotype DJ123

Parameter		S-type	L-type	crown roots	Unit	Source
Root radius	$r_0$	$r_s = 0.0025$	0.01	0.03	cm	WinRhizo and ImageJ plus Micrograph
Outer radius	$r_1$	$r_{h1,surf}$	0.3386	0.3386	cm	outer boundary derived from RLD, Eqn 26
Number of root hairs per unit segment on: $2r_0\pi L$ , $L=1$ cm	$N_h$	$N_{hS} = 520$ (420–620) <sup>a</sup>	700 (600–825) <sup>a</sup>	1440 (1360–1520) <sup>a</sup>	cm <sup>-1</sup>	Nestler and Wissuwa (2016)
Root hair length	$l_h$	$l_{hS} = 0.0125$ (0.01–0.0155) <sup>a</sup>	0.014 (0.012–0.017) <sup>a</sup>	0.02 (0.017–0.0225) <sup>a</sup>	cm	Nestler and Wissuwa (2016)
Root hair radius	$r_h$	0.0008	0.0008	0.0008	cm	Nestler et al. (2016), diameter 10–20 $\mu$ m
Length proportion <sup>b</sup>	$\omega$	0.5	0.33	0.17	fraction	Wissuwa et al. (2020)
Surface area proportion		0.21	0.31	0.48	fraction	(including hairs)

a) Values in brackets are 'field' to 'box' data, we used a value within that range. b) here a 3:2:1-ratio as default.

**TABLE 3** Soil P fractions measured in the experimental soil measured by the Hedley et al. (1982) sequential fractionation scheme

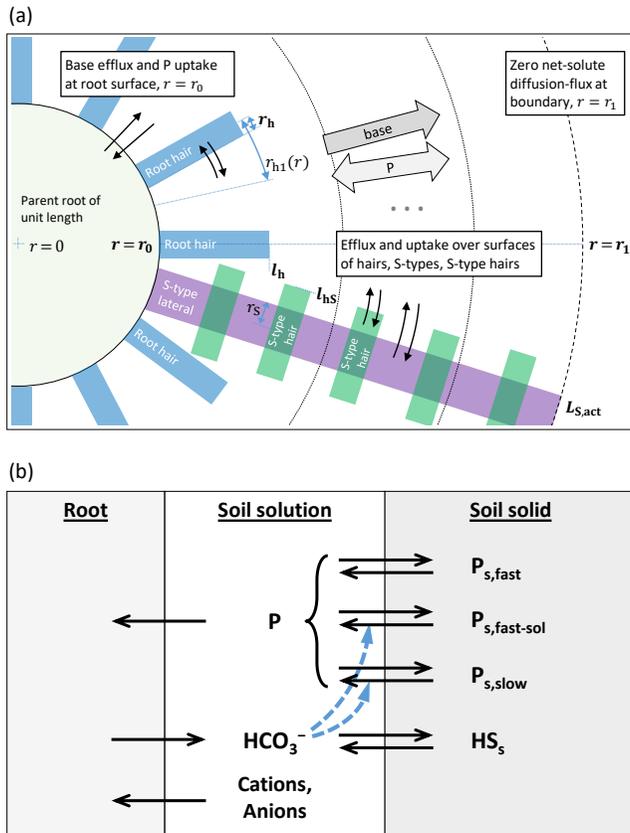
Extraction step	Fraction	Content (mmol kg <sup>-1</sup> )
1 g soil with 26 μmol HCO <sub>3</sub> <sup>-</sup> -form anion exchange resin in H <sub>2</sub> O, recovering P from the resin in 0.5 M HCl	Resin-P	0.04
30 cm <sup>3</sup> of 0.5 M NaHCO <sub>3</sub> , followed by digestion of the extract in H <sub>2</sub> SO <sub>4</sub> and H <sub>2</sub> O <sub>2</sub>	NaHCO <sub>3</sub> -P <sub>i</sub>	0.46
P <sub>o</sub> inferred from total P less P <sub>i</sub>	NaHCO <sub>3</sub> -P <sub>o</sub>	0.71
30 cm <sup>3</sup> of 0.1 M NaOH, followed by digestion of the extract in H <sub>2</sub> SO <sub>4</sub> and H <sub>2</sub> O <sub>2</sub>	NaOH-P <sub>i</sub>	11.29
P <sub>o</sub> inferred from total P less P <sub>i</sub>	NaOH-P <sub>o</sub>	5.72
30 cm <sup>3</sup> of 1 M HCl	HCl-P	0.74
digestion of the residual soil in H <sub>2</sub> SO <sub>4</sub> and H <sub>2</sub> O <sub>2</sub>	Residual-P	21.06
	Total	40.02

P<sub>i</sub>: inorganic, P<sub>o</sub>: organic

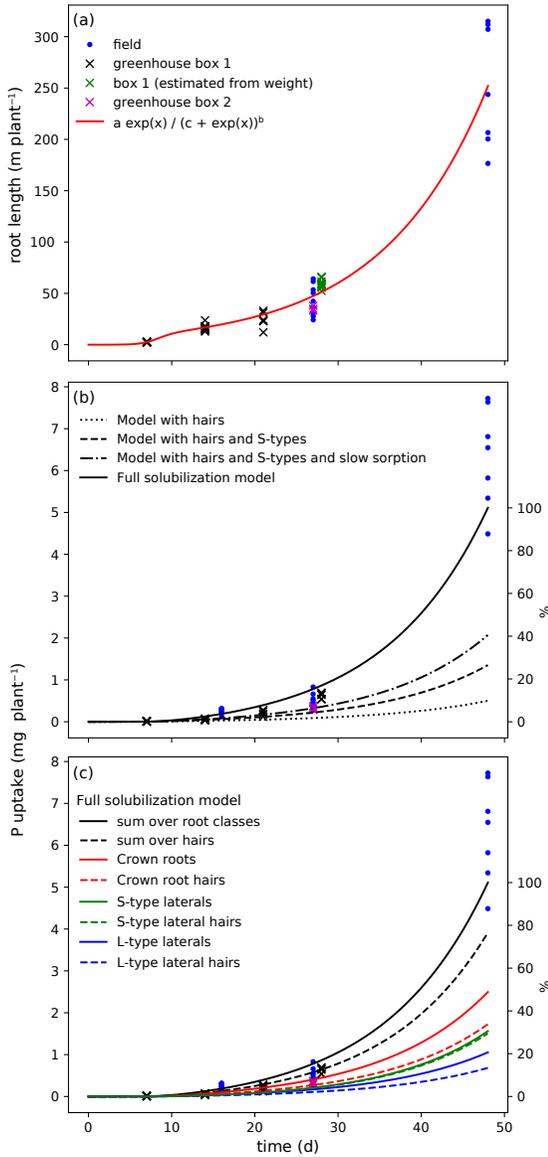
**TABLE 4** Parameters and intermediates for default simulation.

Parameter	Value	Unit	Comments
$V_{\max}$	$8 \times 10^{-12}$	$\text{mol cm}^{-2} \text{s}^{-1}$	maximal P uptake rate (assumed to be fast)
$K_m$	$1 \times 10^{-9}$	$\text{mol cm}^{-3}$	concentration at which P uptake rate is $\frac{1}{2} V_{\max}$ (assumed to be low)
$\rho$	0.87	$\text{g cm}^{-3}$	bulk density for Andosol
$\theta$	0.3	$\text{cm}^3 \text{cm}^{-3}$	volumetric water content
$f$	0.24		diffusion impedance factor for Andosol
$D_P$	$8.9 \times 10^{-6}$	$\text{cm}^2 \text{s}^{-1}$	$\text{H}_2\text{PO}_4^-$ diffusion coefficient in water
$D_B$	$1.23 \times 10^{-5}$	$\text{cm}^2 \text{s}^{-1}$	$\text{HCO}_3^-$ diffusion coefficient in water
$D_H$	$9.55 \times 10^{-5}$	$\text{cm}^2 \text{s}^{-1}$	$\text{H}_3\text{O}^+$ diffusion coefficient in water
$K_1$	$4.45 \times 10^{-10}$	$\text{mol cm}^{-3}$	apparent first dissociation constant of $\text{H}_2\text{CO}_3$
$K_s$	$3.39 \times 10^{-5}$	$\text{mol cm}^{-3} \text{atm}^{-1}$	solubility of $\text{CO}_2$ in water
$p_{\text{CO}_2}$	$4 \times 10^{-3}$	atm	$\text{CO}_2$ pressure in soil air
$\alpha$	0.33		fraction of $[\text{NaOH-P}_i]$ to calculate $\lambda$
$\beta$	0.4		relative change in $P_{s,\text{slow}}$ ( $t = 0$ ) in steady-state after pH change
$\lambda$	$1.49 \times 10^{-3}$		soil $P_e$ - $B_e$ interaction coefficient, $dP_e/dB_e$ at constant $P_e$
$t_{1/2}$	28	d	half-time of slow sorption reaction at pH of 5.8
$t_{1/2}^*$	2	d	half-time of slow sorption reaction at pH of 6.861
$k_1$	$7.86 \times 10^{-2}$	$\text{s}^{-1}$	reaction constant in Eqn 13
$k_2$	$2.87 \times 10^{-7}$	$\text{s}^{-1}$	reaction constant in Eqn 14, half-time of 28 d
$k_1^*$	$1.34 \times 10^5$	$\text{cm}^3 \text{mol}^{-1} \text{s}^{-1}$	reaction constant in Eqn 13
$k_2^*$	9.31	$\text{cm}^3 \text{mol}^{-1} \text{s}^{-1}$	reaction constant in Eqn 14
$b_p$	5440		soil buffer power of P, $b_p = \theta + \rho P_{s,\text{fast}}/P_e$ , at $t = 0$
$b_{\text{HS}}$	$1 \times 10^{-5}$	$\text{mol cm}^{-3} (\text{pH unit})^{-1}$	pH buffer capacity
$E$	$1.83 \times 10^{-12}$	$\text{mol cm}^{-2} \text{s}^{-1}$	efflux flux of $\text{HCO}_3^-$ across root surface
$P_{e,\text{init}}$	$8 \times 10^{-11}$	$\text{mol cm}^{-3}$	initial P in solution
$\text{pH}_{\text{initial}}$	5.8		initial pH (1:5 $\text{H}_2\text{O}$ ) measured for Andosol
$\Delta B_e$	$4 \times 10^{-7}$	$\text{mol cm}^{-3}$	see Eqn 34

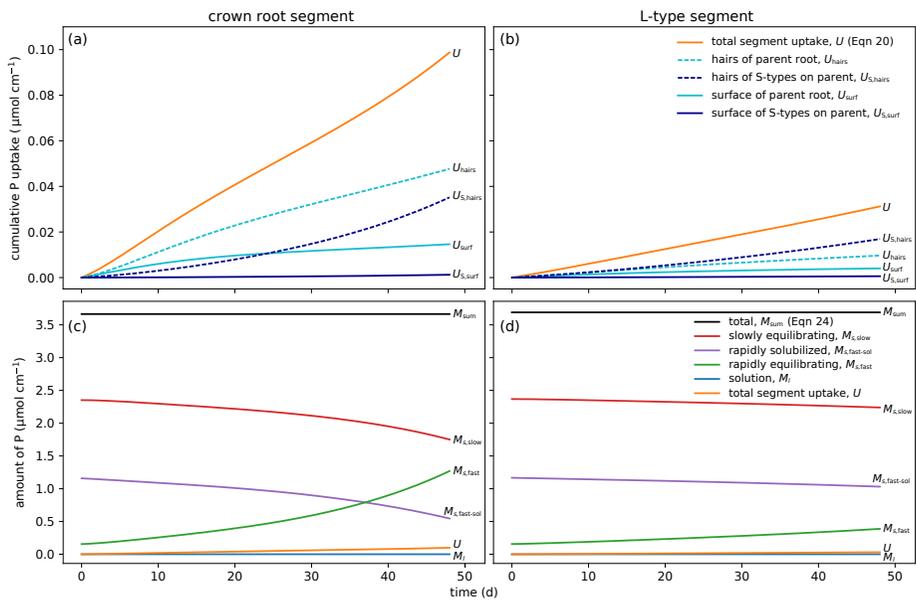
parameters per second are converted to days for the time dimension  $t$ .



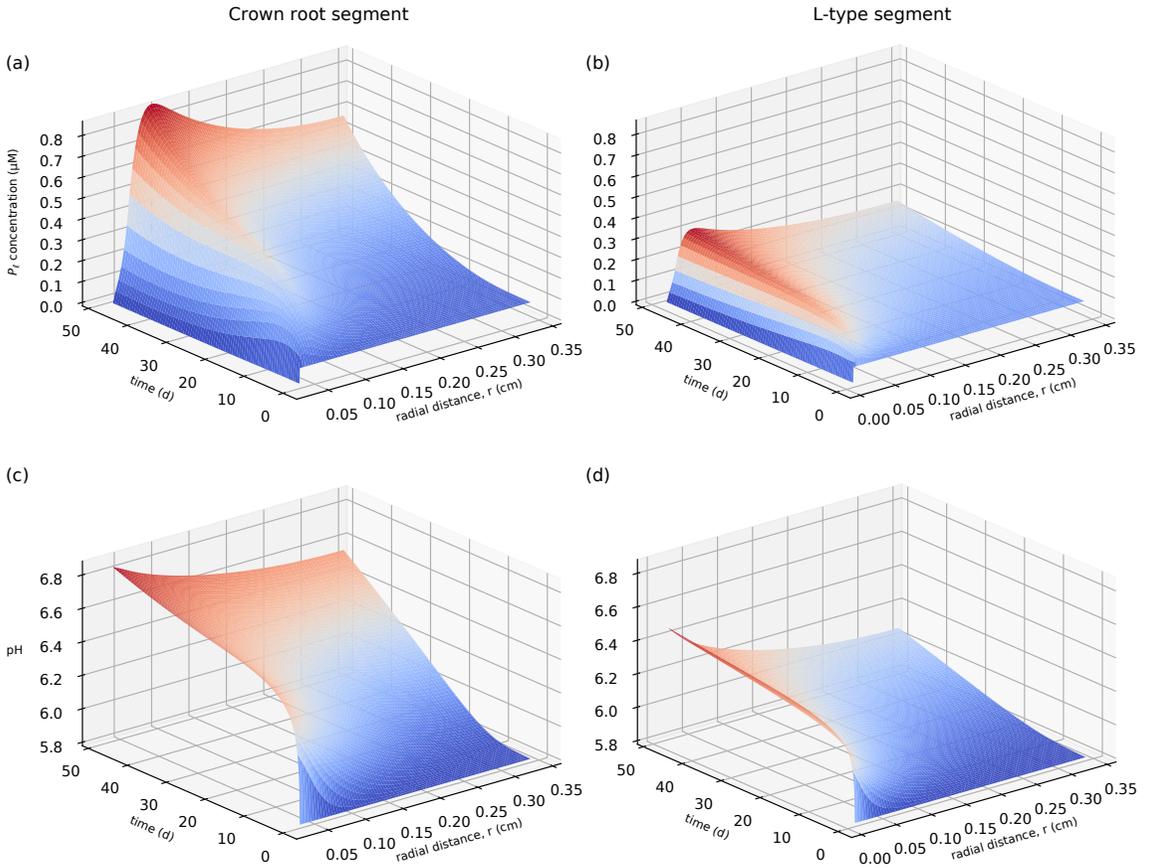
**FIGURE 1** (a) The geometry of the model showing the S-type lateral and the root hairs associated with the parent root (crown or L-type lateral roots); (b) The interchange of P between the soil solid and solution, represented as fast and slow reactions, both sensitive to the concentration of soil bases ( $HCO_3^-$ ); the symbols are defined in Tables 1 and 2.



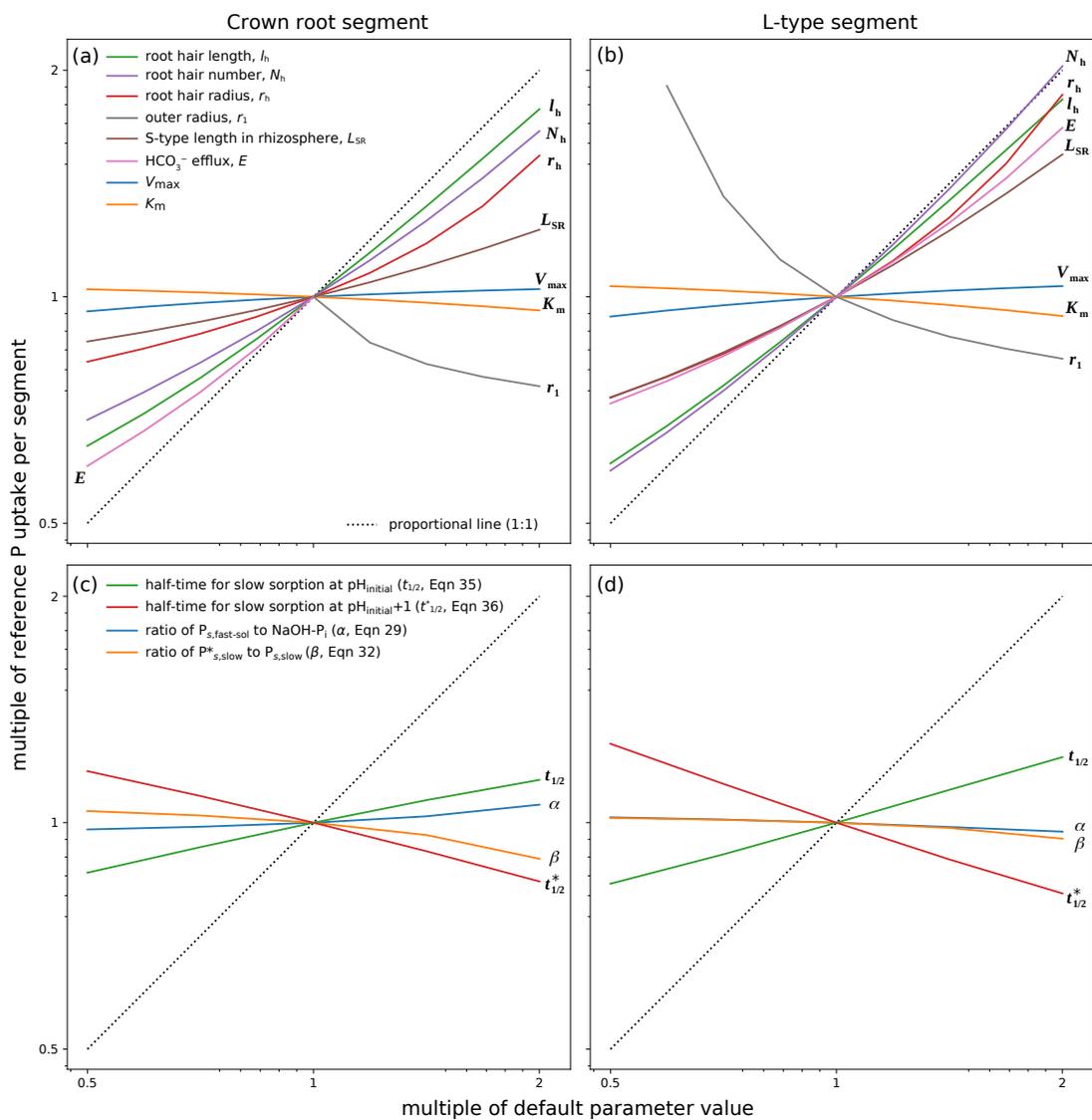
**FIGURE 2** (a) Measured time course of root system length (points) and the fitted function  $L(t) = a \exp(t) / (c + \exp(t))^b$  with parameters  $a = 555$ ,  $b = 0.92$ ,  $c = 3627$  ( $R^2 = 0.928$ ). (b) Simulated P uptake with (solid line) or without (dashed lines) solubilization, and with and without S-type laterals (21% less surface area) and slow P desorption. (c) Simulated contributions of individual root classes including their root hairs (solid lines) and their hairs only (dashed lines) to P uptake.



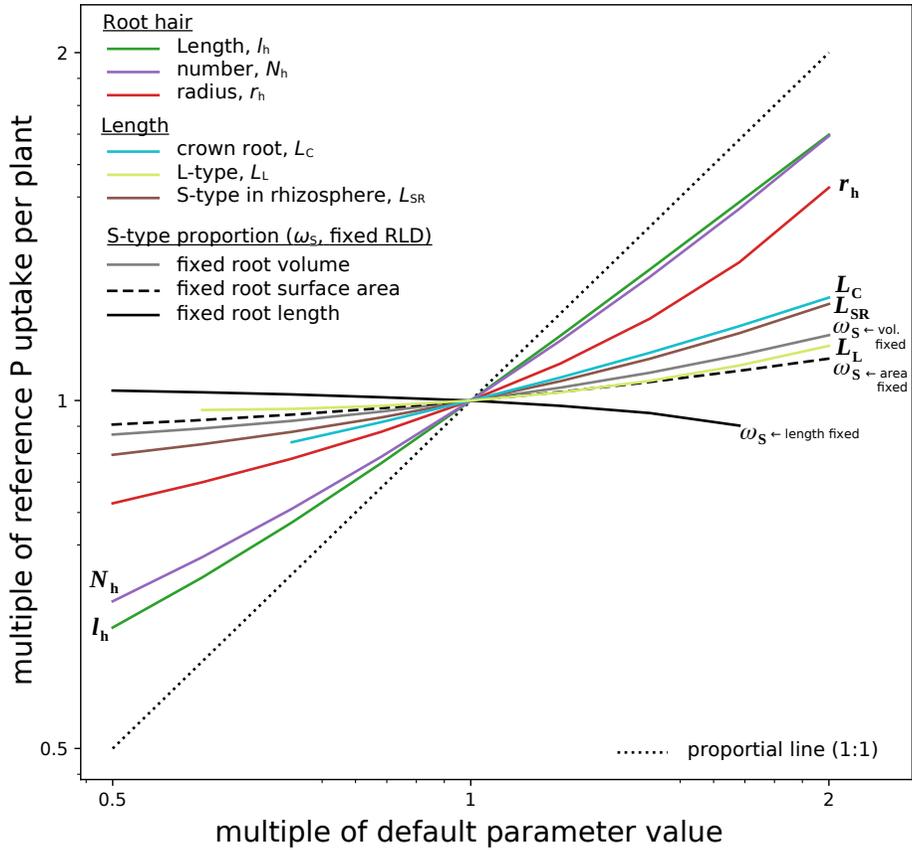
**FIGURE 3** (a), (b) Cumulative P uptake by the surface and hairs of crown roots and L-type laterals of unit length, respectively; and (c), (d) the amount of P in the rhizosphere-volume of crown roots and L-type laterals, respectively.



**FIGURE 4** Concentration profiles over time and radial distance of (a), (b) P in the soil solution and (c), (d) pH for crown roots and L-type laterals. The model parameter values have the standard values given in Tables 2 and 4. Note that the center of the segment is located at  $r = 0$  and the root surface at  $r = r_0$ . Colors follow the contour lines.



**FIGURE 5** Sensitivity of P uptake per unit root length to model input parameters as multiples of the standard values: (a) and (b) root parameters; (c) and (d) soil parameters. The total release of  $\text{HCO}_3^-$  per plant is fixed as the root parameters vary (i. e. for surface area variations), and the total amount of P in soil is kept constant as the soil parameters vary. The parameters  $r_1$  and efflux are varied in a range without changing the pH locally by more than one unit (pink line and grey line). Note axes are logarithmic.



**FIGURE 6** Sensitivity of total P uptake by the root system to model input parameters as multiples of the standard values. Note axes are logarithmic.



710 **Supporting Information: Phosphate solubilization and uptake by upland rice in**  
 711 **strongly sorbing soil**

712 **Derivation of the P-pH coupling**

713 We derive the (rapid) solubilization of in Eqn 6 via the total differential (see Eqn 2 in Nye, 1983),

$$\Delta P_e = \frac{\partial P_e}{\partial P_\ell} \Delta P_\ell + \frac{\partial P_e}{\partial B_\ell} \Delta B_\ell \quad (S1)$$

714 with  $P_e = \theta P_\ell + \rho (P_{s,\text{fast}} + P_{s,\text{fast-sol}})$ . Both gradients are constants, hence, substitution of  $b_p = \partial P_e / \partial P_\ell$   
 715 and  $\lambda = \partial P_e / \partial B_\ell$ , and rearranging gives

$$\Delta P_e = b_p \Delta P_\ell - \lambda b_p \Delta B_\ell \quad (S2)$$

716 which is infinitesimally

$$\frac{dP_e}{dt} = b_p \left( \frac{dP_\ell}{dt} - \lambda \frac{dB_\ell}{dt} \right). \quad (S3)$$

717 Adding  $g = \rho \partial P_{s,\text{slow}} / \partial t$  gives Eqn 6 with left-hand-side as in Eqn 5. Here,  $P_\ell$  does not influence  $P_{s,\text{fast-sol}}$   
 718 ( $\partial P_{s,\text{fast-sol}} / \partial P_\ell = 0$  and  $b_p \partial P_\ell / \partial t = \theta \partial P_\ell / \partial t + \rho \partial P_{s,\text{fast}} / \partial t$ ), but the pH change from an initial pH determines  
 719 the relevant total P concentration. In our simulation, the solubilization of solid phase P is a source according  
 720 to Eqn 33, i. e. the initial concentration of  $P_{s,\text{fast-sol}}$  decreases (Eqn 29). Due to the release of  $\text{HCO}_3^-$  by the  
 721 root, we have  $\partial B_\ell / \partial t > 0$  and  $\partial P_{s,\text{fast-sol}} / \partial t < 0$ . In general, the coupling gradient  $\partial B_\ell / \partial t$  could change sign,  
 722 and when negative (pH declines), the  $P_\ell$  concentration would move to the solid phase.

723 **Parameterization of the pH coupling in the model**

724 For other pH ranges, the model can be adapted by sign-change of  $\lambda$  to perform in the opposite direction of pH  
 725 change, i. e. pH decrease (increase) resulting in increase (decrease) of available P concentration. Note that a  
 726 sign-change of  $\lambda$  implies a sign-change of  $\Delta P_e$  or  $\Delta B_\ell^*$  in Eqn 33, respectively, such that  $\Delta P_e = 0$ .

727 **Uptake and efflux by S-type laterals**

728 The S-type lateral roots were treated as part of the sink term in Eqn 8 analog to Eqn 15 for hairs of parent root.  
 729 This resulted in additional surface per volume for the S-types branching of the 1 cm root segment,  $A_{S,\text{surf}}$ , and  
 730 S-type root hair surface area,  $A_{S,\text{hairs}}$  in the rhizosphere,  $r_0$  to  $r_1$ . The P uptake rates by S-types and their hairs  
 731 are

$$I_S = V \left( r_S, r_{h1S,\text{surf}} \right) A_{S,\text{surf}}, \quad (S4)$$

732 and

$$I_{hS} = V \left( r_h, r_{h1S,\text{hairs}} \right) A_{S,\text{hairs}}, \quad (S5)$$

733 where the influx function  $V = V_{\max} P_{\text{rh}} / (K_m + P_{\text{rh}})$  is evaluated with corresponding  $P_{\text{rh}}(P_{\ell}, r_S, r_{\text{h1S,surf}})$  and  
 734  $P_{\text{rh}}(P_{\ell}, r_h, r_{\text{h1S,hairs}})$  for the radii of S-type,  $r_S$ , and S-type hair,  $r_h$ ; the mid-distances to neighboring S-type  
 735 branches,  $r_{\text{h1S,surf}} = \sqrt{r\pi / (2N_{\text{Scm}})}$ , and S-type root hairs,  $r_{\text{h1S,hairs}} = \sqrt{r_S\pi / (2N_{\text{hS}})}$ ; and the solution  
 736 concentration  $P_{\ell}(r, t)$ , respectively, see Eqn 18 and Eqn 15.

737 The continuous description of surface area per rhizosphere volume for the S-type segment with an  
 738 infinitesimal parameter,  $\delta$ , is

$$A_{\text{S,surf}} = \frac{2r_S N_{\text{Scm}} \delta}{(r + \delta)^2 - r^2} \quad (\text{S6})$$

739 where  $r_S$  is the S-type radius and  $N_{\text{Scm}} = L_{\text{SR}} / L_{\text{S,act}}$  is the number of S-types per cm root. The surface area  
 740 per rhizosphere volume for S-type root hairs is

$$A_{\text{S,hairs}} = \frac{2r_h l_{\text{hS}} N_{\text{Scm}} N_{\text{hS}} \delta}{(r + \delta)^2 - r^2} \quad (\text{S7})$$

741 where  $l_{\text{hS}}$  is the length of an S-type root hair,  $r_h$  the root hair radius, and  $N_{\text{hS}}$  the number of root hairs on  
 742 1 cm S-type (Table 2).

#### 743 Upscaling over root classes

$$\mathcal{P}(t) = \sum_{i \in M} \mathcal{P}_i(t) = \int_0^t \sum_{i \in M} \omega_i \dot{L}(s) U_i(t-s) ds, \quad (\text{S8})$$

744 where  $M := \{\text{L}, \text{C}\}$ , and length-partition weight,  $\omega_i$  (Table 2). Discretized, we derive the upscaled P uptake  
 745 for each root class at  $t_j$  by

$$\mathcal{P}_i(t_j) = \omega_i \sum_{k=1}^j \Delta L_k U_i(t_{j+1-k}), \quad (\text{S9})$$

746 where  $U_i(t_*)$  is the cumulative uptake up to time point  $t_*$  for root class  $i$ . The difference  $\Delta L_k = L(t_k) - L(t_{k-1})$   
 747 is the length that is added to the total root system length at  $t_k$ .

#### 748 Model implementation

749 In the time-direction, the system was solved using BDF (backward differentiation formula as described in  
 750 Shampine and Reichelt, 1997), an implicit variable-order (1–5) method with quasi-constant step size. We  
 751 used Python's Scipy implementation of the BDF, a relative error tolerance of  $1 \times 10^{-5}$ . The initial time step  
 752 was  $1 \times 10^{-10}$  d. We iterated over the line-discretizations of the right-hand-side of the differential equations  
 753 for  $\text{HCO}_3^-$ ,  $P_{\ell}$ , and  $P_{\text{s,slow}}$  in one time-sweep. In this system, hence with BDF and the same stepping, we  
 754 also integrated: the uptake for the root hairs, the root surface, and S-type (hairs and surface separately) of the  
 755 parent.

756 The spatial discretization of the diffusive transport was of second-order (central difference for second  
 757 derivatives, cubic upwind interpolation for the first derivatives). Main root hairs and S-type surface area  
 758 per volume were spatially discretized as in Kuppe et al. (2021a). We used a spatial step size of  $\Delta r =$   
 759  $1.2344 \times 10^{-3}$  cm for crown roots and  $\Delta r = 1.3144 \times 10^{-3}$  cm for L-types.

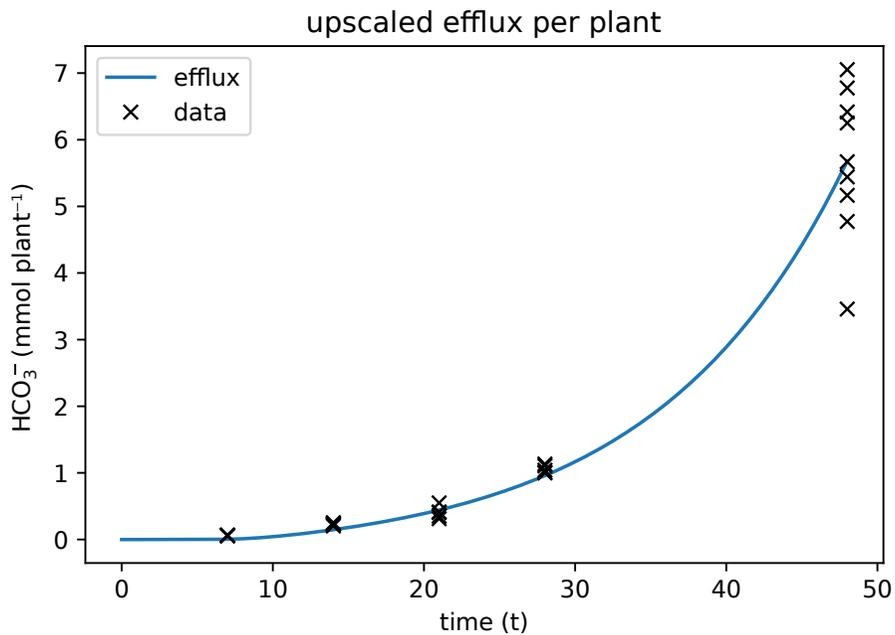
760 **Supplemental Figures**

761 **S1** We calculated the amount of  $\text{HCO}_3^-$  per plant based on the plant biomass, which is supposed to correlate  
762 with nitrate uptake where the roots release  $\text{HCO}_3^-$ .

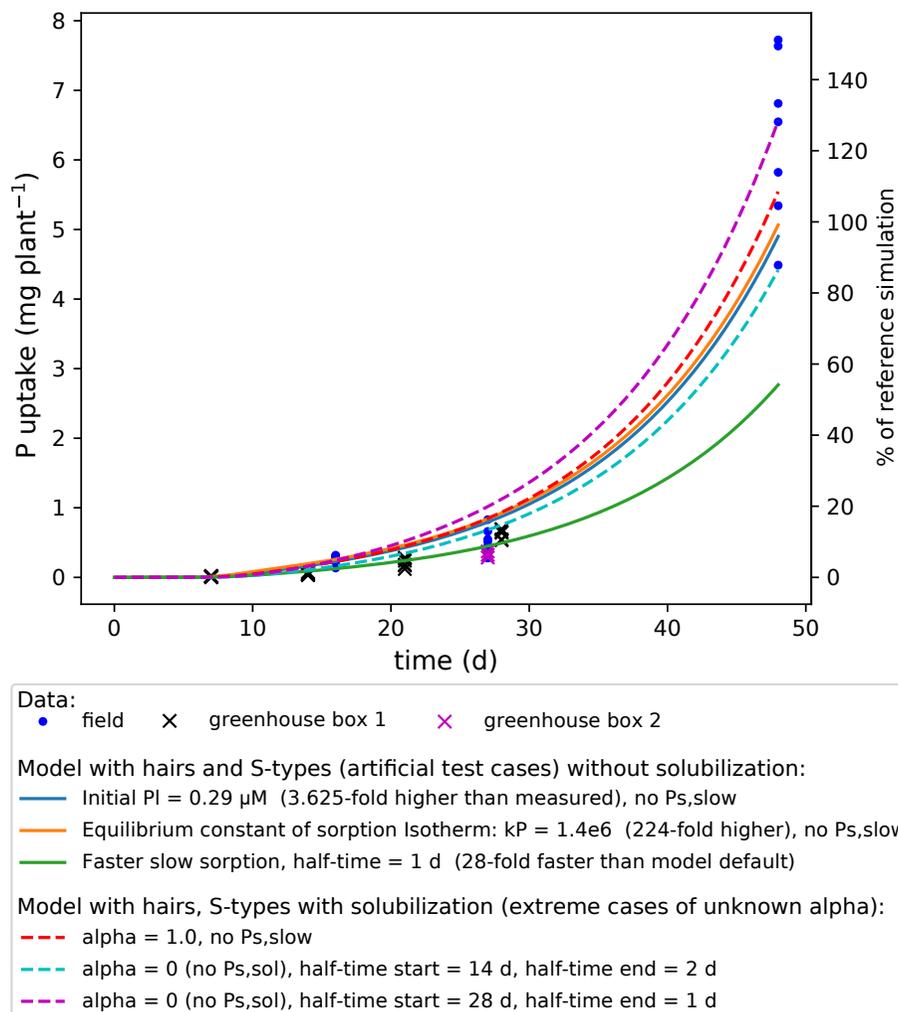
763 **S2** We calculated and compared several parameter combinations, showing that the measured P uptake can be  
764 simulated with parameters that are far away from the measured ones.

765 **S3** When not combining S-type laterals to the parent root rhizosphere but simulating them as separate root  
766 class with the parameters from table 4, we did not achieve the measured P uptake, and the relative  
767 importance of L-types and S-types change places in their ranking.

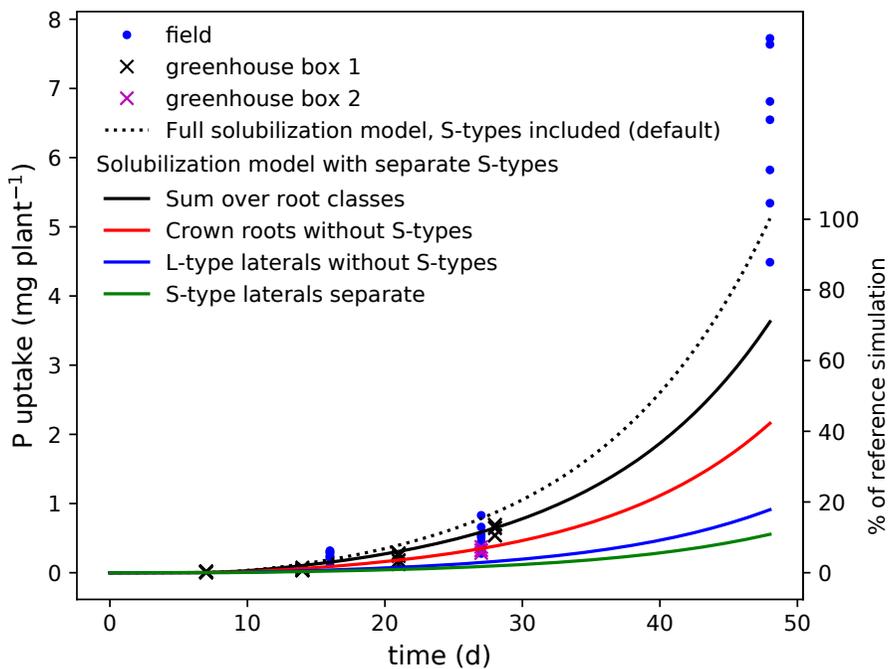
768 **S4** In strongly sorbing soil, the P in solution diffuses slowly.



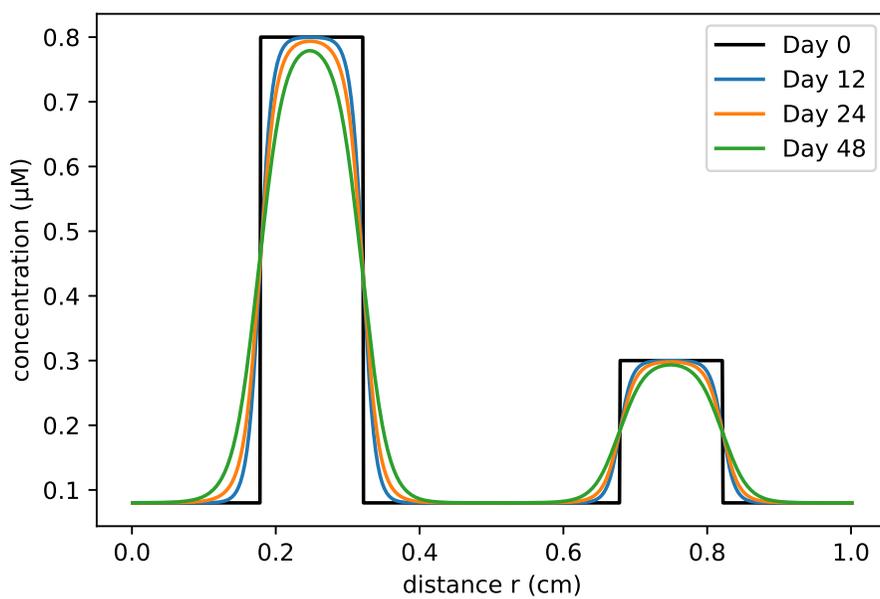
**FIGURE S1** Amount of released  $\text{HCO}_3^-$  of the total root system, integrated (upscaled) and derived from data. This curve was fixed for the sensitivity analysis of the plant traits. Data is based on total dry weight scaled by  $1.49 \text{ mmol g}^{-1}$ , the difference of cation-anion-uptake.



**FIGURE S2** Results of simulation runs. Without coupling we would either need a much higher starting P concentration in solution (red line) or a much higher instantaneous solid P phase (violet line). Therefore it is crucial to know the sizes of the pools. None of the simulations were able to match the shape of the exponential curve fit to the data. To reach a higher concentration at the end of the time interval (day 48), the earlier point would be over-estimated even more.



**FIGURE S3** Results of S-types simulated separately with the default parameter set, not as part of the crown and L-type rhizosphere. S-type, L-type, and crown root length distribution was (50-33-17%) as in table 2. The ranking of the contribution to uptake reflects the relative surface areas of the root classes. Without placing the S-type laterals in the rhizosphere, most important were thicker roots (crown), then L-type and S-type laterals.



**FIGURE S4** Diffusion of P in strongly sorbing soil over 48 days with diffusion coefficient in water of  $8.9 \times 10^{-6} \text{ cm}^2 \text{ s}^{-1}$ , impedance factor of 0.24, and soil buffer power of 5440.