

1 **Sediments in sea ice drive the Canada Basin surface**
2 **Mn maximum: insights from an Arctic Mn ocean**
3 **model**

4 **B. Rogalla¹, S. E. Allen¹, M. Colombo¹, P. G. Myers², K. J. Orians¹**

5 ¹Department of Earth, Ocean, and Atmospheric Sciences, University of British Columbia, Vancouver,
6 British Columbia V6T1Z4, Canada

7 ²Department of Earth and Atmospheric Sciences, University of Alberta, 1-26 ESB, Edmonton, Alberta
8 T6G2E3, Canada

9 **Key Points:**

- 10 • We developed an ocean model of Mn in the Canadian Arctic Archipelago and the
11 Canada Basin that captures observed spatial variations
- 12 • Non-local sediments transported by sea ice are a key source of nutrients such as
13 Mn to the Canada Basin
- 14 • Rivers are not as important for Mn on basin scales as generally identified

Corresponding author: Birgit Rogalla, brogalla@eoas.ubc.ca

Abstract

Biogeochemical cycles in the Arctic Ocean are sensitive to the transport of materials from continental shelves into central basins by sea ice. However, it is difficult to assess the net effect of this supply mechanism due to the spatial heterogeneity in sea ice content. Manganese (Mn) is a micronutrient and tracer which integrates source fluctuations in space and time. The Arctic Ocean surface Mn maximum is attributed to freshwater, but studies struggle to distinguish sea ice and river contributions. Informed by observations from 2015 Canadian GEOTRACES cruises, we developed a three-dimensional dissolved Mn model within a 1/12 degree coupled ocean-ice model centered on the Canada Basin and the Canadian Arctic Archipelago (CAA). Simulations from 2002 to 2019 indicate that annually, 82% of Mn contributed to the Canada Basin upper ocean is released by sea ice, while rivers, although locally significant, contribute only 4%. Downstream, sea ice provides 18% of Mn transported from Parry Channel into Baffin Bay and rivers supply 4%. While rivers are often considered the main source of Mn, our findings suggest that in the Canada Basin they are much less important than sea ice. However, within the shelf-dominated CAA, both rivers and sediment resuspension are important. Climate induced disruption of the transpolar drift may reduce the Canada Basin Mn maximum and supply downstream. Other nutrients found in sediments, such as Fe, may be similarly affected. These results highlight the vulnerability of the biogeochemical supply mechanisms in the Arctic Ocean and the subpolar seas to climatic changes.

Plain Language Summary

Autumn storms on the Siberian side of the Arctic Ocean churn up sediment that freezes into sea ice. The prevailing ocean currents and winds push this sea ice across the Arctic Ocean towards the Canada Basin, where it melts and releases the sediment into the ocean. Sediment contains manganese (Mn) and other nutrients that help support plankton and life. Using our Mn ocean model, 82% of Mn in the Canada Basin comes from “dirty” sea ice from 2002 to 2019, while rivers supply 4%. As a result of climate change, less dirty sea ice may make it across the Arctic Ocean, which could reduce this supply system of Mn and other similar nutrients. This also has impacts downstream: water from the Canada Basin travels through the shallow Canadian Arctic Archipelago (CAA) into Baffin Bay and eventually the North Atlantic. We found that about 18% of Mn transported along this route comes from “dirty” sea ice. In the CAA, other sources contribute

47 as well: tides churn up sediments from the ocean floor and many rivers flow into the chan-
48 nels. Our study highlights ways in which climate change may impact the nutrient sup-
49 ply systems in the Arctic Ocean.

50 **1 Introduction**

51 As the sea ice regime in the Arctic Ocean transitions from multi-year ice to pre-
52 dominantly first-year ice with overall reductions in sea ice extent, thickness and altered
53 drift patterns (Stroeve et al., 2012; Stroeve & Notz, 2018; Spreen et al., 2011; Kwok et
54 al., 2013), biogeochemical cycles and primary productivity are impacted through changes
55 to the sea ice supply mechanism. The Arctic Ocean continental shelves in particular, con-
56 nect land and ocean through the transfer of river runoff and sea ice from near-shore re-
57 gions to the central basins (Charette et al., 2016). Reductions in sea ice export from the
58 shelves weaken the long range transport of ice-rafted matter (Krumpen et al., 2019), in-
59 cluding sediments (Dethleff et al., 2000; Darby et al., 2011), nutrients and trace metals
60 (Tovar-Sánchez et al., 2010; Measures, 1999), pollutants (Pfirman et al., 1995; Peeken
61 et al., 2018) and climate-relevant gases (Damm et al., 2018), to the surface ocean in re-
62 gions far away from boundary sources. It is challenging to quantify the contribution of
63 materials supplied by sea ice with observations alone due to the high spatial and tem-
64 poral variability in the amount of sediment in sea ice and because it is difficult to dis-
65 tinguish from additional contributions to the surface ocean such as river runoff. How-
66 ever, it is clear that changes to the physical processes in the Arctic Ocean will have im-
67 pacts on the biogeochemical cycles and primary productivity of the basins themselves,
68 as well as downstream in subpolar seas (Drinkwater & Harding, 2001; Greene & Persh-
69 ing, 2007).

70 Continental shelves cover half of the area of the Arctic Ocean (Jakobsson, 2002)
71 and their shallow depths facilitate the incorporation of suspended matter into sea ice as
72 it forms (Kempema et al., 1989). The narrow and deeper North American shelves are
73 not as important for basin-wide sea ice sediment transport as the wide Siberian shelves
74 (Eicken et al., 2005). In the Siberian shelf regions, fast ice builds up near shore in the
75 fall, coinciding with storm-related resuspension events, forming sediment-rich sea ice (Nürn-
76 berg et al., 1994). The transpolar drift transports this sea ice towards the North Pole
77 and the anticyclonic Beaufort Gyre redirects a portion into the Canada Basin. This pas-
78 sage takes several years, during which the ice undergoes cycles of melting, freezing and

79 deformation. The materials released by melt alter the geochemical signature of the un-
80 derlying water (Pfirman et al., 1995). While several studies indicate an increase in ex-
81 change of ice-rafted material through increased drift speeds (Spren et al., 2011; New-
82 ton et al., 2017), a recent study indicates a disruption of the long range transport of sed-
83 iments by sea ice due to the melt of first-year ice before it is incorporated into the trans-
84 polar drift (Krumpfen et al., 2019) with implications for the surface ocean of the endmem-
85 bers of this transport pathway: Fram Strait, and indirectly the Canada Basin, the Cana-
86 dian Arctic Archipelago, and the subpolar North Atlantic. In order to establish the im-
87 portance of sediment from sea ice for biogeochemical cycles in the indirectly impacted
88 regions of the Canada Basin and the Canadian Arctic Archipelago, we developed a model
89 of dissolved manganese (Mn).

90 Mn is reactive trace element and an important micronutrient which shares many
91 sources with iron (Fe) in the Arctic Ocean (Brand et al., 1983; Bruland et al., 1991; Jensen
92 et al., 2020). Mn has a scavenged-type profile with high concentrations near sources and
93 low background concentrations. This contrast makes it a convenient source tracer. Over
94 the Arctic Ocean shelves, sediment resuspension contributes Mn to the lower water col-
95 umn (Evans & Nishioka, 2018; Colombo et al., 2020). However, the majority of exter-
96 nal sources supply Mn to the ocean surface, contributing to the surface Mn maximum.
97 In the Arctic Ocean, this surface maximum is attributed to freshwater sources (Campbell
98 & Yeats, 1982; Yeats & Westerlund, 1991; Middag et al., 2011b; Cid et al., 2012; Kondo
99 et al., 2016; Colombo et al., 2020). Observational studies have identified the origin of
100 this freshwater as river discharge (Campbell & Yeats, 1982; Yeats & Westerlund, 1991;
101 Evans & Nishioka, 2018), sea ice meltwater (Measures, 1999; S. Wang et al., 2014) (for
102 Fe), or a combination of both (Middag et al., 2011b; Cid et al., 2012; Kondo et al., 2016;
103 Colombo et al., 2020). Observations of riverine Mn indicate significantly higher concen-
104 trations than in the ocean (Colombo et al., 2019). Similarly, trace metals and nutrients
105 in sea ice occur in concentrations in excess of those in the ocean (Campbell & Yeats, 1982;
106 Hölemann et al., 1999; Granskog et al., 2003; Krachler et al., 2005; Aguilar-Islas et al.,
107 2008; Tovar-Sánchez et al., 2010; Kondo et al., 2016; Evans & Nishioka, 2018). The im-
108 portance of these components depends in part on the distance and pathway of input into
109 the ocean (Fichot et al., 2013). So, while the Canada Basin is relatively distant from land,
110 the narrow and shallow systems of channels that make up the CAA are in close contact

111 with the land-ocean interface and may be more directly impacted by boundary processes
112 such as river discharge and sedimentary inputs.

113 In order to distinguish the individual importance of external Mn sources within the
114 Canada Basin and the CAA, model studies are needed. Past studies have used tracers
115 such as terrestrial dissolved organic matter (Fichot et al., 2013) and the oxygen isotope
116 ratio (Yamamoto-Kawai et al., 2009) to distinguish the contributors to freshwater in the
117 Canada Basin. Mn is an interesting complementary tracer because of its role as a nu-
118 trient and because it integrates processes that fluctuate on a short time scale. As a re-
119 sult, Mn helps address one of the main limitations of the study of sediment entrainment
120 and export events by sea ice: that they are episodic and localized in nature (Eicken et
121 al., 2005). Similarly, while sediment resuspension occurs intermittently, Mn integrates
122 the effect of this component on the lower water column. After establishing the contri-
123 butions of the Mn sources, we use Mn as a tool to study the general role of sea ice trans-
124 port for biogeochemical cycles.

125 In this paper, we present a model of Mn in the Canadian Arctic Archipelago and
126 the Canada Basin, informed by in situ observations collected during the 2015 Canadian
127 GEOTRACES cruises (Colombo et al., 2020). Our work builds on the comprehensive
128 first global model of Mn in the ocean (Van Hulst et al., 2017) and previous smaller scale
129 models of Mn in the North Pacific Ocean (Johnson et al., 1996) and near hydrothermal
130 vents (Lavelle et al., 1992). We incorporate new parameterizations for sediment resus-
131 pension, release of shelf sediments in sea ice, and fluvial contributions, to capture the
132 drivers of Mn distributions in the Canadian Arctic. With this model, we show that the
133 long range transport of sediments by sea ice from the Siberian shelves drives the surface
134 Mn maximum in the Canada Basin while riverine contributions, although locally signif-
135 icant, are not as important as generally identified. Using these results, we discuss im-
136 plications of future sea ice melt on Mn and Fe nutrient budgets in the Canada Basin and
137 downstream in the Canadian Arctic Archipelago and Baffin Bay.

138 **2 Methods**

139 **2.1 Coupled Ocean-Ice Model**

140 For our simulations, we use ocean and ice dynamics calculated by the Arctic and
141 Northern Hemispheric Atlantic (ANHA12) configuration (Hu et al., 2018) of the Nucleus

142 for European Modelling of the Ocean (NEMO) version 3.4 (Madec, 2008). The ANHA12
143 configuration has a nominal horizontal resolution of $1/12^\circ$ which resolves freshwater fluxes
144 associated with coastal currents in the CAA, as well as eddies (Bacon et al., 2014; Chel-
145 ton et al., 1998). The position of the grid’s artificial pole in Northern Canada increases
146 the resolution in the CAA to about 2-3 km (Fig. 1). In the vertical, there are 50 depth
147 levels ranging from 1 m thickness at the surface to 454 m near the bottom. The bottom
148 bathymetry is represented using partial steps.

149 The ANHA12 domain has two open boundaries: one in Bering Strait and the other
150 at 20°S in the Atlantic Ocean. These boundaries are forced with Global Ocean Reanal-
151 yses and Simulations data (Masina et al., 2017). The ocean surface is forced with hourly
152 atmospheric data from the Canadian Meteorological Centre’s global deterministic pre-
153 diction system (Smith et al., 2014) and the rivers are forced with monthly runoff clima-
154 tology with enhanced Greenland melt runoff (Dai et al., 2009; Bamber et al., 2012). The
155 river forcing from 2010 is repeated for the following years (Hu et al., 2019).

156 The sea ice in ANHA12 is represented using the dynamic and thermodynamic Louvain-
157 la-Neuve (LIM2) sea ice model with an elastic-viscous-plastic ice rheology (Fichefet &
158 Maqueda, 1997; Bouillon et al., 2009). An evaluation of LIM2 in the ANHA12 config-
159 uration is provided by Hu et al. (2018). The general spatial distribution of ice thickness
160 within the Archipelago is captured well. In the model, the northern CAA has very thick
161 sea ice (> 4 m), the central parts have intermediate thickness ice (2.5-3 m), and there
162 is thin (< 2 m) sea ice on the east side of the CAA and in southern channels. The ANHA12
163 simulations are limited by the lack of a land-fast ice parameterization, resulting in ice
164 velocities that are higher than observed in Parry Channel, impacting the winter trans-
165 port (Grivault et al., 2018). In addition, tides are not included and as a result, the polynyas
166 which form due to tidally enhanced mixing are not well reproduced (Hughes et al., 2018).

167 The advection and diffusion of tracers are calculated within NEMO by the TOP
168 engine (Gent et al., 1995; Lévy et al., 2001). We use the Flow Relaxation Scheme (FRS)
169 for the tracer boundary conditions and tracer advection is calculated with the Total Vari-
170 ance Dissipation (TVD) scheme (Zalesak, 1979). The vertical diffusion of tracers is cal-
171 culated from the Turbulent Kinetic Energy closure scheme within ANHA12 and the hor-
172 izontal eddy diffusivity parameter is set to $50.0 \text{ m}^2 \text{ s}^{-1}$.

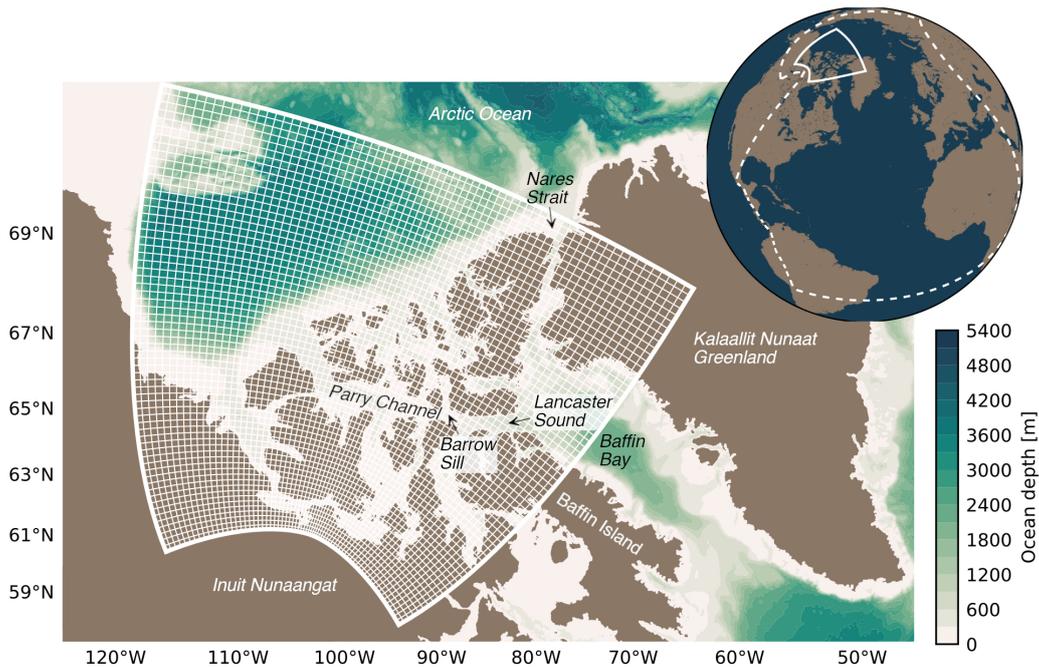


Figure 1. The Mn model domain is centered on the Canadian Arctic Archipelago with highest horizontal resolution in the south (about 3 km). The nominal horizontal resolution of the grid is $1/12^\circ$; the white lines depict one in every ten grid points. The solid white line in the inset globe shows the Mn model domain extent, while the dashed white line delineates the domain of the Arctic and Northern Hemispheric Atlantic configuration (Hu et al., 2018) of the ocean-ice model.

173 2.2 Model of Mn in the Canadian Arctic

174 The Mn model runs offline in NEMO version 3.6 using five day averaged dynam-
 175 ics fields from the ANHA12 reference run from January 2002 to December 2019 (Hu et
 176 al., 2018). The Mn model consists of two main sets of computations: the advection and
 177 diffusion of tracers calculated by the NEMO-TOP engine (Gent et al., 1995; Lévy et al.,
 178 2001), and the source and sink contributions. The source and sink parameterizations were
 179 developed guided by observations from the 2015 Canadian GEOTRACES cruises (Colombo
 180 et al., 2020) and inspired by the first global model of Mn (Van Hulst et al., 2017). In
 181 order to reduce the computational cost, we calculate the model on a sub-domain of ANHA12,
 182 centered on the CAA (Fig. 1). Note that since we are running offline, the physics orig-
 183 inates from the full domain.

184 The known sources and sinks of Mn in the ocean are: rivers, hydrothermal vents,
 185 sediment diffusion, sediment resuspension, reversible scavenging, sinking, uptake and rem-
 186 ineralization, atmospheric dust deposition, and flux from ice (Middag et al., 2011b; Balzer,
 187 1982; Klinkhammer & Bender, 1980; Evans & Nishioka, 2018). From this list, we incor-
 188 porate the processes that are important for dissolved Mn in the Arctic (summarized in
 189 Fig. 2 and Eqn. 1 and 2). In order to model the reversible scavenging of Mn, we incor-
 190 porate Mn oxides (oMn) in addition to dissolved Mn (similar to Van Hulst et al. (2017)).
 191 We do not model particle-bound Mn (pMn), but rather incorporate the indirect effect
 192 of pMn onto dMn through dissolution from the source components. We did not incor-
 193 porate hydrothermal vents as a source of Mn in the Arctic, since the influence of the Gakkel
 194 Ridge is restricted to Nansen and Amundsen Basins due to scavenging nearby the source
 195 (Lavelle et al., 1992; Middag et al., 2011b). We also do not include sediment diffusion
 196 (reductive dissolution) nor biological uptake and remineralization because observations
 197 have indicated that these processes are not significant for Mn in the CAA (Colombo et
 198 al., 2020). The overall Mn model equations are:

$$\frac{\partial dMn}{\partial t} = S_{river} + S_{sediment} + S_{atm} + S_{ice} + S_{sed\ ice} + R_{scav} + advection + diffusion \quad (1)$$

199

$$\frac{\partial oMn}{\partial t} = -R_{scav} - R_{sink} + advection + diffusion \quad (2)$$

200 which includes the contribution of rivers, S_{river} , sediment resuspension (non-reductive
 201 dissolution), $S_{sediment}$, atmospheric dust deposition, S_{atm} , dust flux from ice, S_{ice} , sed-
 202 iment released by ice, $S_{sed\ ice}$, and the reversible scavenging terms, R_{scav} . The details
 203 of the parameterizations are described in the following sections and the parameter val-
 204 ues used for the base run are listed in Table 1.

205 The model was initialized with output from the global Mn model (Van Hulst et
 206 al., 2017) and at the sub-domain boundaries, concentrations are held constant. At the
 207 model boundaries, the ratio of dissolved to oxidised Mn from the global model were not
 208 representative (oxidised Mn was too low) and resulted in unusual scavenging behavior.
 209 To address this, we took values for the dissolved and oxidised Mn concentrations in a
 210 band 15 grid cells towards the interior of the basin (where the model had established nor-
 211 mal scavenging behavior) from a test model run at the end of spin up and used those
 212 values for the boundary conditions.

Table 1. Constants and parameter values used in the Mn model runs.

Parameter	Description	Value	Source
α_0	Solubility of Mn at 4°C	0.65	Fishwick et al. (2018)
$f_{Mn\ crust}$	Mn fraction in Earth's crust	527 ppm	Wedepohl (1995)
$f_{Mn\ sed}$	Mn fraction in marine sediment	270 ppm	Macdonald and Gobeil (2012)
m	Molar mass of Mn	54.938 g mol ⁻¹	—
k_d	Reduction and desorption rate	$4.72 \cdot 10^{-7}$ s ⁻¹	Bruland et al. (1994)
	Photoreduction rate	$2.72 \cdot 10^{-5}$ s ⁻¹	Sunda and Huntsman (1994)
k_p	Oxidation and adsorption rate	$7.00 \cdot 10^{-7}$ s ⁻¹	This study ^a
s_{ox}	Sinking rate	1-10 m day ⁻¹	Van Hulst et al. (2017)
C	Tidal erosion tuning constant	$3.3 \cdot 10^{-6}$	This study
γ	Solubility tuning constant	0.065	This study
R / SPM	River characteristic content		This study ^b
	- Glacial	164 nM / 261 mg L ⁻¹	
	- Continental	30 nM / 12 mg L ⁻¹	
	- Other	5 nM / 4 mg L ⁻¹	

^aUsing data from Colombo et al. (2020); Li (2017).

^bUsing data from Colombo et al. (2019); Brown et al. (2020).

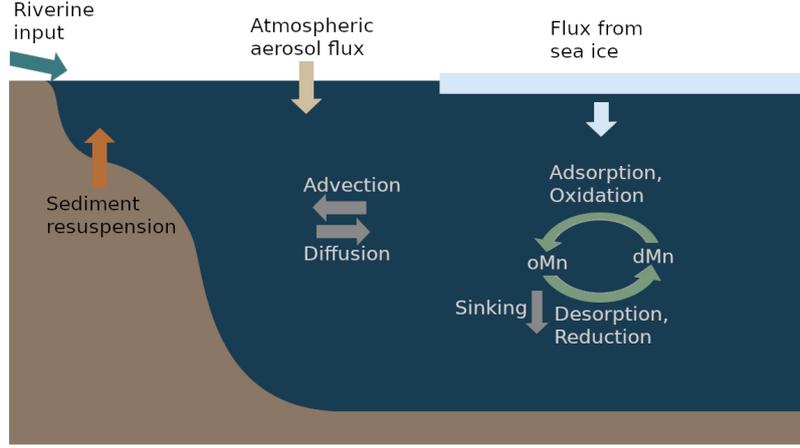


Figure 2. Summary of the processes that affect Mn concentrations in the Canadian Arctic Archipelago and the Canada Basin.

213 **2.2.1 Riverine Source**

214 River discharge contributes Mn to the shelf seas and into the Arctic Ocean (Middag
 215 et al., 2011a). Dissolved Mn is contributed directly in its dissolved form and indirectly
 216 through the dissolution from particle-bound Mn. The contribution of riverine Mn de-
 217 pends on the river discharge, Q , and the concentration in the rivers. These concentra-
 218 tions vary based on properties of the river’s catchment basin: glacial rivers are strongly
 219 enriched in dissolved Mn, continental rivers are somewhat enriched, and in all other rivers,
 220 Mn is not significantly enriched (Colombo et al., 2019). At each time step, the rivers con-
 221 tribute dissolved Mn following:

$$S_{river} = \frac{\overbrace{Q}^{\text{dissolved Mn}}}{\rho_0 \Delta z_{surface}} R_{class} + \beta \frac{\overbrace{Q}^{\text{particle origin dissolved Mn}}}{\rho_0 \Delta z_{surface}} \frac{SPM_{class} \cdot f_{Mn, crust} \cdot \alpha_0}{m} \quad (3)$$

222 where ρ_0 is the density of the river water, $f_{Mn, crust}$ is the crustal abundance of Mn, α_0
 223 is the solubility of Mn, m is the molar mass of Mn, $\Delta z_{surface}$ is the surface grid box thick-
 224 ness, and β is a factor which ranges from 0-1 in our experiments. We use an average value
 225 of the solubility of Mn (65%) measured in seawater at 4°C, since this lower temperature
 226 better reflects the CAA (Fishwick et al., 2018). This solubility falls within the range mea-
 227 sured in samples across the world (Fishwick et al., 2018). Each river is assigned a class
 228 with an associated characteristic trace metal concentration, R_{class} , and suspended par-
 229 ticulate matter content, SPM_{class} , based on catchment basin properties: glacial, con-

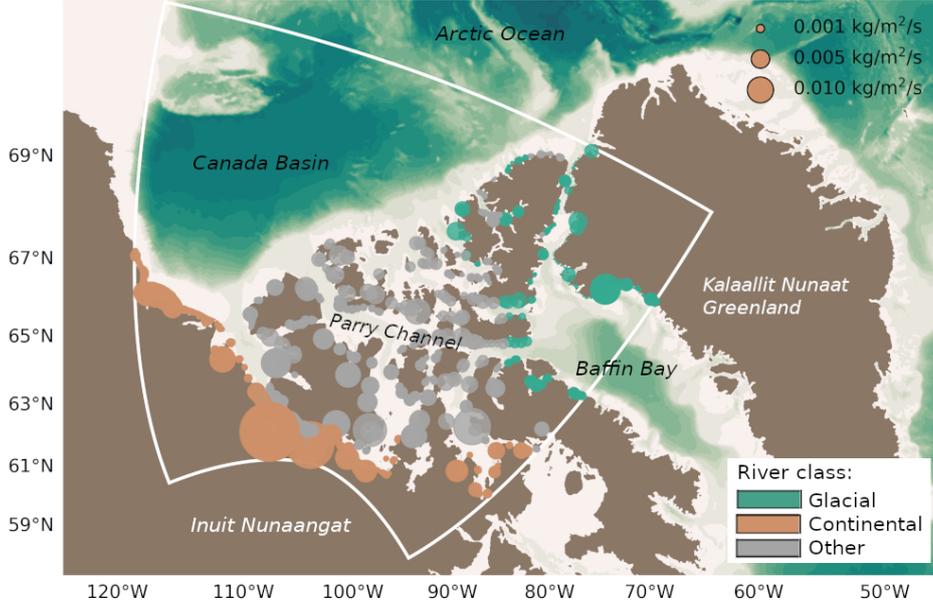


Figure 3. Model rivers were classified based on their drainage basin properties: glacial (green), continental (orange), or other (gray). The points in this map are the locations of the river mouths in the model and their sizes are proportional to the river discharge in September, 2015 (forcing is repeated from year 2010). Note that the river mouths are remapped to accurately represent the ocean freshwater circulation, hence some large rivers are represented as single point sources, while others such as the Mackenzie River consist of multiple point sources along the coastline (Hu et al., 2019; Hayashida et al., 2019).

230 tinal, and other (Fig. 3). The Mn concentrations and SPM content associated with
 231 the classes are determined from rivers sampled in the CAA (Colombo et al., 2019; Brown
 232 et al., 2020). In the base case, the Mn concentrations in glacial rivers are 164 nM, con-
 233 tinal rivers 30 nM, and all other rivers 5 nM. The SPM content is 261 mg L⁻¹ in glacial
 234 runoff, 12 mg L⁻¹ in continental runoff, and 3 mg L⁻¹ in all other rivers.

235 *2.2.2 Atmospheric Aerosol Flux and Release from Sea Ice*

236 Atmospheric aerosols contribute Mn to the ocean through direct deposition to sur-
 237 face waters, Φ_{atm} , or through the deposition onto sea ice and the subsequent release dur-
 238 ing melt, Φ_{ice} . We parameterized this contribution as:

$$S_{atm\ or\ ice} = \frac{\alpha_0 \cdot f_{Mn\ crust}}{m \cdot \Delta z_{surface}} \cdot \Phi_{atm\ or\ ice} \quad (4)$$

239 The atmospheric and sea ice flux terms are derived from monthly Community Earth Sys-
 240 tem Model (CESM) results. The combined monthly dry and wet atmospheric deposi-
 241 tion fluxes originate from historical (1920-2005) and future (2006-2080) runs of the Com-
 242 munity Atmosphere Model with Chemistry (CAM-Chem) downloaded from the Climate
 243 Data Gateway (CESM1 CAM5 BGC Large Ensemble Atmosphere Post Processed Data;
 244 Tilmes et al. (2016)). We estimate tracer fluxes from ice using the monthly Community
 245 Ice Code ensemble results (CICE; Holland et al. (2012); Kay et al. (2015)). These en-
 246 semble run sets have a horizontal atmospheric resolution of $0.9 \times 2.5^\circ$ and ocean/ice res-
 247 olution of $1.6 \times 2.5^\circ$ which we linearly interpolated to the ANHA12 grid.

248 *2.2.3 Sediment Resuspension over the Continental Shelf*

249 Dissolved Mn increases near the ocean floor in the Canadian Arctic as a result of
 250 sediment resuspension (Colombo et al., 2020). Sediment resuspension occurs intermit-
 251 tently, however, Mn integrates the resuspension events and thereby provides a cumula-
 252 tive view. We incorporated sediment resuspension as a continuous process:

$$S_{sediment} = \Phi_{erosion} \cdot \frac{\alpha \cdot f_{Mn\ sed}}{m \cdot \Delta z_{bottom}} \quad (5)$$

253 where $f_{Mn\ sed}$ is the fraction of Mn in marine sediments. The fraction of Mn found in
 254 marine sediments is likely to be lower than those measured in the continental crust (i.e.
 255 Wedepohl (1995)), since it's undergone some amount of chemical transformation. So, we
 256 used the Mn fraction estimated by Macdonald and Gobeil (2012) which is based on sed-
 257 iments in cores on the shelf and slope regions surrounding the Canada Basin. In Eqn. 5,
 258 $\Phi_{erosion}$ is the “erosion ability” (see Fig. S7 for the forcing field). This term incorporates
 259 the spatial differences in dynamics within the CAA. West of Barrow Sill, the system has
 260 lower mixing rates (Hughes et al., 2018) and tidal speeds (Epstein, 2018), than the re-
 261 gion east of Barrow Sill and around the central sill area. These differences impact the
 262 sediment resuspension rates, apparent in the much stronger near-bottom increases of ob-
 263 served dMn in the eastern CAA (Colombo et al., 2020). We estimate the ability of sed-
 264 iment to be eroded with the barotropic tidal speed, U_{tidal} , and a tuning constant, C :

$$\Phi_{erosion} = C \cdot U_{tidal}^2 \quad (6)$$

265 The barotropic tidal speeds are from the MOG2D-G model (Carrère & Lyard, 2003) and
 266 are significantly higher in the eastern CAA, compared to the western CAA (Epstein, 2018).
 267 Locations where the tidal speeds are less than 1 cm s^{-1} are masked, since this is below

268 a critical threshold for motion for particles greater than 0.1 mm (i.e. sand). In areas where
 269 resuspension occurs frequently, the easily accessible Mn on particles has already been
 270 removed, resulting in a lower solubility. So, we reduce the solubility in Eqn. 5 at high
 271 tidal speeds according to:

$$\alpha = \alpha_0 \cdot \frac{\gamma(1 - e^{-U_{tidal}^2/\gamma})}{U_{tidal}^2} \quad (7)$$

272 where γ is a tuning parameter. At small tidal speeds, Eqn. 7 approaches α_0 while at tidal
 273 speeds greater than about 0.14 m s⁻¹, the solubility becomes smaller and as a result, the
 274 overall resuspension rate approaches a constant $\alpha_0\gamma C$ (Fig. S8). The tuning parameters
 275 were estimated based on the best fit to mean observations from several tuning runs.

276 **2.2.4 Sediment Entrained in Sea Ice and Subsequent Melt**

277 Sediment entrained in sea ice has been identified as an important source of reac-
 278 tive trace metals such as aluminum and iron in the ocean, and thus may also be impor-
 279 tant for Mn (Measures, 1999). In order to parameterize this contribution, we couple the
 280 Mn contained in sediments in sea ice and the sea ice melt rate, I_{melt} :

$$S_{sed\ ice} = \frac{\alpha_0 \cdot f_{Mn\ sed}}{m \cdot \Delta z_{surface}} \cdot S_p \cdot I_{melt} \quad (8)$$

281 where S_p is the sediment content in sea ice at each grid point. The sediment content is
 282 spatially variable, and depends on the amount of sediment that was incorporated dur-
 283 ing ice formation on the shelves and on sea ice transport.

284 Through particle tracking experiments with Ocean Parcels (Lange & van Sebille,
 285 2017), we estimated the contribution of sea ice formed over the Siberian shelves during
 286 the stormy fall months (September-December) to the ice in the Canada Basin. We re-
 287 leased parcels every month over the course of a year and traced them backwards for three
 288 years (the average sea ice age in the Canada Basin and the northwestern CAA based on
 289 satellite information). Almost 40% of the sea ice tracks in the northwestern CAA and
 290 Canada Basin region originated from the Siberian shelves via the transpolar drift dur-
 291 ing the fall months, when strong sediment resuspension events coincide with sea ice for-
 292 mation (Fig. 4). The results of the particle tracking experiments were interpolated and
 293 smoothed to create a forcing field which incorporates the spatial variation in sediment
 294 content in sea ice (Fig. S6). In addition, we assumed a low background value of shelf sed-
 295 iments in sea ice in the Archipelago.

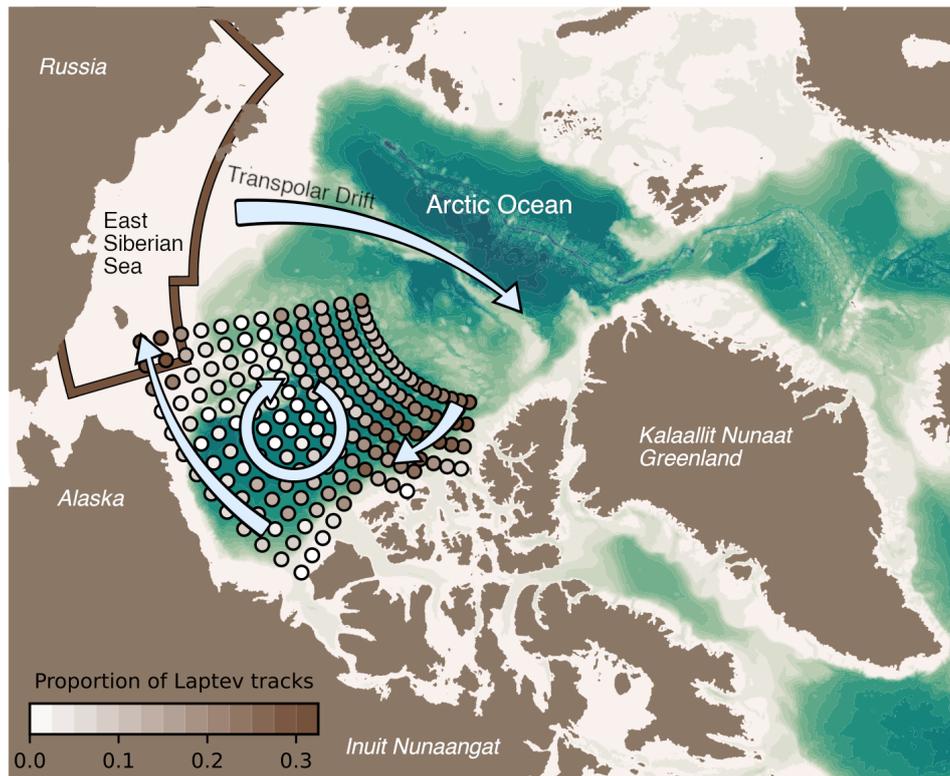


Figure 4. Sediment rich sea ice, produced over the Siberian shelves (East Siberian Sea) in the fall, is transported across the Arctic Ocean via the transpolar drift. From there it is found predominantly along the outer edges of the Beaufort Gyre. As a result, the largest contribution from the Siberian shelves occurs in the northeastern CAA. Ice motion patterns are indicated with light blue arrows. For locations in the Canada Basin, the proportion of parcels traced back to the Siberian shelves (region defined with the brown outline) in the fall months are shown.

296 **2.2.5 Reversible Scavenging and Sinking**

297 Dissolved Mn oxidizes forming larger aggregates and adsorbs to particle surfaces.
 298 dMn is regenerated by the reduction of oxidised Mn and desorption from particles. Since
 299 we do not directly model particle-bound Mn, but rather incorporate its effect on dMn
 300 through dissolution from the source components, we calculate the reversible scavenging
 301 based on the dissolved and oxidised Mn concentrations (Van Hulst et al., 2017):

$$R_{scav} = -k_p \cdot [dMn] + k_d \cdot [oMn] \quad (9)$$

302 where k_p is the adsorption and oxidation rate, and k_d is the desorption and reduction
 303 rate (see Text S1 for the full derivation). The R_{scav} term appears with opposing signs

304 in the dMn and oMn equations (Eqn. 1 and 2). We estimate the rate constants from ob-
 305 servations of dissolved and particulate Mn in the Canadian Arctic (Li, 2017; Colombo
 306 et al., 2020). Assuming steady state, the ratio of the scavenging rates is equal to the ra-
 307 tio of observed dissolved to particulate Mn concentrations. This assumption reduces the
 308 available observations to those far away from sources and sinks, i.e. deep stations in Baf-
 309 fin Bay and the Canada Basin (Fig. S1). The ratio of scavenging rates, k_p/k_d , is esti-
 310 mated to be 1.48 and with a k_d of $4.72 \cdot 10^{-7} \text{ s}^{-1}$ (Bruland et al., 1994), k_p is estimated
 311 as $7.00 \cdot 10^{-7} \text{ s}^{-1}$ (Fig. S2). To account for photoreduction, the reduction rate, k_d , in-
 312 creases from the base rate up to $2.72 \cdot 10^{-5} \text{ s}^{-1}$ near the surface, proportional to the so-
 313 lar flux that penetrates into the ocean (calculated by ANHA12). The scavenging rates
 314 in the model do not depend on the dissolved oxygen concentration since Arctic waters
 315 are generally well oxygenated.

316 The oxidised Mn aggregates sink, R_{sink} , and are removed through burial as in Van
 317 Hulten et al. (2017):

$$R_{sink} = s_{ox} \frac{\partial[oMn]}{\partial z} \quad (10)$$

318 where s_{ox} is the sinking rate.

319 **2.3 Tuning**

320 The model is tuned by altering the sediment content in sea ice and the sediment
 321 resuspension rate constant. The resultant sediment content in sea ice in the Canada Basin
 322 in our model ranges from 0 to 157 g m^{-3} and the average is 32 g m^{-3} . In observations,
 323 the sediment load ranges by several orders of magnitude depending on the location sam-
 324 pled, the type of ice, and is highly variable year-to-year (see Table S1 for a non-comprehen-
 325 sive list of observed sediment content). In the Beaufort Sea, the observed sediment con-
 326 tent in ice cores ranged from 31 to 593 g m^{-3} with an average of 157 g m^{-3} (Reimnitz
 327 et al., 1993). Our tuned ice sediment content is smaller than observed in that study, but
 328 of a similar order of magnitude. The sediment resuspension rates in our model range from
 329 0 to $4425 \text{ g m}^{-2} \text{ yr}^{-1}$ and the average is $149 \text{ g m}^{-2} \text{ yr}^{-1}$. Particulate material collected
 330 in sediment traps over the Beaufort Shelf from spring 1987 to 1988 contained total dry
 331 weight particle fluxes ranging from 20 to $140 \text{ g m}^{-2} \text{ yr}^{-1}$ (O'Brien et al., 2006). The larg-
 332 est particle fluxes occurred during the summer and fall. Our average tuned sediment re-
 333 suspension rate is at the upper end of this range.

2.4 Experimental Design

Three numerical experiments were performed with the Mn model, running from 2002 to 2019: the reference and “clean” sea ice cases, and a sensitivity experiment for the rivers. The reference run includes all model components and uses a lower bound estimate of the river contributions (no particle-bound Mn, $\beta = 0$ in Eqn. 3). The clean sea ice case is the same as the reference run, except that the sea ice does not contain sediment (i.e. $S_{sed\ ice} = 0$). In order to bound the riverine influence, we perform a sensitivity experiment with a distinctly upper bound riverine estimate ($\beta = 1$ in Eqn. 3) and compare this with the reference experiment which provides a lower bound estimate. The treatment of riverine input of Mn introduces uncertainties in the model due to the complex estuarine cycling and the influence of particulate matter on dissolved Mn concentrations. In the “upper bound” river experiment, we include the contribution from riverine sediments on the Mn concentrations in addition to the dissolved Mn. Each experiment is spun up by repeating the year 2002 seven times, before starting the full run. The run is considered spun up when the year-to-year change in Mn profiles is minimal (Fig. S9). Analysis was performed using Python 3 (<https://anaconda.com>) within Jupyter Notebooks with the NumPy, Pandas, SciPy, Matplotlib, Seaborn, scikit-learn, and cmocean packages (Pedregosa et al., 2011; Hunter, 2007; Kluyver et al., 2016; Oliphant, 2006; The Pandas development team, 2020; Thyng et al., 2016; Virtanen et al., 2020; Waskom & the Seaborn development team, 2020).

3 Results

The Mn profiles throughout our domain are typical for a scavenged type element: concentrations are higher near sources with a low and homogeneous background (Fig. 5). The background concentrations are controlled by scavenging, advection and mixing, and the resultant redistribution of materials throughout the water column, while the surface Mn maximum is a result of the contributions from river runoff, sea ice melt, dust deposition, photoreduction, and sediment that is resuspended directly into the polar mixed layer. Sediment resuspension leads to near-bottom increases in some regions.

362

3.1 Model Evaluation

363

364

365

366

367

368

369

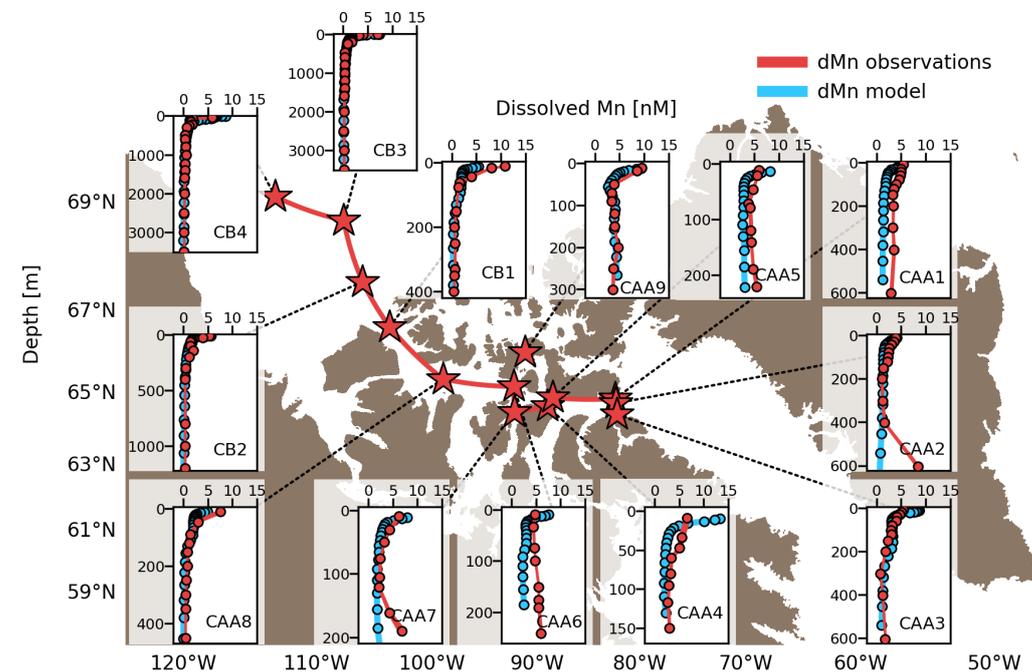


Figure 5. Simulated Mn profiles (blue) from the reference run compared to observed concentrations (red) measured during Canadian GEOTRACES cruises in September, 2015. Station locations are marked by red stars and the station names are labelled on the profiles. The thickness of the red line is the standard deviation of the measurements of the Mn samples.

370

371

372

373

374

The modelled Mn concentrations are highest at the surface and low in the background, as in the observations (Fig. 5). The model captures the regional variation of Mn concentrations along a transect from the Canada Basin into the Canadian Arctic Archipelago (Fig. 6). From 0-900 km along this transect (the Canada Basin), background concentrations are low (0-2 nM) and increase in the upper 300 m of the water column to

375 7-12 nM. At station CB4 in the Canada Basin, both the model and observations indi-
 376 cate high surface concentrations that extend deeper into the water column relative to
 377 the rest of the Canada Basin (Fig. 6). Surface concentrations at CB3 are the highest ob-
 378 served in the central part of the Canada Basin. In the model, a band of high concentra-
 379 tions is also found at the surface around station CB3. At 100-200 m depth in the Canada
 380 Basin, observed Mn concentrations are slightly higher than the background concentra-
 381 tions. This increase is associated with the winter Bering Sea Water and is not captured
 382 by the model. From 1300-2000 km along the transect (the eastern CAA), the background
 383 Mn concentrations are higher, ranging from 1-4 nM, due to the prevalence of resuspen-
 384 sion and other Mn sources. The increased role of sediment resuspension is particularly
 385 noticeable in the near-bottom increase in concentrations in the western CAA (stations
 386 CAA4-7; Fig. 6). Surface concentrations at CB1 and CAA8 in the Western CAA are un-
 387 derestimated (Fig. 5) and receive outflow from the Canada Basin. In the eastern CAA,
 388 surface concentrations range from 5-10 nM in observations and in the model. However,
 389 in the central CAA, surface concentrations are overestimated at stations CAA3-6, mostly
 390 at the southern side of Parry Channel and downstream from shallow regions with strong
 391 resuspension (Fig. 5). In Lancaster Sound, the model captures the higher surface con-
 392 centrations at CAA3 on the south side of the channel relative to CAA2 and CAA1.

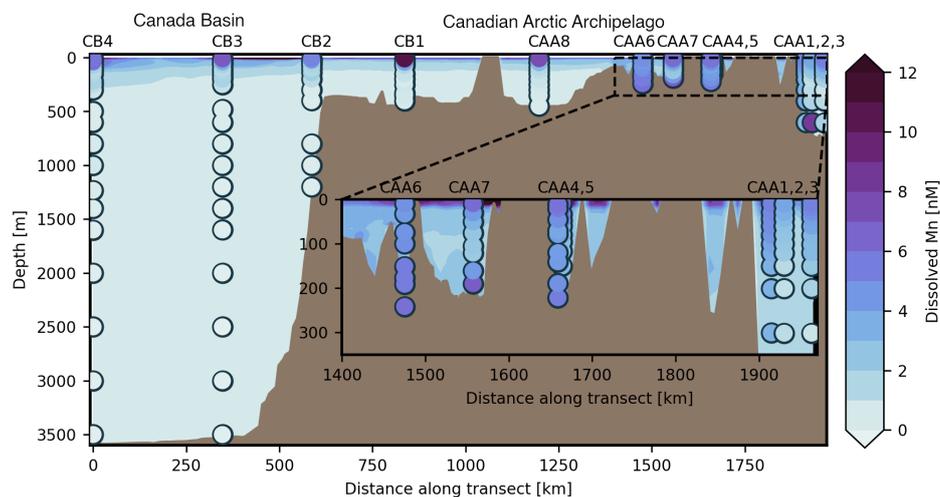


Figure 6. A transect of Mn concentrations from the Canada Basin through Parry Channel in the Canadian Arctic Archipelago to Baffin Bay (path shown in Fig. 5). Observations of dissolved Mn are shown within the circles, with the simulated concentrations for September, 2015 in the background. The inset expands on the Parry Channel region east of Barrow Sill.

393 The low and homogeneous background Mn concentrations are well represented in
394 shallow (< 600 m) as well as deep stations (1000-3500 m depth). However, the background
395 concentrations are underestimated at stations CAA1 and CAA6 in the central and east-
396 ern portions of Parry Channel (Fig. 5). At these stations, strong mixing results in con-
397 stant, “vertical” Mn profiles (Hughes et al., 2018). During tuning, we saw that in these
398 well-mixed regions, a locally higher sediment resuspension rate increased dMn concen-
399 trations from the bottom up to about 50 m depth where the surface stratification lim-
400 its vertical mixing. At CAA1 and CAA6, although the high tidal speeds force some sed-
401 iment resuspension to the west upstream in the model (Fig. 8b), it does not directly im-
402 pact these stations. CAA1 also receives Baffin Bay water that recirculates into Lancaster
403 Sound which is low in Mn content in the model. Nevertheless, sediment resuspension is
404 well-represented in other regions such as station CAA9 in Penny Strait (sediment resus-
405 pension is the main source of Mn here; Fig. 8b), an area known for its strong tidal speeds
406 and mixing (Hughes et al., 2018). At stations CAA2 and CAA7, on the south side of Parry
407 Channel, observed Mn concentrations increase up to 10 nM near the ocean bottom. These
408 peaks in the observations are attributed to sediment resuspension (Colombo et al., 2020),
409 although the specific mechanism is unclear. The model does not reproduce these local
410 extreme increases, which likely vary on much smaller spatial scales than our parameter-
411 izations can resolve. A small increase over the 40 m above the bottom is reproduced by
412 the model at stations CAA5 and CAA7.

413 **3.2 Importance of Sediment in Sea Ice**

414 In the upper water column (above 50 m), the observed Mn concentrations are strongly
415 variable and range from 2-11 nM. The “clean” sea ice experiment underestimates these
416 concentrations by several nM (the mean of this experiment falls 3 nM below the 1:1 line),
417 especially when observed concentrations exceed 6 nM (Fig. 7a). The reference run, which
418 includes sediment in sea ice, represents observed concentrations reasonably well for con-
419 centrations below 6 nM and the mean falls close to the 1:1 line. In the eastern CAA, the
420 reference run somewhat overestimates concentrations between 6-7 nM (especially near-
421 est the surface), and underestimates the observed concentrations above 7 nM in the west-
422 ern CAA (stations CB1 and CAA8) which receive relatively direct outflow from the Arc-
423 tic Ocean. In the upper bound river experiment, surface Mn concentrations are overes-
424 timated at most stations in the CAA, particularly in the eastern CAA, with a slight im-

425 provement at CAA8 in the western CAA (Fig. 7b). In the subsurface, CAA1, CAA2, and
 426 CAA5 in the eastern CAA are somewhat better represented. The Mn estimates for sta-
 427 tions in the Canada Basin are unaffected by the addition of Mn originating from par-
 428 ticulate matter in rivers.

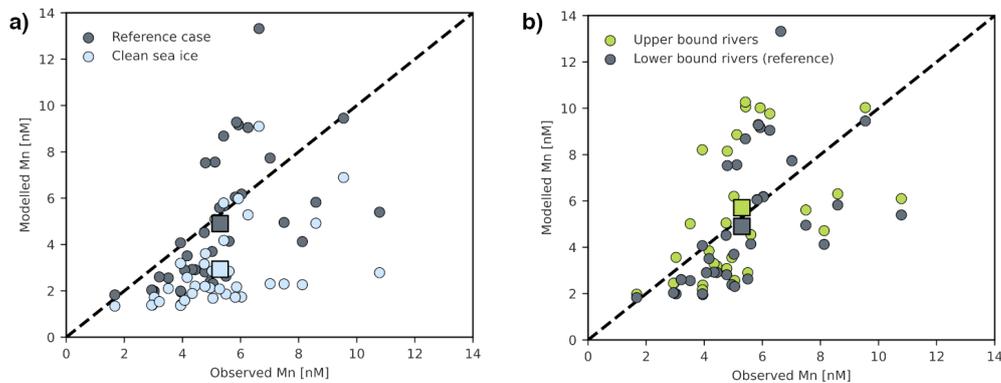


Figure 7. The nearest-depth modelled Mn concentrations compared with the observed concentrations at depths shallower than 50 m for all stations. Square markers indicate the averages of the experiments and observations. (a) The modelled Mn concentrations at the evaluation stations most closely resemble the observations in the reference experiment with sediment in sea ice compared to the “clean” sea ice experiment. Both of these experiments use the lower bound river estimate. (b) The lower and upper bound river experiments, which include sediment within the sea ice, indicate that a small additional contribution from riverine particulate matter content may be most representative.

429 In the near surface (around 10 m depth), observed concentrations range from 5-
 430 7 nM in the Canada Basin, up to 11 nM at CB1, while elsewhere in Parry Channel they
 431 are around 5 nM (Fig. 5 and 6). Within Parry Channel, the environment east of Bar-
 432 row Sill is different from the regions west of Barrow Sill which share characteristics with
 433 the Canada Basin (Colombo et al., 2020). The model also displays this trend, however,
 434 the model overestimates the surface concentrations at some stations in eastern Parry Chan-
 435 nel (CAA4-6), particularly in the upper bound river experiment. We expect substantial
 436 gradients in concentrations in the surface layer in the Arctic Ocean as a result of the strong
 437 stratification. The shallowest observations are collected at around 10 m below the sur-
 438 face, while the model surface estimate is at 0.5 m depth, so it is difficult to assess the
 439 uppermost modelled concentrations. However, the Mn-salinity relationship within the

440 upper 50 m of the water column in the model is similar to the observations for the ex-
441 periment with sediment in sea ice (Fig. S10). Without sediment in sea ice, the model un-
442 derestimates the low-salinity Mn endmember.

443 While the model is limited in its representation of regions with strongly variable
444 resuspension rates, it performs well within a range of environments: from deep regions
445 in the Canada Basin to shallow areas in the CAA. The model is configured to ask ques-
446 tions about the drivers of Mn variability; it is important to keep in mind that our pa-
447 rameterizations are limited by the spatial and temporal resolution of available informa-
448 tion, so small scale variations are unlikely to be captured by the model.

449 **3.3 Contributions from External Sources of Mn**

450 To assess the relative contributions of each of the external Mn sources, we calcu-
451 lated the annual contribution and flux from these model components in the reference ex-
452 periment. We are most interested in the surface layer, so our estimate is for the upper
453 55 m of the water column. An estimate of the full water column differs by including the
454 effects of resuspension in regions deeper than 55 m, thus increasing the importance of
455 resuspension (Table S2). Estimates from the upper bound river experiment, which does
456 not account for any removal of particulate or dissolved Mn in estuaries, are indicated in
457 brackets. By calculating the Willmott skill score (Willmott, 1981), we estimate the op-
458 timal riverine suspended matter content for the model at the observed locations as $\beta =$
459 0.05 in Eqn. 3 (Willmott of 0.68). We did not include the contributions from (photo)reduction
460 as a source of dMn in these calculations since it is part of the internal cycling of Mn. In
461 order to identify regional differences, we separated the domain into the Canada Basin
462 and the Canadian Arctic Archipelago (details in Fig. S3) and subdivided the CAA into
463 western and eastern halves along 100°W. Overall, the Canada Basin is more isolated and
464 receives a lower annual contribution of Mn than the CAA: 146 (162) versus 421 (547)
465 $\mu\text{mol m}^{-2} \text{ yr}^{-1}$ (Table 2).

466 In our model, the dominant source of Mn in the Canada Basin is the release of sed-
467 iment by sea ice melt (Table 2). This component accounts for 82 (74)% of the average
468 yearly addition of Mn. The amount of melt that occurs fluctuates interannually, simi-
469 lar to sea ice extent changes observed with satellite data. Nevertheless, from 2002 to 2019,
470 sea ice melt is consistently the largest contributor of Mn in our model in the Canada Basin.

471 Sediment resuspension contributes about 15 (13)% in the Canada Basin, mainly over the
 472 Beaufort shelf, and river discharge, predominantly from the Mackenzie River, contributes
 473 3.6 (13)%. Atmospheric dust deposited onto the ocean surface, or released during sea
 474 ice melt, is not a significant source of Mn anywhere in the domain.

Table 2. The spatial average annual dissolved Mn contributed by external model source components to the upper 55 m of the water column ($\mu\text{mol m}^{-2} \text{yr}^{-1}$) in the reference experiment, averaged over the years 2002-2019, separated by region (Fig. S3). Sediment release by sea ice is the only component that varies significantly year-to-year. Estimates from the upper bound river contribution experiment are indicated in brackets.

Component contribution	Canada Basin		Canadian Arctic Archipelago	
	$\mu\text{mol m}^{-2} \text{yr}^{-1}$	%	$\mu\text{mol m}^{-2} \text{yr}^{-1}$	%
River discharge	5.2 (21)	3.6 (13)	15 (141)	3.5 (26)
Sediment resuspension	21	15 (13)	361	86 (66)
Sediment from sea ice	120	82 (74)	44	11 (8.1)
Dust released by sea ice	0.2	0.1	0.3	0.1
Direct dust deposition	0.0	0.0	0.0	0.0
Total	146 (162)	100	421 (547)	100

Table 3. Same as Table 2, but with the Canadian Arctic Archipelago (CAA) subdivided into western and eastern halves along 100°W (near Barrow Sill).

Component contribution	Western CAA		Eastern CAA	
	$\mu\text{mol m}^{-2} \text{yr}^{-1}$	%	$\mu\text{mol m}^{-2} \text{yr}^{-1}$	%
River discharge	5.8 (24)	1.8 (6.8)	22 (228)	4.4 (33)
Sediment resuspension	273	83 (79)	427	87 (61)
Sediment from sea ice	48	15 (14)	41	8.4 (5.9)
Dust released by sea ice	0.2	0.1	0.3	0.1
Direct dust deposition	0.0	0.0	0.0	0.0
Total	328 (345)	100	490 (696)	100

475 In the CAA, sediment resuspension contributes 86 (66)% of the annual external
476 addition of Mn to the water column (Table 2). Sediment released by sea ice accounts for
477 11 (8.1)% of Mn; a combination of relatively “clean” sea ice with high melt rates. The
478 importance of river contributions cover a broader range from 3.5 (26)% in the CAA, com-
479 pared to 3.6 (13)% in the Canada Basin, however since these regions differ in total an-
480 nual contributions, rivers contribute more dMn to the CAA overall. Although the Canada
481 Basin receives runoff from the Mackenzie River, the CAA has many rivers of a range of
482 sizes that drain into it, including glacial rivers with high Mn concentrations.

483 Within the CAA, there is a significant difference in dynamical regime west and east
484 of the approximately 120 m deep Barrow Sill (Table 3; Hughes et al. (2017); Colombo
485 et al. (2020); Q. Wang et al. (2012)). The overall contribution of Mn to the water col-
486 umn in the eastern CAA is 490 (696) $\mu\text{mol m}^{-2} \text{ yr}^{-1}$, compared to 328 (345) $\mu\text{mol m}^{-2} \text{ yr}^{-1}$
487 in the western CAA. The main contributor to this difference is the 1.5 times stronger
488 sediment resuspension in the eastern CAA. In addition, rivers contribute more strongly
489 to the eastern CAA relative to the western CAA, 4.4 (33)% versus 1.8 (6.8)%, with a
490 broad range in the estimate of their role in the eastern CAA. The eastern CAA receives
491 contributions from the high Mn content glacial rivers that drain Greenland, Ellesmere
492 Island, and Baffin Island.

493 Throughout our domain, Mn concentrations are highest in the summer months as
494 a result of seasonally fluctuating components (Fig. 8a). Sea ice melt is largest in July,
495 while the river runoff peak occurs during the freshet in May-June. Due to the large sup-
496 ply of dissolved Mn in the summer months and the increased solar flux, (photo)reduction
497 and oxidation are stronger from July through September. For the month of September,
498 we identified which model component is on average the dominant control of Mn for each
499 horizontal grid cell over the full time series (Fig. 8b). Note that this figure shows where
500 the model adds the contribution from a component; where the Mn ends up depends on
501 the advection and diffusion of the tracer as well.

502 Within the Canada Basin and portions of the western CAA (the Amundsen Gulf
503 and western Parry Channel), sea ice melt controls the simulated Mn concentrations (Fig. 8b).
504 In the interior of the Beaufort Gyre region, far away from sources and with relatively
505 “clean” sea ice, none of the components contribute significantly. Over the Beaufort Shelf,
506 the Mackenzie River is a regionally important source of Mn; generally river runoff is a

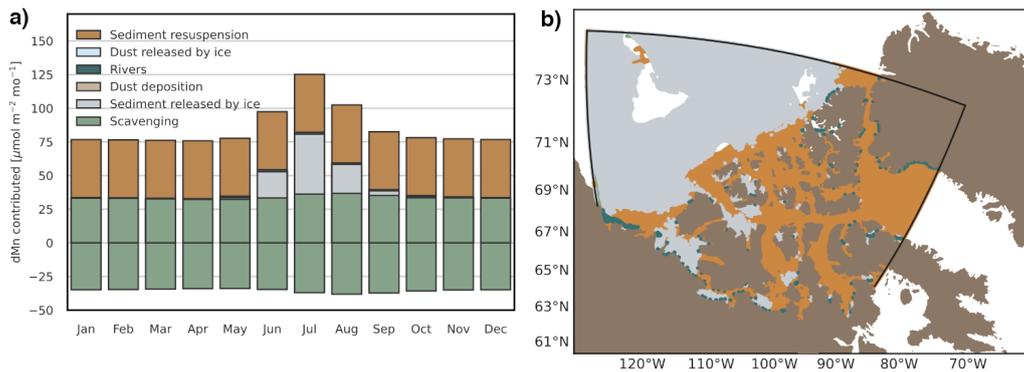


Figure 8. In the full water column, sediment released by sea ice dominates in the Canada Basin, while sediment resuspension is prevalent over shelf areas including the Canadian Arctic Archipelago. (a) Climatology of the seasonal cycle of Mn contributions for the full water column. The oxidation (removal) and reduction (addition) of Mn through scavenging are calculated as the average throughout the water column. Sediment resuspension is added at the bottom grid cell, while all other sources act directly on the ocean surface. Note that the contributions from dust deposition and release of dust from ice are too small to appear. (b) Most important Mn contributors to the water column in September based on climatology. At each grid cell, the color represents the most important model forcing component. The importance of scavenging is based on the average net combined effect of reduction and oxidation throughout the water column. Places where the net contributions are smaller than $0.5 \mu\text{mol m}^{-2} \text{mo}^{-1}$ are masked with white.

507 significant source at river mouths. In the shallower shelf regions, such as the Beaufort
 508 Shelf and the CAA, sediment resuspension is prevalent.

509 The magnitudes of annual Mn fluxes from sources in this Arctic Model (AM; Ta-
 510 ble 2) are comparable to those in the first global model of Mn by Van Hulst et al. (2017)
 511 (VH). In the global model, dust contributes $0\text{-}2 \mu\text{mol m}^{-2} \text{yr}^{-1}$ in the Arctic Ocean, whereas
 512 in AM it ranged from $0\text{-}1 \mu\text{mol m}^{-2} \text{yr}^{-1}$ (combining direct dust deposition from the
 513 atmosphere and indirect release from ice). AM riverine fluxes were $5.2 (21) \mu\text{mol m}^{-2} \text{yr}^{-1}$
 514 in the Canada Basin and $15 (141) \mu\text{mol m}^{-2} \text{yr}^{-1}$ in the CAA; higher than the VH es-
 515 timate of $0\text{-}2 \mu\text{mol m}^{-2} \text{yr}^{-1}$. This range likely reflects a combination of the high Mn
 516 content of rivers in the Arctic used in this model (Colombo et al., 2019) and alternate
 517 treatment of rivers; VH assumes a relation between Fe and Mn content, while AM uses
 518 observations specific to the Arctic rivers and their catchment basins. In the global model,

519 the flux of Mn from bottom sediments in the Arctic Ocean was 5-75 $\mu\text{mol m}^{-2} \text{yr}^{-1}$; AM
520 has 21-361 $\mu\text{mol m}^{-2} \text{yr}^{-1}$. The difference in the upper limit of the range likely reflects
521 the distinctive processes considered by the models: the global model considers sediment
522 diffusion for the flux from sediments, whereas AM considers sediment resuspension be-
523 cause it is more important in the CAA (Colombo et al., 2020). It is also challenging to
524 resolve the large continental shelf regions in the Canadian Arctic in a global model. Lastly,
525 on a global scale, hydrothermal input of Mn at spreading ridges is important (Van Hul-
526 ten et al., 2017), however the spreading ridges in the Arctic are far away from the AM
527 domain, so that contribution is not included.

528 **3.4 Simulated Surficial Mn During the Summer and the Polar Night**

529 The most significant seasonal and interannual changes in Mn concentrations oc-
530 cur in the polar mixed layer, defined here as the upper 35 m of the water column. For
531 the following characterizations of the simulated concentrations, we will focus on this layer.
532 The upper few meters of the ocean have a strong gradient in Mn concentrations (sim-
533 ulated profile in Fig. S12). It is not possible to measure this layer with conventional meth-
534 ods from a large ship. As such, we exclude the surface 1 m in the results presented here
535 (see Fig. S11 for the surface Mn field) to allow for more direct comparison with exist-
536 ing observations.

537 During the summer months, the surface Mn concentrations in the Canada Basin
538 mirror the areas of strong sea ice melt (Fig. 9a). The highest Mn values are found along
539 the outer edges of the Beaufort Gyre (up to 14 nM), while the interior has lower con-
540 centrations of Mn. The Beaufort Shelf does not receive a strong contribution of Mn from
541 sea ice melt (Fig. 8b). The location of the areas with highest sea ice melt contributions
542 in the Canada Basin shift from year-to-year with the position of the Beaufort Gyre.

543 Although rivers contribute only a few percent annually to Mn in the Canada Basin
544 (Table. 2), over the continental shelf, plumes of high Mn concentrations extend along the
545 coastline in the summer months (Fig. 9a). The plume from the Mackenzie River, the largest
546 river in our domain, extends eastward along the shelf in July, 2015. In the southwest-
547 ern CAA, high Mn concentrations are prevalent during the summer months in regions
548 fed by continental rivers (Fig. 8b). Glacial river drainage is apparent in surface Mn con-
549 centrations in a number of coastal regions (Fig. 9a). A plume of runoff from Baffin Is-

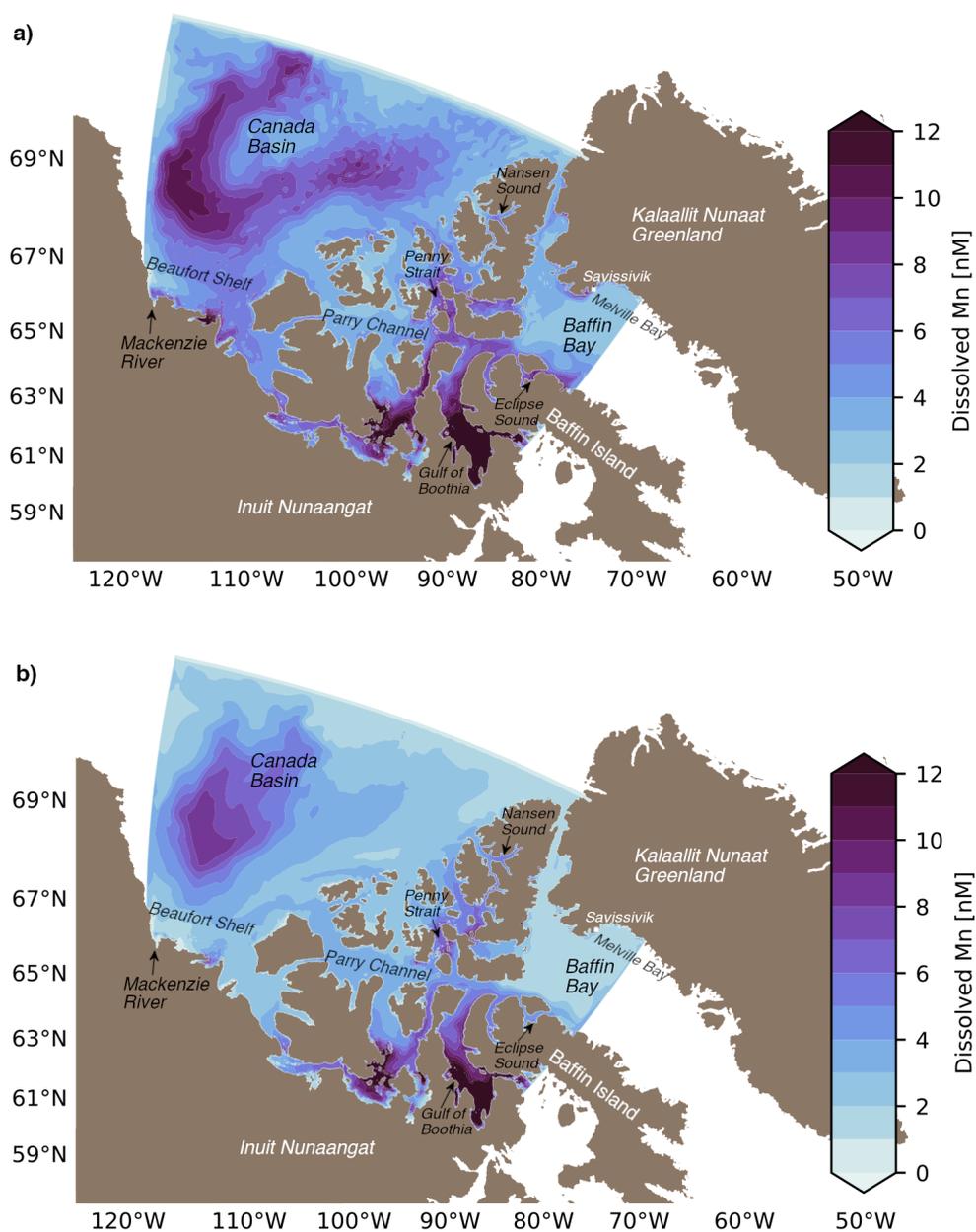


Figure 9. Simulated monthly average Mn concentrations in the Polar Mixed Layer. (a) July, 2015. In the summer, sea ice melt and sediment resuspension dominate the Mn concentrations in the Canada Basin and the Canadian Arctic Archipelago, while freshwater sources such as the Mackenzie River and Greenland meltwater are important regionally. (b) January, 2015. During the Polar night, simulated Mn concentrations are more homogeneous and low, however sediment resuspension continues to drive higher concentrations in the Gulf of Boothia and south-central CAA.

550 land extends from Eclipse Sound into Baffin Bay, where it is incorporated into the West
551 Greenland Current. Along the coast of Greenland, high concentration Mn rivers drain
552 the ice sheet and a number of plumes are visible. A notable plume is located near Savis-
553 sivik into Melville Bay and extends along the shoreline. In the northern CAA, higher
554 surface concentrations are a combination of the sea ice melt and glacial rivers (Fig. 8b).
555 In particular, Nansen Sound receives large contributions from runoff from Ellesmere Is-
556 land.

557 Mn concentrations exhibit strong spatial variability in the CAA (Fig. 9). In the
558 western CAA, concentrations are low (2-6 nM) and homogeneous. Southern regions, in-
559 cluding the Gulf of Boothia, have some of the highest concentrations (8-12+ nM) and
560 flow into the Parry Channel east of Barrow Sill. In this section of central and eastern
561 Parry Channel (and in Penny Strait), intermediate concentrations (6-9 nM) are present.
562 In Lancaster Sound, the outflow from Parry Channel, with relatively high Mn concen-
563 trations, follows the southern half of the channel while lower concentration waters from
564 Baffin Bay (1-4 nM) recirculate along the northern half of Lancaster Sound. Baffin Bay
565 and the Nares Strait regions are characterized by lower surface concentrations.

566 During the Polar Night, fewer sources contribute Mn and there is less spatial con-
567 trast in surface concentrations (Fig. 9b). In the winter, surface concentrations range from
568 1-10 nM (excluding the Gulf of Boothia region), while in the summer they ranged up to
569 14 nM. Regions where Mn is most impacted by sediment resuspension (Fig. 8b), such
570 as the Gulf of Boothia, still have high concentrations in the winter as this component
571 does not vary seasonally. In the Canada Basin, a core of 4-10 nM concentrations is lo-
572 cated towards the center of the Beaufort Gyre region, while during the summer months,
573 the high concentrations are located around the outer edges (Fig. 9). The winter core forms
574 from the advection of the summer Mn signal during the winter months when surface in-
575 put in the Canada Basin is low, and from the reduction of oxidised Mn which was trans-
576 ported from regions with high summer dissolved Mn concentrations. The signature of
577 this winter core reduces in the spring and early summer as the seasonal sea ice melt once
578 again dominates the signal.

4 Discussion

In the Arctic Ocean, the maximum Mn concentrations occur near the surface in the polar mixed layer. These high concentrations are commonly attributed to freshwater sources such as river discharge and sea ice melt (Campbell & Yeats, 1982; Yeats & Westerlund, 1991; Middag et al., 2011b; Cid et al., 2012; Kondo et al., 2016; Colombo et al., 2020). However, the relative contributions from rivers and sea ice to this low-salinity maximum are not easily distinguished. In this paper, we present a regional model of Mn in the Canadian Arctic which incorporates river input and sediment release by sea ice, as well as atmospheric inputs, sediment resuspension, and scavenging. With these components, the model captures the spatial variability and magnitude of observed concentrations and we are able to use the model to assess the controls on Mn in the Arctic. With results from three Mn model experiments (reference, “clean” sea ice, and upper bound river), we identified the dominance of non-local sediment released by sea ice in the Canada Basin, while rivers had a more regional importance. These findings suggest that future changes to sea ice transport across the Arctic Ocean may have a significant impact on the supply of Mn and other nutrients to the Canada Basin and downstream to the CAA. Within the CAA, the dynamical differences between the western and eastern CAA translated into distinctive mean Mn concentrations and component contribution patterns with more influence from rivers and sediment resuspension.

4.1 Ice-rafted Sediments are the Predominant Source of Mn in the Canada Basin

With our model, we found that 82 (74)% of Mn in the Canada Basin (the main estimate is from the reference experiment with the upper bound river estimate in brackets) is supplied by sediment from sea ice and 11 (8.1)% in the CAA (Table 2). Sediments released by sea ice melt dominate the Mn concentrations in the polar mixed layer during the summer months (Fig. 9a), while in the winter, sea ice blocks the direct surface input of Mn and a relatively lower, more homogeneous distribution results (Fig. 9b). These results indicate that sediment transport and release by sea ice is the main source of Mn (and likely other similar nutrients) within the Canada Basin, and plays a role within the CAA as well. The sea ice in the interior of the Canada Basin originates from the Siberian shelf regions and traverses the Arctic Ocean via the transpolar drift. The sea ice spends several years in transit, during which it undergoes freeze-thaw cycles and loses sediment.

611 In the Canada Basin, the highest Mn concentrations (and relatively younger ice) are found
612 along the outer edges of the Beaufort Gyre, while older ice transported to the interior
613 of the Gyre by convergence has lower Mn concentrations (Fig. 9a). Sea ice is also formed
614 in the Beaufort Shelf region, however this ice is transported towards Siberia and does
615 not directly impact the Mn concentrations in the Canada Basin.

616 Mn sources from the land-ocean interface, such as rivers and sediments, were more
617 important in the CAA than in the Canada Basin, and dynamical differences between the
618 western and eastern CAA translated into distinctive Mn concentrations and component
619 contribution patterns. This separation in dynamics is bounded by the ≈ 120 m deep Bar-
620 row Sill and has been noted in several studies (Hughes et al., 2017; Colombo et al., 2020).
621 In the western CAA, surface concentrations range from 3-6 nM (Fig. 9) and Mn com-
622 ponent contributions share characteristics with the Canada Basin: similar overall river
623 contributions, some influence from sediments in sea ice, and weaker contributions from
624 sediment resuspension than in the eastern CAA (Table 3). In contrast, in the eastern
625 CAA, Mn concentrations are high (6-9 nM; Fig. 9) and dominated by sediment resus-
626 pension associated with strong tidal speeds and river discharge plays a more important
627 role. The estimate of the component contributions is most sensitive in the eastern CAA:
628 the importance of rivers ranges from 4.4% in the reference experiment up to a maximum
629 of 33% in the upper bound river experiment. Rivers are prevalent in the eastern CAA
630 and many of these drain glaciated regions associated with high suspended particulate
631 matter and dissolved Mn. As a result, rivers have the potential to play an important role
632 for Mn in the eastern CAA. However, the available information for river input and es-
633 tuarine removal limits our ability to constrain the most likely river contribution. Based
634 on the model versus observation surface comparisons (Fig. 7), the upper bound river ex-
635 periment overestimates the contribution of rivers because it does not consider removal
636 in estuaries, but the lower bound river experiment underestimates it. The most realis-
637 tic value is somewhere in between. The uncertainties associated with these estimates high-
638 light the need for studies looking at the estuarine cycling in the CAA.

639 Besides sea ice melt, Pacific water inflow from the Bering Strait and river runoff
640 (Eurasian runoff and North American runoff) contribute freshwater to the Arctic Ocean
641 (Proshutinsky et al., 2019; Krishfield et al., 2014) and could contribute Mn to the sur-
642 face maximum. The central Canada Basin contains significant amounts of meteoric wa-
643 ter and sea ice melt (Guay et al., 2009) which feed its freshening (Yamamoto-Kawai et

644 al., 2009). Several studies have looked into the composition of this water. Fichot et al.
645 (2013) did not identify much river runoff in the central basin and Kelly et al. (2019) found
646 that the freshwater contribution from Siberian rivers has decreased since 1997 as a re-
647 sult of the mainly anticyclonic atmospheric circulation pattern over the Canada Basin.
648 Similarly, model trajectories of floats released from Siberian rivers since 1985 do not gen-
649 erally reach the Canada Basin by 2007 (Proshutinsky et al., 2019). In our reference and
650 upper bound river simulations, rivers contribute only 3.6 (13)% to the total budget of
651 Mn in the Canada Basin and 3.5 (26)% in the CAA (Table 2). However, freshwater sources
652 such as the Mackenzie River on the Beaufort shelf and glacial melt off the coast of Green-
653 land (Fig. 9) do dominate areas nearby coastlines. The supply of relatively fresh Pacific
654 Water from Bering Strait to the Canada Basin is also affected by the atmospheric cir-
655 culation in the Canada Basin (Kelly et al., 2019) and floats released from Bering Strait
656 since 2000 do not enter the central Canada Basin by 2012 (Proshutinsky et al., 2019).
657 Thus, inputs outside of our domain that originate from Siberian runoff and Pacific wa-
658 ter are unlikely to significantly contribute to the freshwater-associated surface Mn max-
659 imum in the Canada Basin. It is important to note that our simulated profiles (Fig. 5)
660 do not capture the subtle increase in Mn concentrations associated with the winter Bering
661 Sea Water around 100-200 m depth in the Canada Basin. This limitation is likely be-
662 cause our western boundary condition does not fully capture the higher concentrations
663 of Mn found in the Alaskan Coastal Current and in waters from the Chukchi Shelf. This
664 difference mainly affects the Beaufort Shelf and as mentioned the winter Bering Sea Wa-
665 ter layer in the Canada Basin.

666 In order to assess whether we overestimated the sediment content of sea ice, we per-
667 formed an experiment with “clean” sea ice. In the “clean” ice experiment, the surface
668 Mn concentrations are underestimated by 3 nmol L⁻¹ or about 30-50% in the Canada
669 Basin relative to observations (Fig. 7a). If we assume that all of the missing Mn comes
670 from sediment and that Mn added at the surface mixes down to the turbocline, we miss
671 a source that supplies 10-160 grams of sediment per squared meter to the surface ocean
672 across the Canada Basin (range based on model turbocline depths in 2015). The mag-
673 nitude of this component is similar to the average sediment load measured in sea ice cores
674 (Reimnitz et al., 1993; Stierle & Eicken, 2002; Eicken et al., 2005). Rivers would be un-
675 able to contribute the total amount missing since it must occur over a large area and since
676 the upper bound river experiment shows that additional contributions from rivers do not

677 significantly affect the Canada Basin or the overall surface representation (Fig. 7b). In
678 the “clean” sea ice experiment, the freshwater endmember of Mn is also underestimated
679 (Fig. S10). The Mn-salinity relationship in the Canada Basin and the CAA is more ac-
680 curately represented in the experiment with sediment contained in sea ice and the re-
681 gional differences are also reproduced.

682 Our results demonstrate that the long range transport of sediments by sea ice from
683 the Siberian shelves is an important source of Mn in the Canada Basin and the Cana-
684 dian Arctic Archipelago. These findings provide support for the sea ice trace metal trans-
685 port mechanism proposed by Measures (1999). Measures (1999) found that the highest
686 Al and Fe concentrations in the central Arctic Ocean coincided with areas with high con-
687 centrations of ice-rafted sediments, instead of river input, and so they hypothesized that
688 transport of ice rafted sediments and the subsequent seasonal melt supplies reactive el-
689 ements to the surface Arctic Ocean. However, their data set did not allow the quantifi-
690 cation of annual fluxes of material to the central Arctic Ocean and so they were unable
691 to quantify the exact contribution of this component to the observed trace metal con-
692 centrations.

693 **4.2 Declining Long Range Sea Ice Transport Could Reduce the Canada** 694 **Basin and Canadian Arctic Archipelago Nutrient Supply**

695 Based on the importance of non-local sediments transported by sea ice (particu-
696 larly from the Siberian shelves), the distributions of trace metals, nutrients, and their
697 biogeochemical cycles in the Arctic basins are likely to be significantly impacted by cli-
698 mate change associated reductions in sea ice. Rising oceanic and atmospheric temper-
699 atures delay the freeze-up period and induce earlier melt of sea ice (Stroeve et al., 2012;
700 Stroeve & Notz, 2018). In addition, in the relatively “quiet” dynamics of the Arctic Ocean,
701 increased mixing may bring warmer Atlantic water (or Pacific Water; Kodaira et al. (2020))
702 to the surface and further increase sea ice melt (D’Asaro & Morison, 1992; Liang & Losch,
703 2018). These factors may significantly reduce the amount of first-year ice that survives
704 in the Kara Sea, East Siberian sea, and western Laptev Sea (Krumpfen et al., 2019).

705 Investigations of long range transport by the transpolar ice drift indicate an increase
706 in exchange of ice-rafted material between regions, associated with faster ice drift from
707 thinning ice cover (Sprenen et al., 2011; Kwok et al., 2013; Kipp et al., 2018). However,

708 Krumpfen et al. (2019) found that summer ice extents in the marginal ice zones have been
709 low enough over the past few years, that most of the ice exported from shelves melts be-
710 fore it enters the transpolar drift. Only ice formed in polynyas between October and Jan-
711 uary was advected far enough to survive the following summer melt season, suggesting
712 a reduction in the transport of matter towards the central Arctic Ocean and Fram Strait.

713 In our study, we saw a steady increase in the Mn content of the Canada Basin po-
714 lar mixed layer from 2002-2019 (Fig. 10), and the primary source of this Mn is sea ice
715 melt (correlation R-squared of 0.57). Note that in our model experiments, we do not ac-
716 count for changes in sea ice supply regions, but focus only on sea ice melt. The addition
717 of Mn by melt in our model mirrors satellite observations of sea ice loss in the Beaufort
718 Sea (Fig. 10; correlation R-squared of 0.54). So, the short term increase in sea ice melt
719 may increase the Mn content of the Canada Basin. However, the long term reduction
720 in supply of sediment rich sea ice from the Siberian shelves via the transpolar drift and
721 the reduction in total sea ice volume would result in a decreased supply of Mn. Whereas
722 in the short term, there may be an increase in nutrients supplied by sea ice into the Canada
723 Basin through increased sea ice exchange and melt, in the long run, we expect a decrease
724 in supply and a subsequent decline in the surface maximum of Mn in the Canada Basin.
725 Confounding this is the likely increase in transport of riverine and shelf-derived trace el-
726 ements in the ocean by the transpolar drift as a result of an intensification of the Arc-
727 tic hydrological cycle and permafrost degradation (Charette et al., 2020).

728 A reduction in nutrient supply to the Canada Basin may also have an impact down-
729 stream in Baffin Bay. With our experiments, we calculated the transport of Mn through
730 Parry Channel and the contribution of sediment released by sea ice melt and rivers to
731 this transport (see Text S2 for details). About 45% of the net Mn transported into Parry
732 Channel from the Canada Basin is contributed by sediments from sea ice and 7% by rivers
733 (Fig. S4 and S5). Rivers contribute 4% to the Mn transported from Parry Channel into
734 Baffin Bay and sea ice contributes around 18%. The reduction in the contribution of these
735 components does not indicate loss in the CAA; it reflects the additional contributions
736 from sediment resuspension from the CAA shelf environment. The sea ice contribution
737 in the water column is significant downstream, however, it is important to note that the
738 sea ice transport in the CAA in the ocean-ice model is stronger than observed due to the
739 lack of a land-fast ice parameterization (Grivault et al., 2018). As a result, we may over-
740 estimate the sea ice transport and thus melt in Parry Channel, particularly for the out-

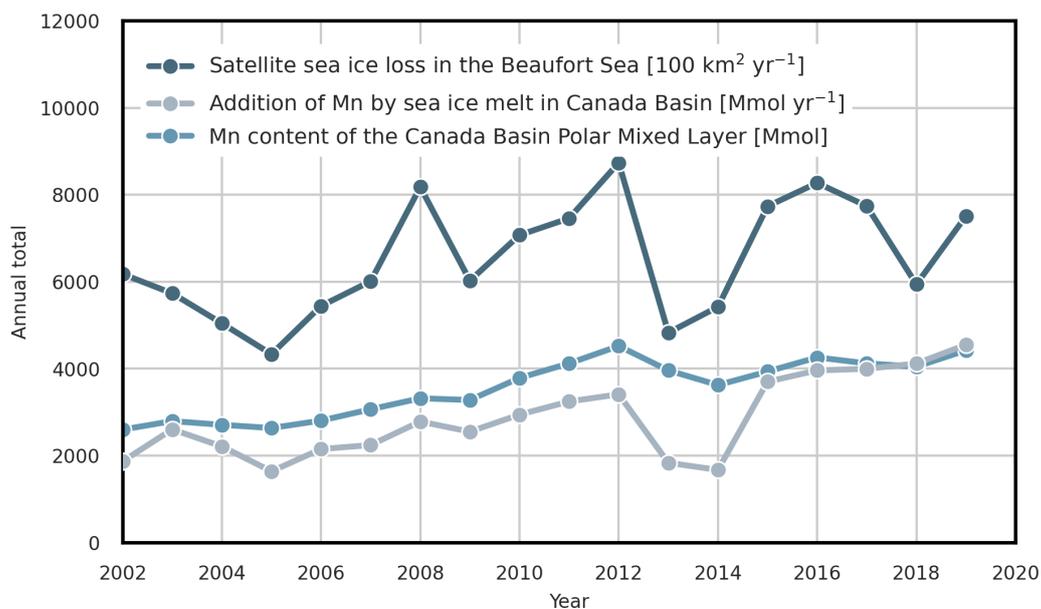


Figure 10. Interannual variations in sea ice melt contribute strongly to Mn supply to the Canada Basin. Conversely, surface Mn concentration changes in the Canada Basin can be used as an indicator of the volume of sediments released by sea ice melt. The sea ice loss is calculated from regional monthly sea ice extent changes measured by the Defense Meteorological Satellite Program series of passive microwave remote sensing instruments (Fetterer et al., 2017). The Mn added by external sources to the Canada Basin is calculated from outputs of the regional Mn model presented in this study.

741 flow from Parry Channel into Baffin Bay. There are also further factors contributing Mn
742 within the CAA which confound this finding. The acceleration of the hydrological cy-
743 cle and permafrost thaw may increase the contributions of riverine Mn to the CAA; our
744 experiments do not take these changes into account. On the other hand, sea ice melt is
745 associated with an increase in stratification which may reduce the depth up to which re-
746 suspended sediment can mix, reducing the Mn supplied into the upper water column (and
747 productive areas) by sediment resuspension in the CAA. However, reduced sea ice cover
748 is also associated with increased wind-driven mixing.

749 Our findings for Mn in the Arctic have implications for nutrients which share sim-
750 ilar sources. In the Arctic Ocean, iron (Fe) behaves similarly to Mn, although Fe is less
751 soluble than Mn and oxidizes more rapidly (Landing and Bruland (1987); Colombo et
752 al. (2020); for a comprehensive discussion, read Jensen et al. (2020)). Fe is an essential
753 micronutrient and in some regions of the ocean, such as the Southern Ocean, parts of
754 the North Atlantic, and the Pacific Northwest, it limits primary productivity (Martin
755 & Gordon, 1988; Hawkings et al., 2014; Tagliabue et al., 2017). Generally, iron is not
756 growth limiting in the Arctic (S. Wang et al., 2014), but there is evidence that Fe is lim-
757 ited in some specific regions: on the outer shelf and shelf break in the Bering Sea (Aguilar-
758 Islas et al., 2008), as well as in the Barents Sea and Nansen Basin (Rijkenberg et al., 2018).
759 Past studies have indicated that sea ice contributes to the flux of Fe into the ocean (Measures,
760 1999; Lannuzel et al., 2007; Aguilar-Islas et al., 2008; Kanna et al., 2020). Based on the
761 expected changes to the Mn cycle and supply with sea ice melt over the next decades,
762 the supply of Fe to the Canada Basin may be reduced as well. Meanwhile, the increase
763 in simulated Mn content in the Canada Basin from 2002-2019 due to sea ice melt may
764 also have supplied nutrients such as Fe and drive some of the observed increased Arc-
765 tic Ocean primary production (Lewis et al., 2020). Changes to Fe availability impact the
766 community composition and the timing of the spring phytoplankton bloom (Aguilar-Islas
767 et al., 2008), which in turn has consequences for biological productivity, Arctic ecosys-
768 tems, and the carbon cycle.

769 **4.3 Limitations of Results**

770 The findings in this study are limited by the parameterizations for scavenging, sed-
771 iment in sea ice, sediment resuspension, and river runoff. Overall, the model is best con-
772 strained for summer months, the southern CAA, and the Canada Basin due to the avail-

773 ability of observations. Scavenging rates are important throughout the water column and
774 are most likely to affect our results in coastal regions. We assumed steady state to es-
775 timate the adsorption and desorption rates from observations; this assumption is least
776 likely to hold in coastal regions and near the surface where scavenging rates are both im-
777 portant and variable. For the sediment released by sea ice, we did not account for vari-
778 ations in transport of sediment (and its origin) across the Arctic Ocean over the course
779 of the time series. Sea ice drift patterns vary interannually, and so could the source re-
780 gions for sediment transported to the Canada Basin by sea ice. The sediment content
781 would more accurately be represented as a time dependent variable. The total Mn con-
782 tent in the Canada Basin would increase (decrease) with a higher (lower) sediment con-
783 tent in sea ice, while sediment in sea ice would be more (less) important overall. How-
784 ever, observed sediment sea ice loads range several orders of magnitude by location sam-
785 pled and properties of the ice, and these fluctuations make it challenging to quantify an-
786 nual changes in overall sediment content and path travelled. Similarly, sediment resus-
787 pension varies interannually and seasonally and may be better represented as a time de-
788 pendent variable. We do not take into account the contributions from breaking of inter-
789 nal waves, storm generated currents, and surface waves on sediment resuspension. As
790 a result, we likely underestimate sediment resuspension contributions in some areas, par-
791 ticularly during the summer ice-free period. Our treatment of rivers was simplistic and
792 did not account for the complexity of transformations that occur in the estuarine zone.
793 Our results indicate a lower and upper bound of the riverine contributions, however we
794 are unable to indicate what the actual contribution is. We also did not account for the
795 projected seasonal ranges in riverine Mn concentrations with discharge (Colombo et al.,
796 2019); the river discharge varies seasonally, but the characteristic Mn concentrations of
797 the rivers are constant. This approximation could underestimate the riverine contribu-
798 tions during the spring freshet in coastal areas. The regions most likely to be impacted
799 by these limitations are within the northern CAA and Greenland coast, where glacial
800 rivers are most important.

801 While the numbers presented here should be taken as an estimate of magnitude
802 rather than as exact values, the key results are robust to the uncertainties described above.
803 The only way we were able to close the Mn budget (particularly in the Canada Basin)
804 was by incorporating the sediment in sea ice component. Similarly, the only way to rep-
805 resent the higher concentrations of Mn found in the lower water column at some stations

806 in the CAA was through the sediment resuspension term. Overall, while the model is
807 limited in its representation of these processes, it provides a platform to ask questions
808 about the drivers of Mn variability and to perform larger scale estimates of the processes
809 that contribute Mn to the Arctic Ocean. Improvements to the estimates of sediment con-
810 tent in sea ice from, for example, satellite products would strengthen future predictions,
811 while the model accuracy would be improved by more comprehensive estimates of the
812 scavenging and sediment resuspension rates. Observations of Mn along a transect from
813 an estuary into the ocean would help constrain the riverine contributions.

814 **5 Conclusions**

815 New trace metal datasets collected in the Arctic Ocean as part of the Canadian GEO-
816 TRACES program have provided an essential base for studying biogeochemical cycling
817 in this unique region. Using in situ observations from Colombo et al. (2020), we devel-
818 oped the first model of Mn in the Canadian Arctic Archipelago and the Canada Basin.
819 With three experiments from 2002-2019, we looked at (1) the drivers of Mn distributions
820 in the CAA and the Canada Basin and (2) implications of future sea ice transport changes
821 on the biogeochemical cycles of nutrients in the Arctic Ocean.

822 (1) While sediment transport by sea ice is identified as important in the Arctic Ocean
823 (Measures, 1999; Eicken et al., 2005), this mechanism is commonly considered less sig-
824 nificant for Mn than riverine input. However, without the contribution from sediment
825 in sea ice to Mn, we were unable to accurately represent the Mn concentrations in the
826 Canada Basin with our model. Sediments transported in sea ice by the transpolar drift
827 account for up to 82% of the total annual Mn added in the Canada Basin and up to 15%
828 in the western CAA, driving Mn surface maxima. These results support the hypothe-
829 sis that “ice-rafted sediment may be an important transport mechanism for supplying
830 reactive trace elements,” proposed by Measures (1999). Rivers are certainly locally im-
831 portant, but contribute only 3.6 (13)% annually in the Canada Basin. Within the CAA,
832 our estimates for river contributions ranged from 3.5% up to 26% in the upper bound
833 river experiment. This broad range is the result of the limited information available re-
834 garding estuarine cycling in the Arctic. A clear divide is present in the CAA: west of Bar-
835 row Sill, the mean concentrations are lower and the behaviour of Mn is more similar to
836 the Canada Basin, while in the eastern CAA, sediments resuspended by high tidal speeds,
837 as well as many glacial rivers drive higher Mn concentrations.

838 (2) Sea ice transport via the transpolar drift is interrupted by Arctic warming (Krum-
839 pen et al., 2019) and the decline in this long range transport could reduce the Canada
840 Basin and the CAA nutrient supply. These changes not only impact the Arctic, but also
841 sub-arctic seas, with 18% of the Mn transported from Parry Channel into Baffin Bay added
842 by sea ice melt. Mn behaves similarly to Fe in the Arctic Ocean and both of these nu-
843 trients support phytoplankton growth. The importance of sea ice for nutrient supply to
844 the photic zone in the Canada Basin, as well as downstream, is concerning given the re-
845 cent changes in the Arctic Ocean sea ice regime (reduced summer minimum ice extent,
846 ice thinning, reduction in multi-year ice extent, and altered drift paths). There are many
847 competing factors that will contribute to changes in the biogeochemical cycles; combined
848 model-observation studies are highly valuable to understand the individual contribution
849 of these factors.

850 **Acronyms**

851 **CAA** Canadian Arctic Archipelago

852 **NEMO** Nucleus for European Modelling of the Ocean

853 **ANHA12** Arctic and Northern Hemispheric Atlantic 1/12 degree

854 **LIM2** Louvain-la-Neuve version 2

855 **TOP** Tracers in the Ocean Paradigm

856 **TVD** Total Variance Dissipation scheme

857 **CESM** Community Earth System Model

858 **CAM-Chem** Community Atmosphere Model with Chemistry

859 **Acknowledgments**

860 We thank Marco van Hulst for openly sharing his model code and results, Jacqui-Lee
861 Epstein for extracting the tidal speeds for the sediment resuspension parameterization,
862 and Genevieve Parton for helpful discussions regarding sediment resuspension. This work
863 was funded by the National Sciences and Engineering Council (NSERC) Climate Change
864 and Atmospheric Research Grant: GEOTRACES (RGPC 433848-12) and VITALS (RG-
865 PCC 433898), an NSERC Discovery Grant (RGPIN-2016-03865) to SEA, and by the Uni-
866 versity of British Columbia through a four year fellowship to BR. Computing resources
867 were provided by Compute Canada (RRG 2648 RAC 2019, RRG 2969 RAC 2020, RRG

868 1541 RAC 2021). The model configuration, code, results, and analysis code are archived
 869 on FRDR at <https://doi.org/10.20383/102.0388>. Analysis code is also available via Github
 870 at <https://github.com/brogalla/Mn-sea-ice-paper>.

871 **References**

- 872 Aguilar-Islas, A. M., Rember, R. D., Mordy, C. W., & Wu, J. (2008). Sea ice-
 873 derived dissolved iron and its potential influence on the spring algal bloom in
 874 the Bering Sea. *Geophys. Res. Lett.*, *35*(24). doi: 10.1029/2008GL035736
- 875 Bacon, S., Marshall, A., Holliday, N. P., Aksenov, Y., & Dye, S. R. (2014). Seasonal
 876 variability of the East Greenland Coastal Current. *J. Geophys. Res.-Ocean.*,
 877 *119*(6), 3967–3987. doi: 10.1002/2013JC009279
- 878 Balzer, W. (1982). On the distribution of iron and manganese at the sediment/water
 879 interface: Thermodynamic versus kinetic control. *Geochem. Cosmochim. Acta*,
 880 *46*(7), 1153–1161. doi: 10.1016/0016-7037(82)90001-1
- 881 Bamber, J., Van Den Broeke, M., Ettema, J., Lenaerts, J., & Rignot, E. (2012).
 882 Recent large increases in freshwater fluxes from Greenland into the North
 883 Atlantic. *Geophys. Res. Lett.*, *39*(19).
- 884 Bouillon, S., Morales Maqueda, M. A., Legat, V., & Fichet, T. (2009). An elastic-
 885 viscous-plastic sea ice model formulated on Arakawa B and C grids. *Ocean*
 886 *Model.*, *27*(3-4), 174–184. doi: 10.1016/j.ocemod.2009.01.004
- 887 Brand, L. E., Sunda, W. G., & Guillard, R. R. L. (1983). Limitation of marine
 888 phytoplankton reproductive rates by zinc, manganese, and iron. *Limnol.*
 889 *Oceanogr.*, *28*(6), 1182–1198. doi: 10.4319/lo.1983.28.6.1182
- 890 Brown, K. A., Williams, W. J., Carmack, E. C., Fiske, G., François, R., McLen-
 891 nan, D., & Peucker-Ehrenbrink, B. (2020). Geochemistry of small Canadian
 892 Arctic rivers with diverse geological and hydrological settings. *J. Geophys.*
 893 *Res.-Biogeosciences*, *125*(1). doi: 10.1029/2019JG005414
- 894 Bruland, K. W., Donat, J. R., & Hutchins, D. A. (1991). Interactive influences
 895 of bioactive trace metals on biological production in oceanic waters. *Limnol.*
 896 *Oceanogr.*, *36*(8), 1555–1577. doi: 10.4319/lo.1991.36.8.1555
- 897 Bruland, K. W., Orians, K. J., & Cowen, J. P. (1994). Reactive trace metals in
 898 the stratified central North Pacific. *Geochem. Cosmochim. Acta*, *58*(15), 3171–
 899 3182. doi: 10.1016/0016-7037(94)90044-2

- 900 Campbell, J. A., & Yeats, P. A. (1982). The distribution of manganese, iron, nickel,
 901 copper and cadmium in the waters of Baffin Bay and the Canadian Arctic
 902 Archipelago. *Oceanol. Acta*, 5(2), 161–168.
- 903 Carrère, L., & Lyard, F. (2003). Modeling the barotropic response of the global
 904 ocean to atmospheric wind and pressure forcing-comparisons with observa-
 905 tions. *Geophys. Res. Lett.*, 30(6).
- 906 Charette, M. A., Kipp, L. E., Jensen, L. T., Dabrowski, J. S., Whitmore, L. M.,
 907 Fitzsimmons, J. N., ... others (2020). The Transpolar Drift as a source
 908 of riverine and shelf-derived trace elements to the central Arctic Ocean. *J.*
 909 *Geophys. Res.-Ocean.*, 125(5). doi: 10.1029/2019jc015920
- 910 Charette, M. A., Lam, P. J., Lohan, M. C., Kwon, E. Y., Hatje, V., Jeandel, C., ...
 911 Garcia-Orellana, J. (2016). Coastal ocean and shelf-sea biogeochemical cycling
 912 of trace elements and isotopes: lessons learned from GEOTRACES. *Philos.*
 913 *Trans. Roy. Soc. A*, 374(2081), 20160076. doi: 10.1098/rsta.2016.0076
- 914 Chelton, D. B., de Szoeke, R. A., Schlax, M. G., El Naggar, K., & Siwertz,
 915 N. (1998). Geographical variability of the first baroclinic Rossby radi-
 916 us of deformation. *J. Phys. Oceanogr.*, 28(3), 433–460. doi: 10.1175/
 917 1520-0485(1998)028%3C0433:GVOTFB%3E2.0.CO;2
- 918 Cid, A. P., Nakatsuka, S., & Sohrin, Y. (2012). Stoichiometry among bioactive trace
 919 metals in the Chukchi and Beaufort Seas. *J. Oceanogr.*, 68(6), 985–1001. doi:
 920 10.1007/s10872-012-0150-8
- 921 Colombo, M., Brown, K. A., De Vera, J., Bergquist, B. A., & Orians, K. J. (2019).
 922 Trace metal geochemistry of remote rivers in the Canadian Arctic Archipelago.
 923 *Chem. Geol.*, 525, 479–491. doi: 10.1016/j.chemgeo.2019.08.006
- 924 Colombo, M., Jackson, S. L., Cullen, J. T., & Orians, K. J. (2020). Dissolved
 925 iron and manganese in the Canadian Arctic Ocean: on the biogeochemical
 926 processes controlling their distributions. *Geochem. Cosmochim. Acta*, 277,
 927 150–174. doi: 10.1016/j.gca.2020.03.012
- 928 Dai, A., Qian, T., Trenberth, K. E., & Milliman, J. D. (2009). Changes in continen-
 929 tal freshwater discharge from 1948 to 2004. *J. Climate*, 22(10), 2773–2792. doi:
 930 10.1175/2008JCLI2592.1
- 931 Damm, E., Bauch, D., Krumpfen, T., Rabe, B., Korhonen, M., Vinogradova, E.,
 932 & Uhlig, C. (2018). The Transpolar Drift conveys methane from the

- 933 Siberian Shelf to the central Arctic Ocean. *Sci. Rep.*, 8(1), 1–10. doi:
934 10.1038/s41598-018-22801-z
- 935 Darby, D. A., Myers, W. B., Jakobsson, M., & Rigor, I. (2011). Modern dirty sea ice
936 characteristics and sources: The role of anchor ice. *J. Geophys. Res.-Ocean.*,
937 116(9). doi: 10.1029/2010JC006675
- 938 D'Asaro, E. A., & Morison, J. H. (1992). Internal waves and mixing in the Arctic
939 Ocean. *Deep Sea Res. Pt. I*, 39(2), S459–S484. doi: 10.1016/S0198-0149(06)
940 80016-6
- 941 Dethleff, D., & Kuhlmann, G. (2009). Entrainment of fine-grained surface deposits
942 into new ice in the southwestern Kara Sea, Siberian Arctic. *Cont. Shelf Res.*,
943 29(4), 691–701. doi: 10.1016/j.csr.2008.11.009
- 944 Dethleff, D., & Kuhlmann, G. (2010). Fram Strait sea-ice sediment provinces based
945 on silt and clay compositions identify Siberian Kara and Laptev seas as main
946 source regions. *Polar Sci.*, 29(3). doi: 10.3402/polar.v29i3.6070
- 947 Dethleff, D., Rachold, V., Tintelnot, M., & Antonow, M. (2000). Sea-ice transport
948 of riverine particles from the Laptev Sea to Fram Strait based on clay mineral
949 studies. *Intl. J. Earth Sci.*, 89(3), 496–502. doi: 10.1007/s005310000109
- 950 Drinkwater, K. F., & Harding, G. C. (2001). Effects of the Hudson Strait outflow on
951 the biology of the Labrador Shelf. *Can. J. Fish. Aquat. Sci.*, 58(1), 171–184.
952 doi: 10.1139/f00-210
- 953 Eicken, H., Gradinger, R., Gaylord, A., Mahoney, A., Rigor, I., & Melling, H.
954 (2005). Sediment transport by sea ice in the Chukchi and Beaufort Seas:
955 Increasing importance due to changing ice conditions? *Deep Sea Res. Pt. II*,
956 52, 3281–3302. doi: 10.1016/j.dsr2.2005.10.006
- 957 Eicken, H., Kolatschek, J., Freitag, J., Lindemann, F., Kassens, H., & Dmitrenko,
958 I. (2000). A key source area and constraints on entrainment for basin-scale
959 sediment transport by Arctic sea ice. *Geophys. Res. Lett.*, 27(13), 1919–1922.
960 doi: 10.1029/1999GL011132
- 961 Eicken, H., Reimnitz, E., Alexandrov, V., Martin, T., Kassens, H., & Viehoff,
962 T. (1997). Sea-ice processes in the Laptev Sea and their importance
963 for sediment export. *Cont. Shelf Res.*, 17(2), 205–233. doi: 10.1016/
964 S0278-4343(96)00024-6
- 965 Epstein, J.-L. (2018). *The impact of internal tide mixing parameterizations in*

- 966 *an eddy-permitting model of the Arctic Ocean* (Master's thesis, University of
967 British Columbia). doi: 10.14288/1.0365809
- 968 Evans, L. K., & Nishioka, J. (2018). Quantitative analysis of Fe, Mn and Cd from
969 sea ice and seawater in the Chukchi Sea, Arctic Ocean. *Polar Sci.*, *17*, 50–58.
970 doi: 10.1016/j.polar.2018.07.002
- 971 Fetterer, F., Knowles, K., Meier, W. N., Savoie, M., & Windnagel, A. K. (2017).
972 Updated daily: Sea ice index, version 3. Boulder, Colorado USA. *NSIDC:*
973 *National Snow and Ice Data Center*. doi: 10.7265/N5K072F8
- 974 Fichefet, T., & Maqueda, M. A. M. (1997). Sensitivity of a global sea ice model to
975 the treatment of ice thermodynamics and dynamics. *J. Geophys. Res.-Ocean.*,
976 *102*(C6), 12609–12646. doi: 10.1029/97JC00480
- 977 Fichot, C. G., Kaiser, K., Hooker, S. B., Amon, R. M., Babin, M., Bélanger, S., ...
978 Benner, R. (2013). Pan-Arctic distributions of continental runoff in the Arctic
979 Ocean. *Sci. Rep.*, *3*(1), 1–6. doi: 10.1038/srep01053
- 980 Fishwick, M. P., Ussher, S. J., Sedwick, P. N., Lohan, M. C., Worsfold, P. J., Buck,
981 K. N., & Church, T. M. (2018). Impact of surface ocean conditions and aerosol
982 provenance on the dissolution of aerosol manganese, cobalt, nickel and lead in
983 seawater. *Mar. Chem.*, *198*, 28–43. doi: 10.1016/J.MARCHEM.2017.11.003
- 984 Gent, P. R., Willebrand, J., McDougall, T. J., & McWilliams, J. C. (1995). Param-
985 eterizing eddy-induced tracer transport in ocean circulation models. *J. Phys.*
986 *Oceanogr.*, *25*(4), 463–474. doi: 10.1175/1520-0485(1995)025%3C0463:PEITTI%
987 3E2.0.CO;2
- 988 Granskog, M. A., Kaartokallio, H., & Shirasawa, K. (2003). Nutrient status of Baltic
989 Sea ice: Evidence for control by snow-ice formation, ice permeability, and ice
990 algae. *J. Geophys. Res.-Ocean.*, *108*(C8). doi: 10.1029/2002jc001386
- 991 Greene, C. H., & Pershing, A. J. (2007). Climate drives sea change. *Science*,
992 *315*(5815), 1084–1085. doi: 10.1126/science.1136495
- 993 Grivault, N., Hu, X., & Myers, P. G. (2018). Impact of the surface stress on the
994 volume and freshwater transport through the Canadian Arctic Archipelago
995 from a high-resolution numerical simulation. *J. Geophys. Res.-Ocean.*, *123*(12),
996 9038–9060. doi: 10.1029/2018JC013984
- 997 Guay, C. K. H., McLaughlin, F. A., & Yamamoto-Kawai, M. (2009). Differentiating
998 fluvial components of upper Canada Basin waters on the basis of measure-

- 999 ments of dissolved barium combined with other physical and chemical tracers.
1000 *J. Geophys. Res.-Ocean.*, *114*(C1). doi: 10.1029/2008JC005099
- 1001 Hawkings, J. R., Wadham, J. L., Tranter, M., Raiswell, R., Benning, L. G.,
1002 Statham, P. J., ... Telling, J. (2014). Ice sheets as a significant source of
1003 highly reactive nanoparticulate iron to the oceans. *Nat. Commun.*, *5*(1), 1–8.
1004 doi: 10.1038/ncomms4929
- 1005 Hayashida, H., Christian, J. R., Holdsworth, A. M., Hu, X., Monahan, A. H.,
1006 Mortenson, E., ... Steiner, N. S. (2019). CSIB v1 (Canadian Sea-ice Bio-
1007 geochemistry): a sea-ice biogeochemical model for the NEMO community
1008 ocean modelling framework. *Geosci. Model Dev.*, *12*(5), 1965–1990.
- 1009 Hölemann, J., Wegener, A., & Schirmacher, M. (1999). Dissolved and particulate
1010 major and trace elements in newly formed ice from the Laptev Sea (Trans-
1011 drift III, October 1995). In *Land-ocean systems in the Siberian Arctic* (pp.
1012 101–111). Springer Berlin Heidelberg. doi: 10.1007/978-3-642-60134-7_11
- 1013 Holland, M. M., Bailey, D. A., Briegleb, B. P., Light, B., & Hunke, E. (2012). Im-
1014 proved sea ice shortwave radiation physics in CCSM4: The impact of melt
1015 ponds and aerosols on Arctic sea ice. *J. Climate*, *25*(5), 1413–1430. doi:
1016 10.1175/JCLI-D-11-00078.1
- 1017 Hu, X., Myers, P. G., & Lu, Y. (2019). Pacific water pathway in the Arctic Ocean
1018 and Beaufort Gyre in two simulations with different horizontal resolutions. *J.*
1019 *Geophys. Res.-Ocean.*, *124*(8), 6414–6432.
- 1020 Hu, X., Sun, J., Chan, T. O., & Myers, P. G. (2018). Thermodynamic and dy-
1021 namic ice thickness contributions in the Canadian Arctic Archipelago in
1022 NEMO-LIM2 numerical simulations. *Cryosphere*, *12*, 1233–1247. doi:
1023 10.5194/tc-12-1233-2018
- 1024 Hughes, K. G., Klymak, J. M., Hu, X., & Myers, P. G. (2017). Water mass
1025 modification and mixing rates in a 1/12 simulation of the Canadian Arc-
1026 tic Archipelago. *J. Geophys. Res.-Ocean.*, *122*, 803–820. doi: 10.1002/
1027 2016JC012235
- 1028 Hughes, K. G., Klymak, J. M., Williams, W. J., & Melling, H. (2018). Tidally
1029 modulated internal hydraulic flow and energetics in the central Canadian
1030 Arctic Archipelago. *J. Geophys. Res.-Ocean.*, *123*(8), 5210–5229. doi:
1031 10.1029/2018JC013770

- 1032 Hunter, J. D. (2007). Matplotlib: A 2d graphics environment. *Comput. Sci. Eng.*,
1033 9(3), 90–95.
- 1034 Jakobsson, M. (2002). Hypsometry and volume of the Arctic Ocean and
1035 its constituent seas. *Geochem. Geophys. Geosystems*, 3(5), 1–18. doi:
1036 10.1029/2001GC000302
- 1037 Jensen, L. T., Morton, P., Twining, B. S., Heller, M. I., Hatta, M., Measures, C. I.,
1038 ... Fitzsimmons, J. N. (2020). A comparison of marine Fe and Mn cycling:
1039 U.S. GEOTRACES GN01 Western Arctic case study. *Geochem. Cosmochim.*
1040 *Acta*. doi: 10.1016/j.gca.2020.08.006
- 1041 Johnson, K. S., Coale, K. H., Berelson, W. M., & Michael Gordon, R. (1996). On
1042 the formation of the manganese maximum in the oxygen minimum. *Geochem.*
1043 *Cosmochim. Acta*, 60(8), 1291–1299. doi: 10.1016/0016-7037(96)00005-1
- 1044 Kanna, N., Lannuzel, D., van der Merwe, P., & Nishioka, J. (2020). Size fraction-
1045 ation and bioavailability of iron released from melting sea ice in a subpolar
1046 marginal sea. *Mar. Chem.*, 221, 103774. doi: 10.1016/j.marchem.2020.103774
- 1047 Kay, J. E., Deser, C., Phillips, A., Mai, A., Hannay, C., Strand, G., ... Vertenstein,
1048 M. (2015). The community earth system model (CESM) large ensemble
1049 project : A community resource for studying climate change in the presence
1050 of internal climate variability. *B. Am. Meteorol. Soc.*, 96(8), 1333–1349. doi:
1051 10.1175/BAMS-D-13-00255.1
- 1052 Kelly, S. J., Proshutinsky, A., Popova, E. K., Aksenov, Y. K., & Yool, A. (2019).
1053 On the origin of water masses in the Beaufort Gyre. *J. Geophys. Res.-Ocean.*,
1054 124(7), 4696–4709. doi: 10.1029/2019JC015022
- 1055 Kempema, E. W., Reimnitz, E., & Barnes, P. (1989). Sea ice sediment entrainment
1056 and rafting in the Arctic. *J. Sediment. Petrol.*, 59(2), 308–317.
- 1057 Kipp, L. E., Charette, M. A., Moore, W. S., Henderson, P. B., & Rigor, I. G.
1058 (2018). Increased fluxes of shelf-derived materials to the central Arctic Ocean.
1059 *Sci. Advances*, 4(1). doi: 10.1126/sciadv.aao1302
- 1060 Klinkhammer, G. P., & Bender, M. L. (1980). The distribution of manganese in
1061 the Pacific Ocean. *Earth Planet Sc. Lett.*, 46(3), 361–384. doi: 10.1016/0012-
1062 -821X(80)90051-5
- 1063 Kluyver, T., Ragan-Kelley, B., Pérez, F., Granger, B., Bussonnier, M., Frederic, J.,
1064 ... others (2016). *Jupyter notebooks – a publishing format for reproducible*

- 1065 *computational workflows* (F. Loizides & B. Schmidt, Eds.). IOS Press.
- 1066 Kodaira, T., Waseda, T., Nose, T., & Inoue, J. (2020). Record high Pacific Arctic
1067 seawater temperatures and delayed sea ice advance in response to episodic
1068 atmospheric blocking. *Sci. Rep.*, *10*(1), 1–12.
- 1069 Kondo, Y., Obata, H., Hioki, N., Ooki, A., Nishino, S., Kikuchi, T., & Kuma, K.
1070 (2016). Transport of trace metals (Mn, Fe, Ni, Zn and Cd) in the western
1071 Arctic Ocean (Chukchi Sea and Canada Basin) in late summer 2012. *Deep Sea*
1072 *Res. Pt. I*, *116*, 236–252. doi: 10.1016/J.DSR.2016.08.010
- 1073 Krachler, M., Zheng, J., Fisher, D., & Shotyk, W. (2005). Analytical procedures for
1074 improved trace element detection limits in polar ice from Arctic Canada using
1075 ICP-SMS. *Anal. Chim. Acta*, *530*(2), 291–298. doi: 10.1016/j.aca.2004.09.024
- 1076 Krishfield, R. A., Proshutinsky, A., Tateyama, K., Williams, W. J., Carmack, E. C.,
1077 McLaughlin, F. A., & Timmermans, M. L. (2014). Deterioration of perenni-
1078 al sea ice in the Beaufort Gyre from 2003 to 2012 and its impact on the
1079 oceanic freshwater cycle. *J. Geophys. Res.-Ocean.*, *119*(2), 1271–1305. doi:
1080 10.1002/2013JC008999
- 1081 Krumpfen, T., Belter, H. J., Boetius, A., Damm, E., Haas, C., Hendricks, S., ...
1082 Stein, R. (2019). Arctic warming interrupts the Transpolar Drift and affects
1083 long-range transport of sea ice and ice-rafted matter. *Sci. Rep.*, *9*(1), 1–9. doi:
1084 10.1038/s41598-019-41456-y
- 1085 Kwok, R., Spreen, G., & Pang, S. (2013). Arctic sea ice circulation and drift speed:
1086 Decadal trends and ocean currents. *J. Geophys. Res.-Ocean.*, *118*(5), 2408–
1087 2425. doi: 10.1002/jgrc.20191
- 1088 Landing, W. M., & Bruland, K. W. (1987). The contrasting biogeochemistry of iron
1089 and manganese in the Pacific Ocean. *Geochem. Cosmochim. Acta*, *51*(1), 29–
1090 43. doi: 10.1016/0016-7037(87)90004-4
- 1091 Lange, M., & van Sebille, E. (2017). Parcels v0.9: Prototyping a Lagrangian ocean
1092 analysis framework for the petascale age. *Geosci. Model Dev.*, *10*(11), 4175–
1093 4186. doi: 10.5194/gmd-10-4175-2017
- 1094 Lannuzel, D., Schoemann, V., de Jong, J., & Tison, J.-L. (2007). Distribution and
1095 biogeochemical behaviour of iron in the East Antarctic sea ice. *Mar. Chem.*,
1096 *106*(1-2), 18–32. doi: 10.1016/J.MARCHEM.2006.06.010
- 1097 Lavelle, J. W., Cowen, J. P., & Massoth, G. J. (1992). A model for the deposition

- 1098 of hydrothermal manganese near ridge crests. *J. Geophys. Res.*, *97*(C5), 7413.
1099 doi: 10.1029/92JC00406
- 1100 Lévy, M., Estublier, A., & Madec, G. (2001). Choice of an advection scheme for bio-
1101 geochemical models. *Geophys. Res. Lett.*, *28*(19), 3725–3728. doi: 10.1029/
1102 2001GL012947
- 1103 Lewis, K. M., van Dijken, G. L., & Arrigo, K. R. (2020). Changes in phytoplankton
1104 concentration now drive increased Arctic Ocean primary production. *Science*,
1105 *369*(6500), 198–202. doi: 10.1126/science.aay8380
- 1106 Li, J. (2017). *Particulate trace metals and iron availability to phytoplankton in a*
1107 *changing Arctic Ocean* (Master’s thesis, University of British Columbia). doi:
1108 10.14288/1.0348666
- 1109 Liang, X., & Losch, M. (2018). On the effects of increased vertical mixing on the
1110 Arctic Ocean and sea ice. *J. Geophys. Res.-Ocean.*, *123*(12). doi: 10.1029/
1111 2018JC014303
- 1112 Macdonald, R. W., & Gobeil, C. (2012). Manganese sources and sinks in the Arc-
1113 tic Ocean with reference to periodic enrichments in basin sediments. *Aquat.*
1114 *Geochem.*, *18*(6), 565–591. doi: 10.1007/s10498-011-9149-9
- 1115 Madec, G. (2008). NEMO ocean engine. *Note du Pôle de modélisation, Insti-*
1116 *tut Pierre-Simon Laplace*, *27*(1288-1619). Retrieved from [https://www.nemo-](https://www.nemo-ocean.eu/wp-content/uploads/NEMO_book.pdf)
1117 [ocean.eu/wp-content/uploads/NEMO_book.pdf](https://www.nemo-ocean.eu/wp-content/uploads/NEMO_book.pdf)
- 1118 Martin, J. H., & Gordon, R. M. (1988). Northeast Pacific iron distributions in rela-
1119 tion to phytoplankton productivity. *Deep Sea Res. Pt. I*, *35*(2), 177–196. doi:
1120 10.1016/0198-0149(88)90035-0
- 1121 Masina, S., Storto, A., Ferry, N., Valdivieso, M., Haines, K., Balmaseda, M., ...
1122 Parent, L. (2017). An ensemble of eddy-permitting global ocean reanal-
1123 yses from the MyOcean project. *Clim. Dynam.*, *49*(3), 813–841. doi:
1124 10.1007/s00382-015-2728-5
- 1125 Measures, C. I. (1999). The role of entrained sediments in sea ice in the distribution
1126 of aluminium and iron in the surface waters of the Arctic Ocean. *Mar. Chem.*,
1127 *68*, 59–70. doi: 10.1016/S0304-4203(99)00065-1
- 1128 Middag, R., de Baar, H. J. W., Laan, P., Cai, P. H., & van Ooijen, J. C. (2011a).
1129 Dissolved manganese in the Atlantic sector of the Southern Ocean. *Deep Sea*
1130 *Res. Pt. II*, *58*(25-26), 2661–2677. doi: 10.1016/J.DSR2.2010.10.043

- 1131 Middag, R., de Baar, H. J. W., Laan, P., & Klunder, M. B. (2011b). Fluvial and hy-
 1132 drothermal input of manganese into the Arctic Ocean. *Geochem. Cosmochim.*
 1133 *Acta*, 75(9), 2393–2408. doi: 10.1016/J.GCA.2011.02.011
- 1134 Newton, R., Pfirman, S., Tremblay, B., & DeRepentigny, P. (2017). Increasing
 1135 transnational sea-ice exchange in a changing Arctic Ocean. *Earths Future*,
 1136 5(6), 633–647. doi: 10.1002/2016EF000500
- 1137 Nürnberg, D., Wollenburg, I., Dethleff, D., Eicken, H., Kassens, H., Letzig, T.,
 1138 & Reimnitz, E. (1994). Sediments in Arctic sea ice: Implications for
 1139 entrainment, transport and release. *Mar. Geol.*, 119, 185–214. doi:
 1140 10.1016/0025-3227(94)90181-3
- 1141 O’Brien, M. C., Macdonald, R. W., Melling, H., & Iseki, K. (2006). Particle
 1142 fluxes and geochemistry on the Canadian Beaufort Shelf: Implications for
 1143 sediment transport and deposition. *Cont. Shelf Res.*, 26(1), 41–81. doi:
 1144 10.1016/J.CSR.2005.09.007
- 1145 Oliphant, T. E. (2006). *A guide to numpy* (Vol. 1). Trelgol Publishing USA.
- 1146 Pedregosa, F., Varoquaux, G., Gramfort, A., Michel, V., Thirion, B., Grisel, O., ...
 1147 Duchesnay, E. (2011). Scikit-learn: Machine learning in Python. *J. Mach.*, 12,
 1148 2825–2830.
- 1149 Peeken, I., Primpke, S., Beyer, B., Gütermann, J., Katlein, C., Krumpen, T.,
 1150 ... Gerdtts, G. (2018). Arctic sea ice is an important temporal sink and
 1151 means of transport for microplastic. *Nat. Commun.*, 9(1), 1–12. doi:
 1152 10.1038/s41467-018-03825-5
- 1153 Pfirman, S. L., Eicken, H., Bauch, D., & Weeks, W. (1995). The potential transport
 1154 of pollutants by Arctic sea ice. *Sci. Total Environ.*, 159(2-3), 129–146. doi: 10
 1155 .1016/0048-9697(95)04174-Y
- 1156 Proshutinsky, A., Krishfield, R., Toole, J. M., Timmermans, M. L., Williams, W.,
 1157 Zimmermann, S., ... Zhao, J. (2019). Analysis of the Beaufort Gyre freshwa-
 1158 ter content in 2003–2018. *J. Geophys. Res.-Ocean.*, 124(12), 9658–9689. doi:
 1159 10.1029/2019JC015281
- 1160 Reimnitz, E., McCormick, M., Mcdougall, K., & Brouwers, E. (1993). Sediment ex-
 1161 port by ice rafting from a coastal polynya. *Arct. Antarct. Alp. Res.*, 25(2), 83–
 1162 98. doi: 10.1080/00040851.1993.12002988
- 1163 Rijkenberg, M. J. A., Slagter, H. A., van der Loeff, M., van Ooijen, J., & Gerringa,

- 1164 L. J. A. (2018). Dissolved Fe in the deep and upper Arctic Ocean with a
1165 focus on Fe limitation in the Nansen Basin. *Front. Mar. Sci.*, *5*, 88. doi:
1166 10.3389/fmars.2018.00088
- 1167 Smith, G. C., Roy, F., Mann, P., Dupont, F., Brasnett, B., Lemieux, J.-F., ...
1168 Bélair, S. (2014). A new atmospheric dataset for forcing ice-ocean models:
1169 Evaluation of reforecasts using the Canadian global deterministic prediction
1170 system. *Q. J. R. Meteorol. Soc.*, *140*(680), 881–894. doi: 10.1002/qj.2194
- 1171 Spreen, G., Kwok, R., & Menemenlis, D. (2011). Trends in Arctic sea ice drift and
1172 role of wind forcing: 1992-2009. *Geophys. Res. Lett.*, *38*(19). doi: 10.1029/
1173 2011GL048970
- 1174 Stierle, A. P., & Eicken, H. (2002). Sediment inclusions in Alaskan coastal
1175 sea ice: Spatial distribution, interannual variability, and entrainment re-
1176 quirements. *Arct. Antarct. Alp. Res.*, *34*(4), 465–476. doi: 10.1080/
1177 15230430.2002.12003518
- 1178 Stroeve, J. C., & Notz, D. (2018). Changing state of Arctic sea ice across all sea-
1179 sons. *Environ. Res. Lett.*, *13*(10). doi: 10.1088/1748-9326/aade56
- 1180 Stroeve, J. C., Serreze, M. C., Holland, M. M., Kay, J. E., Malanik, J., & Barrett,
1181 A. P. (2012). The Arctic’s rapidly shrinking sea ice cover: A research synthe-
1182 sis. *Clim. Change*, *110*(3-4), 1005–1027. doi: 10.1007/s10584-011-0101-1
- 1183 Sunda, W. G., & Huntsman, S. A. (1994). Photoreduction of manganese oxides in
1184 seawater. *Mar. Chem.*, *46*(1-2), 133–152. doi: 10.1016/0304-4203(94)90051-5
- 1185 Tagliabue, A., Bowie, A. R., Boyd, P. W., Buck, K. N., Johnson, K. S., & Saito,
1186 M. A. (2017). The integral role of iron in ocean biogeochemistry. *Nature*,
1187 *543*(7643), 51–59. doi: 10.1038/nature21058
- 1188 The Pandas development team. (2020). Pandas-dev/pandas: Pandas. *Zenodo*, *21*, 1–
1189 9.
- 1190 Thyng, K. M., Greene, C. A., Hetland, R. D., Zimmerle, H. M., & DiMarco, S. F.
1191 (2016). True colors of oceanography: Guidelines for effective and accurate
1192 colormap selection. *Oceanogr.*, *29*(3), 9–13.
- 1193 Tilmes, S., Lamarque, J. F., Emmons, L. K., Kinnison, D. E., Marsh, D., Garcia,
1194 R. R., ... Blake, N. (2016). Representation of the Community Earth System
1195 Model (CESM1) CAM4-chem within the Chemistry-Climate Model Initiative
1196 (CCMI). *Geosci. Model Dev.*, *9*(5), 1853–1890. doi: 10.5194/gmd-9-1853-2016

- 1197 Tovar-Sánchez, A., Duarte, C. M., Alonso, J. C., Lacorte, S., Tauler, R., & Galbán-
1198 Malagón, C. (2010). Impacts of metals and nutrients released from melt-
1199 ing multiyear Arctic sea ice. *J. Geophys. Res.*, *115*(C7), C07003. doi:
1200 10.1029/2009JC005685
- 1201 Tucker, W. B., Gow, A. J., Meese, D. A., Bosworth, H. W., & Reimnitz, E. (1999).
1202 Physical characteristics of summer sea ice across the Arctic Ocean. *J. Geophys.*
1203 *Res.-Ocean.*, *104*(C1), 1489–1504. doi: 10.1029/98jc02607
- 1204 Van Hulten, M., Middag, R., Dutay, J.-C., De Baar, H., Roy-Barman, M., Gehlen,
1205 M., ... Sterl, A. (2017). Manganese in the west Atlantic Ocean in the context
1206 of the first global ocean circulation model of manganese. *Biogeosciences*, *14*,
1207 1123–1152. doi: 10.5194/bg-14-1123-2017
- 1208 Virtanen, P., Gommers, R., Oliphant, T. E., Haberland, M., Reddy, T., Courn-
1209 peau, D., ... others (2020). Scipy 1.0: fundamental algorithms for scientific
1210 computing in python. *Nature methods*, *17*(3), 261–272.
- 1211 Wang, Q., Myers, P. G., Hu, X., & Bush, A. B. G. (2012). Flow constraints on
1212 pathways through the Canadian Arctic Archipelago. *Atmos.-Ocean*, *50*(3),
1213 373–385.
- 1214 Wang, S., Bailey, D., Lindsay, K., Moore, J. K., & Holland, M. (2014). Impact of sea
1215 ice on the marine iron cycle and phytoplankton productivity. *Biogeosciences*,
1216 *11*(17), 4713–4731. doi: 10.5194/bg-11-4713-2014
- 1217 Waskom, M., & the Seaborn development team. (2020). *Seaborn*. Zenodo. doi: 10
1218 .5281/zenodo.592845
- 1219 Wedepohl, H. K. (1995). The composition of the continental crust. *Geochem. Cos-*
1220 *mochim. Acta*, *59*(7), 1217–1232. doi: 10.1016/0016-7037(95)00038-2
- 1221 Willmott, C. J. (1981). On the validation of models. *Phys. Geogr.*, *2*(2), 184–194.
- 1222 Yamamoto-Kawai, M., McLaughlin, F. A., Carmack, E. C., Nishino, S., Shimada,
1223 K., & Kurita, N. (2009). Surface freshening of the Canada Basin, 2003–2007:
1224 River runoff versus sea ice meltwater. *J. Geophys. Res.*, *114*(C1), C00A05. doi:
1225 10.1029/2008JC005000
- 1226 Yeats, P. A., & Westerlund, S. (1991). Trace metal distributions at an Arctic Ocean
1227 ice island. *Mar. Chem.*, *33*(3), 261–277. doi: 10.1016/0304-4203(91)90071-4
- 1228 Zalesak, S. T. (1979). Fully multidimensional flux-corrected transport algorithms for
1229 fluids. *J. Comput. Phys.*, *31*(3), 335–362. doi: 10.1016/0021-9991(79)90051-2