



**Abstract**

Biogeochemical cycles in the Arctic Ocean are sensitive to the transport of materials from continental shelves into central basins by sea ice. However, it is difficult to assess the net effect of this supply mechanism due to the spatial heterogeneity of sea ice content. Manganese (Mn) is a micronutrient and tracer which integrates source fluctuations in space and time. The Arctic Ocean surface Mn maximum is attributed to freshwater, but studies struggle to distinguish sea ice and river contributions. Informed by observations from 2009 IPY and 2015 Canadian GEOTRACES cruises, we developed a three-dimensional dissolved Mn model within a 1/12 degree coupled ocean-ice model centered on the Canada Basin and the Canadian Arctic Archipelago (CAA). Simulations from 2002-2019 indicate that annually, 93% of Mn contributed to the Canada Basin upper ocean is released by sea ice, while rivers, although locally significant, contribute only 2%. Downstream, sea ice provides 34% of Mn transported from Parry Channel into Baffin Bay. While rivers are often considered the main source of Mn, our findings suggest that in the Canada Basin they are less important than sea ice. However, within the shelf-dominated CAA, both rivers and sediment resuspension are important. Climate induced disruption of the transpolar drift may reduce the Canada Basin Mn maximum and supply downstream. Other nutrient elements found in sediments, such as Fe, may be similarly affected. These results highlight the vulnerability of the biogeochemical supply mechanisms in the Arctic Ocean and the subpolar seas to climatic changes.

**Plain Language Summary**

Autumn storms on the Siberian side of the Arctic Ocean churn up sediment that freezes into sea ice. The prevailing ocean currents and winds push this sea ice across the Arctic Ocean towards the Canada Basin, where it melts and releases the sediment into the ocean. Sediment contains manganese (Mn) and other nutrient elements that help support plankton and life. Using our Mn ocean model, 93% of Mn in the Canada Basin comes from “dirty” sea ice from 2002 to 2019, while rivers supply 2%. As a result of climate change, less dirty sea ice may make it across the Arctic Ocean, which could reduce this supply system of Mn and other similar nutrients. This change also has potential impacts downstream: water from the Canada Basin travels through the shallow Canadian Arctic Archipelago (CAA) into Baffin Bay and eventually the North Atlantic. We found that about 34% of Mn transported along this route comes from “dirty” sea ice. In the CAA,

47 other sources contribute as well: tides churn up sediments from the ocean floor and many  
48 rivers flow into the channels. Our study highlights ways in which climate change may  
49 impact the nutrient supply systems in the Arctic Ocean.

## 50 **1 Introduction**

51 As the sea ice regime in the Arctic Ocean transitions from multi-year ice to pre-  
52 dominantly first-year ice with overall reductions in sea ice extent, thickness and altered  
53 drift patterns (Stroeve et al., 2012; Stroeve & Notz, 2018; Spreen et al., 2011; Kwok et  
54 al., 2013), biogeochemical cycles and primary productivity are impacted through changes  
55 to the sea ice supply mechanism. The Arctic Ocean continental shelves connect land and  
56 ocean through the transfer of river runoff and sea ice from near-shore regions to the cen-  
57 tral basins (Charette et al., 2016). Reductions in sea ice export from the shelves weak-  
58 ens the long range transport of ice-rafted matter (Krumpen et al., 2019), including sed-  
59 iments (Dethleff et al., 2000; Darby et al., 2011), nutrients and trace metals (Tovar-Sánchez  
60 et al., 2010; Measures, 1999), pollutants (Pfirman et al., 1995; Peeken et al., 2018) and  
61 climate-relevant gases (Damm et al., 2018), to the surface ocean in regions far away from  
62 boundary sources. It is challenging to quantify the contribution of materials supplied by  
63 sea ice with observations alone due to the high spatial and temporal variability in the  
64 amount of sediment in sea ice and because it is difficult to distinguish it from additional  
65 contributions to the surface ocean such as river runoff. However, it is clear that changes  
66 to the physical processes in the Arctic Ocean will have impacts on the biogeochemical  
67 cycles and primary productivity of the basins themselves, as well as downstream in sub-  
68 polar seas (Drinkwater & Harding, 2001; Greene & Pershing, 2007).

69 Continental shelves cover half of the area of the Arctic Ocean (Jakobsson, 2002)  
70 and their shallow depths facilitate the incorporation of suspended matter into sea ice as  
71 it forms (Kempema et al., 1989). The narrow and deeper North American shelves are  
72 not as important for basin-wide sea ice sediment transport as the wide Siberian shelves  
73 (Eicken et al., 2005). In the Siberian shelf regions, fast ice builds up near shore in the  
74 fall, coinciding with storm-related resuspension events, forming sediment-rich sea ice (Nürn-  
75 berg et al., 1994). The transpolar drift transports this sea ice towards the North Pole  
76 and the anticyclonic Beaufort Gyre redirects a portion into the Canada Basin. This pas-  
77 sage takes several years, during which the ice undergoes cycles of melting, freezing and  
78 deformation. The materials released by melt alter the geochemical signature of the un-

79 derlying water (Pfirman et al., 1995). Several studies indicate an increase in sea ice ex-  
80 change through increased drift speeds (Spren et al., 2011; Kwok et al., 2013; Newton  
81 et al., 2017; Kipp et al., 2018), however a recent study indicates a disruption of the long  
82 range transport of sediments by sea ice due to the melt of first-year ice before it is in-  
83 corporated into the transpolar drift (Krumpen et al., 2019) with implications for the sur-  
84 face ocean of the endmembers of this transport pathway: Fram Strait, and indirectly the  
85 Canada Basin, the Canadian Arctic Archipelago, and the subpolar North Atlantic. In  
86 order to establish the importance of sediment from sea ice for biogeochemical cycles in  
87 the indirectly impacted regions of the Canada Basin and the Canadian Arctic Archipelago,  
88 we developed a model of dissolved manganese (Mn).

89 Mn is a reactive trace element and an important micronutrient which shares many  
90 sources with iron (Fe) in the Arctic Ocean (Brand et al., 1983; Bruland et al., 1991; Jensen  
91 et al., 2020). Mn has a scavenged-type profile with high concentrations near sources and  
92 low background concentrations. This contrast makes it a convenient source tracer. Over  
93 the Arctic Ocean shelves, sediment resuspension contributes Mn to the lower water col-  
94 umn (Evans & Nishioka, 2018; Colombo et al., 2020). However, the majority of exter-  
95 nal sources supply Mn to the ocean surface, contributing to the surface Mn maximum.  
96 In the Arctic Ocean, this surface maximum is attributed to freshwater sources (Campbell  
97 & Yeats, 1982; Yeats & Westerlund, 1991; Middag et al., 2011b; Cid et al., 2012; Kondo  
98 et al., 2016; Colombo et al., 2020). Observational studies have identified the origin of  
99 this freshwater as river discharge (Campbell & Yeats, 1982; Yeats & Westerlund, 1991;  
100 Evans & Nishioka, 2018), sea ice meltwater (Measures, 1999; S. Wang et al., 2014) (for  
101 Fe), or a combination of both (Middag et al., 2011b; Cid et al., 2012; Kondo et al., 2016;  
102 Colombo et al., 2020). Observations of riverine Mn indicate significantly higher concen-  
103 trations than in the ocean (Colombo et al., 2019). Similarly, trace metals and nutrients  
104 in sea ice occur in concentrations in excess of those in the ocean (Campbell & Yeats, 1982;  
105 Hölemann et al., 1999; Granskog et al., 2003; Krachler et al., 2005; Aguilar-Islas et al.,  
106 2008; Tovar-Sánchez et al., 2010; Kondo et al., 2016; Evans & Nishioka, 2018). The im-  
107 portance of these components depends in part on the distance and pathway of input into  
108 the ocean (Fichot et al., 2013). So, while the Canada Basin is relatively distant from land,  
109 the narrow and shallow systems of channels that make up the CAA are in close contact  
110 with the land-ocean interface and may be more directly impacted by boundary processes  
111 such as river discharge and sedimentary inputs (Colombo et al., 2021).

112 In order to distinguish the individual importance of external Mn sources within the  
113 Canada Basin and the CAA, model studies are needed. Past studies have used tracers  
114 such as terrestrial dissolved organic matter (Fichot et al., 2013) and the oxygen isotope  
115 ratio (Yamamoto-Kawai et al., 2009) to distinguish the contributors to freshwater in the  
116 Canada Basin. Mn is an interesting complementary tracer because of its role as a nu-  
117 trient and because it integrates processes that fluctuate on a short time scale. As a re-  
118 sult, Mn helps address one of the main limitations of the study of sediment entrainment  
119 and export events by sea ice: that they are episodic and localized in nature (Eicken et  
120 al., 2005). Similarly, while sediment resuspension occurs intermittently, Mn integrates  
121 the effect of this component on the lower water column. After establishing the contri-  
122 butions of the Mn sources, we use Mn as a tool to study the general role of sea ice trans-  
123 port for biogeochemical cycles.

124 In this paper, we present a model of Mn in the Canadian Arctic Archipelago and  
125 the Canada Basin, informed by in situ observations collected during the 2009 IPY GEO-  
126 TRACES cruise (Sim, 2018) and the 2015 Canadian GEOTRACES cruises (Colombo  
127 et al., 2020). Our work builds on the comprehensive first global model of Mn in the ocean  
128 (Van Hulten et al., 2017) and previous smaller scale models of Mn in the North Pacific  
129 Ocean (Johnson et al., 1996) and near hydrothermal vents (Lavelle et al., 1992). We in-  
130 corporate new parameterizations for sediment resuspension, release of shelf sediments  
131 in sea ice, and fluvial contributions, to capture the drivers of Mn distributions in the Cana-  
132 dian Arctic. With this model, we show that the long range transport of sediments by  
133 sea ice from the Siberian shelves drives the surface Mn maximum in the Canada Basin  
134 while riverine contributions, although locally significant, are not as important as gen-  
135 erally identified. Using these results, we discuss implications of future sea ice melt on  
136 Mn and Fe nutrient budgets in the Canada Basin and downstream in the Canadian Arc-  
137 tic Archipelago and Baffin Bay.

## 138 **2 Methods**

### 139 **2.1 Coupled Ocean-Ice Model**

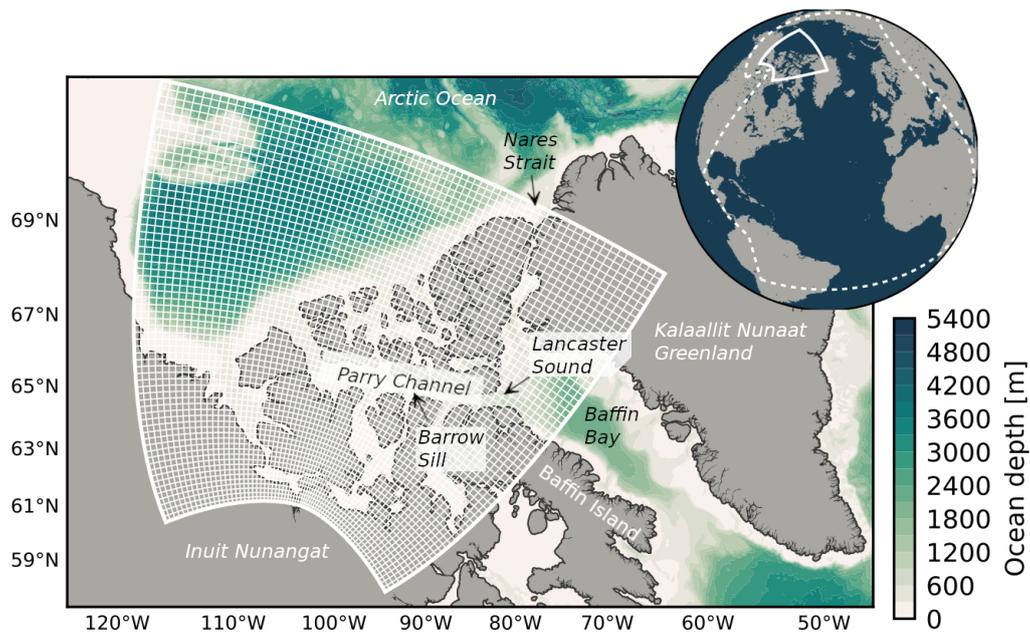
140 For our simulations, we use ocean and ice dynamics calculated by the Arctic and  
141 Northern Hemispheric Atlantic (ANHA12) configuration (Hu et al., 2018) of the Nucleus  
142 for European Modelling of the Ocean (NEMO) version 3.4 (Madec, 2008). The ANHA12

143 configuration has a nominal horizontal resolution of  $1/12^\circ$  which resolves freshwater fluxes  
144 associated with coastal currents in the CAA, as well as eddies (Bacon et al., 2014; Chel-  
145 ton et al., 1998). The position of the grid’s artificial pole in Northern Canada increases  
146 the resolution in the CAA to about 2-3 km (Fig. 1). In the vertical, there are 50 depth  
147 levels ranging from 1 m thickness at the surface to 454 m near the bottom. The bottom  
148 bathymetry is represented using partial steps.

149 The ANHA12 domain has two open boundaries: one in Bering Strait and the other  
150 at  $20^\circ\text{S}$  in the Atlantic Ocean. These boundaries are forced with Global Ocean Reanal-  
151 yses and Simulations data (Masina et al., 2017). The ocean surface is forced with hourly  
152 atmospheric data from the Canadian Meteorological Centre’s global deterministic pre-  
153 diction system (Smith et al., 2014) and the rivers are forced with monthly runoff clima-  
154 tology with enhanced Greenland melt runoff (Dai et al., 2009; Bamber et al., 2012). The  
155 river forcing from 2010 is repeated for the following years (Hu et al., 2019).

156 The sea ice in ANHA12 is represented using the dynamic and thermodynamic Louvain-  
157 la-Neuve (LIM2) sea ice model with an elastic-viscous-plastic ice rheology (Fichefet &  
158 Maqueda, 1997; Bouillon et al., 2009). An evaluation of LIM2 in the ANHA12 config-  
159 uration is provided by Hu et al. (2018). The general spatial distribution of ice thickness  
160 within the Archipelago is captured well. In the model, the northern CAA has very thick  
161 sea ice ( $> 4$  m), the central parts have intermediate thickness ice (2.5-3 m), and there  
162 is thin ( $< 2$  m) sea ice on the east side of the CAA and in southern channels. The ANHA12  
163 simulations are limited by the lack of a land-fast ice parameterization, resulting in ice  
164 velocities that are higher than observed in Parry Channel, impacting the winter trans-  
165 port (Grivault et al., 2018). In addition, tides are not included and as a result, the polynyas  
166 which form due to tidally enhanced mixing are not well reproduced (Hughes et al., 2018).

167 The advection and diffusion of tracers are calculated within NEMO by the TOP  
168 engine (Gent et al., 1995; Lévy et al., 2001). Tracer advection is calculated with the To-  
169 tal Variance Dissipation (TVD) scheme (Zalesak, 1979) and we use the Flow Relaxation  
170 Scheme (FRS) for the tracer boundary conditions. The vertical diffusion of tracers is cal-  
171 culated from the Turbulent Kinetic Energy closure scheme within ANHA12 and the hor-  
172 izontal eddy diffusivity parameter is set to  $50.0 \text{ m}^2 \text{ s}^{-1}$ .



**Figure 1.** The Mn model domain is centered on the Canadian Arctic Archipelago with highest horizontal resolution in the south (about 3 km). The nominal horizontal resolution of the grid is  $1/12^\circ$ ; the white lines depict one in every ten grid points. The solid white line in the inset globe shows the Mn model domain extent, while the dashed white line delineates the domain of the Arctic and Northern Hemispheric Atlantic configuration (Hu et al., 2018) of the ocean-ice model.

## 173 2.2 Model of Mn in the Canadian Arctic

174 The Mn model runs offline in NEMO version 3.6 using five day averaged dynam-  
 175 ics fields from the ANHA12 reference run from January 2002 to December 2019 (Hu et  
 176 al., 2018). The Mn model consists of two main sets of computations: the advection and  
 177 diffusion of tracers calculated by the NEMO-TOP engine (Gent et al., 1995; Lévy et al.,  
 178 2001), and the source and sink contributions. The source and sink parameterizations were  
 179 developed guided by observations from the 2015 Canadian GEOTRACES cruises (Colombo  
 180 et al., 2020) and inspired by the first global model of Mn (Van Hulst et al., 2017). In  
 181 order to reduce the computational cost, we calculate the model on a sub-domain of ANHA12,  
 182 centered on the CAA (Fig. 1). Note that since we run offline, the physics originates from  
 183 the full domain.

184 The known sources and sinks of Mn in the ocean are: rivers, hydrothermal vents,  
 185 sediment diffusion, sediment resuspension, reversible scavenging, sinking, uptake and rem-  
 186 ineralization, atmospheric dust deposition, and flux from ice (Middag et al., 2011b; Balzer,  
 187 1982; Klinkhammer & Bender, 1980; Evans & Nishioka, 2018). From this list, we incor-  
 188 porate the processes that are important for dissolved Mn in the Arctic (summarized in  
 189 Fig. 2 and Eqn. 1 and 2). In order to model the reversible scavenging of Mn, we incor-  
 190 porate Mn oxides (oMn) in addition to dissolved Mn (similar to Van Hulst et al. (2017);  
 191 Eqn. 2). We do not model particle-bound Mn (pMn), but rather incorporate the indi-  
 192 rect effect of pMn onto dMn through dissolution from the source components. We did  
 193 not incorporate hydrothermal vents as a source of Mn in the Arctic, since the influence  
 194 of the Gakkel Ridge is restricted to Nansen and Amundsen Basins due to scavenging nearby  
 195 the source (Lavelle et al., 1992; Middag et al., 2011b). We also do not include sediment  
 196 diffusion (reductive dissolution) because observations have indicated that these processes  
 197 are not significant for Mn in the CAA (Colombo et al., 2020). The Mn model equations  
 198 are:

$$\frac{\partial dMn}{\partial t} = S_{river} + S_{sediment} + S_{atm} + S_{ice} + S_{sed\ ice} + S_{bio} + R_{scav} + \text{advection} + \text{diffusion} \quad (1)$$

199

$$\frac{\partial oMn}{\partial t} = -R_{scav} - R_{sink} + \text{advection} + \text{diffusion} \quad (2)$$

200 which include the contribution of rivers,  $S_{river}$ , sediment resuspension (non-reductive  
 201 dissolution),  $S_{sediment}$ , atmospheric dust deposition,  $S_{atm}$ , dust flux from ice,  $S_{ice}$ , sed-  
 202 iment released by ice,  $S_{sed\ ice}$ , biological uptake and remineralization,  $S_{bio}$ , the reversible  
 203 scavenging terms,  $R_{scav}$ , and sinking,  $R_{sink}$ . The details of the parameterizations are  
 204 described in the following sections and the parameter values used for the runs are listed  
 205 in Table 1.

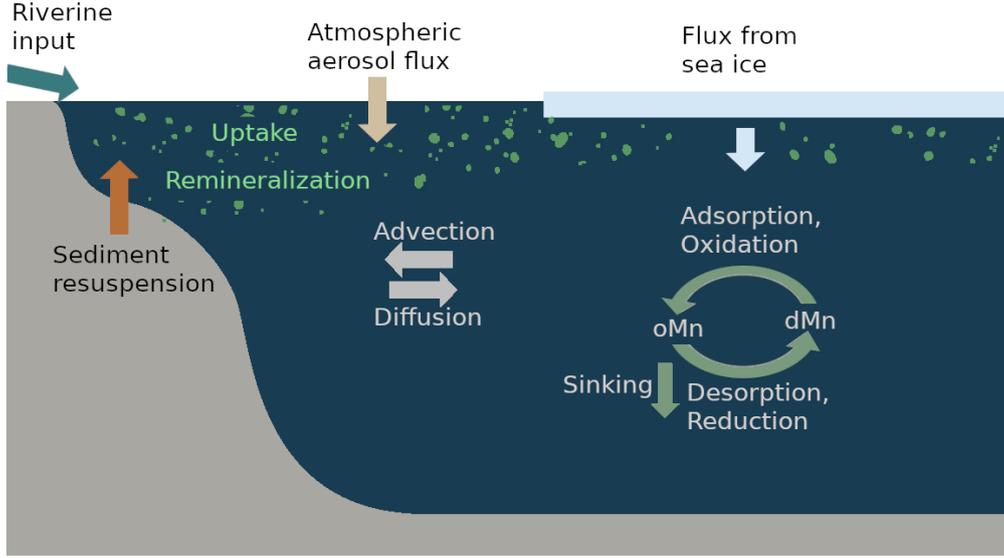
206 The model was initialized with output from the global Mn model (Van Hulst et  
 207 al., 2017) and concentrations are held constant at the sub-domain boundaries. At these  
 208 boundaries, the ratio of dissolved to oxidised Mn from the global model were not rep-  
 209 resentative (oxidised Mn was too low) and resulted in unusual scavenging behavior. To  
 210 address this, we took values for the dissolved and oxidised Mn concentrations in a band  
 211 15 grid cells towards the interior of the basin (where the model had established normal  
 212 scavenging behavior) from a test model run at the end of spin up and used those val-  
 213 ues for the boundary conditions.

**Table 1.** Constants and parameter values used in the Mn model runs.

Parameter	Description	Value	Source
$\alpha_0$	Solubility of Mn at 4°C	0.65	Fishwick et al. (2018)
$f_{Mn\ crust}$	Mn fraction in Earth's crust	527 ppm	Wedepohl (1995)
$f_{Mn\ sed}$	Mn fraction in marine sediment	270 ppm	Macdonald and Gobeil (2012)
$m$	Molar mass of Mn	54.938 g mol <sup>-1</sup>	—
$R_{Mn:N}$	Extended Redfield ratio Mn:N	1.6 : 23,000	Kuss and Kremling (1999)
$k_d$	Reduction and desorption rate	$4.7 \cdot 10^{-7} \text{ s}^{-1}$	Bruland et al. (1994)
	Photo-enhanced reduction rate	$2.7 \cdot 10^{-5} \text{ s}^{-1}$	Sunda and Huntsman (1994)
$k_p$	Oxidation and adsorption rate	$7.0 \cdot 10^{-7} \text{ s}^{-1}$	This study <sup>a</sup>
$s_{ox}$	Sinking rate	0.6 m day <sup>-1</sup>	Roy-Barman (2009) / This study
$C$	Tidal erosion tuning constant	$2.1 \cdot 10^{-6}$	This study
$\gamma$	Solubility tuning constant	0.065	This study
$R / SPM$	River characteristic content		This study <sup>b</sup>
	- Glacial	164 nM / 261 mg L <sup>-1</sup>	
	- Continental	30 nM / 12 mg L <sup>-1</sup>	
	- Other	2 nM / 4 mg L <sup>-1</sup>	

<sup>a</sup>Using data from Colombo et al. (2020); Li (2017).

<sup>b</sup>Using data from Colombo et al. (2019); Brown et al. (2020).



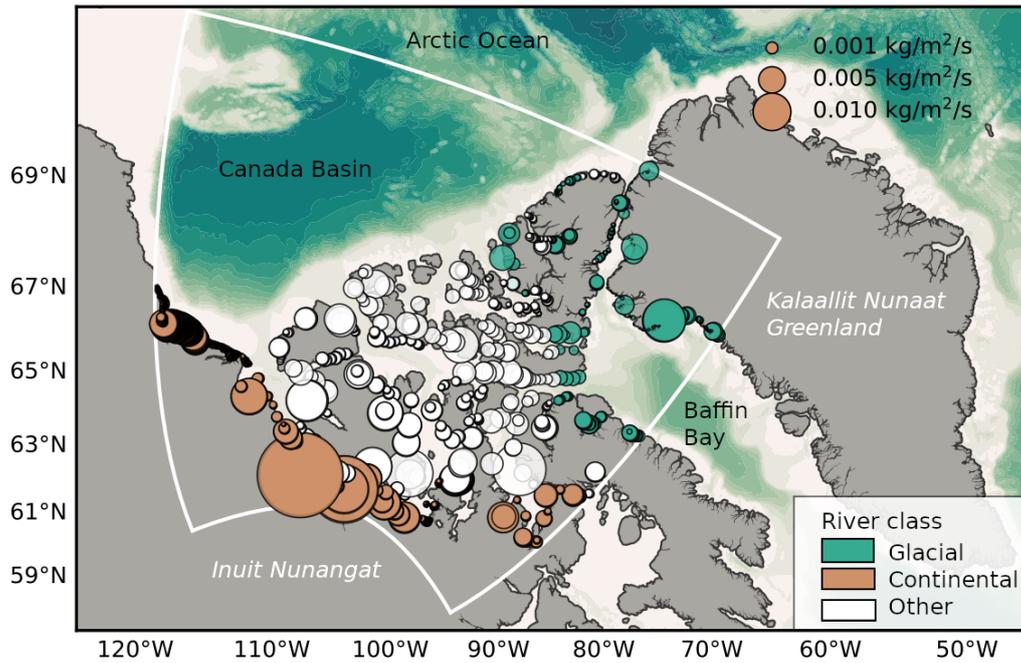
**Figure 2.** Summary of the processes that affect Mn concentrations in the Canadian Arctic Archipelago and the Canada Basin.

### 2.2.1 Riverine Source

River discharge contributes Mn to the shelf seas and into the Arctic Ocean (Middag et al., 2011a). Dissolved Mn is contributed directly in its dissolved form and indirectly through the dissolution from particle-bound Mn. The contribution of riverine Mn depends on the river discharge,  $Q$ , and the concentration in the rivers. These concentrations vary based on properties of the river’s catchment basin: glacial rivers are strongly enriched in dissolved Mn, continental rivers are somewhat enriched, and in all other rivers, Mn is not significantly enriched (Colombo et al., 2019). At each time step, the rivers contribute dissolved Mn following:

$$S_{river} = \frac{\overbrace{Q}^{\text{dissolved Mn}}}{\rho_0 \Delta z_{surface}} R_{class} + \beta \frac{\overbrace{Q}^{\text{particle origin dissolved Mn}}}{\rho_0 \Delta z_{surface}} \frac{SPM_{class} \cdot \alpha_0 \cdot f_{Mn, crust}}{m} \quad (3)$$

where  $\rho_0$  is the density of the river water,  $\Delta z_{surface}$  is the surface grid box thickness,  $\beta$  is a factor which ranges from 0-1 in our experiments (not tuned, but tested in the upper bound river experiment),  $f_{Mn, crust}$  is the crustal abundance of Mn,  $m$  is the molar mass of Mn, and  $\alpha_0$  is the solubility of Mn. We use an average value for the solubility (65%) measured in seawater at 4°C, since this lower temperature better reflects the



**Figure 3.** Model rivers were classified based on their drainage basin properties: glacial (green), continental (orange), or other (white). The points on this map are the locations of the river water input in the model and their sizes are proportional to the river discharge in September, 2015 (forcing is repeated from year 2010). Note that the river freshwater flux is remapped to prevent negative model salinities, hence some large rivers are represented as single point sources, while others such as the Mackenzie River consist of multiple point sources along the coastline (Hu et al., 2019).

228 CAA (Fishwick et al., 2018). This solubility falls within the range measured in samples  
 229 across the world (Fishwick et al., 2018). Each river is assigned a class with an associ-  
 230 ated characteristic trace metal concentration,  $R_{class}$ , and suspended particulate matter  
 231 content,  $SPM_{class}$ , based on catchment basin properties: glacial, continental, and other  
 232 (Fig. 3 and Table 1). The Mn concentrations and SPM content associated with the classes  
 233 are determined from rivers sampled in the CAA (Colombo et al., 2019; Brown et al., 2020).

### 234 *2.2.2 Atmospheric Aerosol Flux and Release from Sea Ice*

235 Atmospheric aerosols contribute Mn to the ocean through direct deposition to sur-  
 236 face waters,  $\Phi_{atm}$ , or through the deposition onto sea ice and the subsequent release dur-

237 ing melt,  $\Phi_{ice}$ . We parameterized these contributions as:

$$S_{atm\ or\ ice} = \frac{\alpha_0 \cdot f_{Mn\ crust}}{m \cdot \Delta z_{surface}} \cdot \Phi_{atm\ or\ ice} \quad (4)$$

238 The atmospheric and sea ice flux terms are derived from monthly Community Earth Sys-  
 239 tem Model (CESM) results. The combined monthly dry and wet atmospheric deposi-  
 240 tion fluxes originate from historical (1920-2005) and future (2006-2080) runs of the Com-  
 241 munity Atmosphere Model with Chemistry (CAM-Chem) downloaded from the Climate  
 242 Data Gateway (CESM1 CAM5 BGC Large Ensemble Atmosphere Post Processed Data;  
 243 Tilmes et al. (2016)). We estimate tracer fluxes from ice using the monthly Community  
 244 Ice Code ensemble results (CICE; Holland et al. (2012); Kay et al. (2015)). These en-  
 245 semble run sets have a horizontal atmospheric resolution of  $0.9 \times 2.5^\circ$  and ocean/ice res-  
 246 olution of  $1.6 \times 2.5^\circ$  which we linearly interpolated to the ANHA12 grid. We do not tune  
 247 any of the parameters in this process.

### 248 ***2.2.3 Sediment Resuspension over the Continental Shelf***

249 Dissolved Mn increases near the ocean floor in the Canadian Arctic as a result of  
 250 sediment resuspension (Colombo et al., 2020). Sediment resuspension occurs intermit-  
 251 tently, however, Mn integrates the resuspension events and thereby provides a cumula-  
 252 tive view of its prevalence. We incorporated sediment resuspension as a continuous pro-  
 253 cess:

$$S_{sediment} = \Phi_{erosion} \cdot \frac{\alpha \cdot f_{Mn\ sed}}{m \cdot \Delta z_{bottom}} \quad (5)$$

254 where  $f_{Mn\ sed}$  is the fraction of Mn in marine sediments. This fraction is likely to be lower  
 255 than measured in the continental crust, i.e. Wedepohl (1995), since it's undergone some  
 256 amount of chemical transformation. We used the Mn fraction estimated by Macdonald  
 257 and Gobeil (2012) from sediments in cores on the shelf and slopes surrounding the Canada  
 258 Basin. In Eqn. 5,  $\Phi_{erosion}$  is the "erosion ability" (see Fig. S1 for the forcing field). This  
 259 term incorporates the spatial differences in dynamics within the CAA. West of Barrow  
 260 Sill, the system has lower mixing rates (Hughes et al., 2018) and tidal speeds (Epstein,  
 261 2018), than the region east of Barrow Sill and around the central sills area. These dif-  
 262 ferences impact the sediment resuspension rates, apparent in the much stronger near-  
 263 bottom increases of observed dMn in the eastern CAA (Colombo et al., 2020). We es-  
 264 timate the ability of sediment to be eroded with the barotropic tidal speed,  $U_{tidal}$ , and

265 a tuning constant,  $C$ :

$$\Phi_{erosion} = C \cdot U_{tidal}^2 \quad (6)$$

266 The barotropic tidal speeds are from the MOG2D-G model (Carrère & Lyard, 2003) and  
 267 are significantly higher in the eastern CAA, compared to the western CAA (Epstein, 2018).  
 268 Locations where the tidal speeds are less than  $1 \text{ cm s}^{-1}$  are masked, since they are be-  
 269 low a critical threshold for motion for particles greater than 0.1 mm, i.e. sand. In ar-  
 270 eas where resuspension occurs frequently, the easily accessible Mn on particles has al-  
 271 ready been removed, resulting in a lower solubility. We reduce the solubility in Eqn. 5  
 272 at high tidal speeds according to:

$$\alpha = \alpha_0 \cdot \frac{\gamma(1 - e^{-U_{tidal}^2/\gamma})}{U_{tidal}^2} \quad (7)$$

273 where  $\gamma$  is a tuning parameter. At small tidal speeds, Eqn. 7 approaches  $\alpha_0$  while at tidal  
 274 speeds greater than  $0.1 \text{ m s}^{-1}$ , solubility decreases and the overall resuspension rate ap-  
 275 proaches a constant  $\alpha_0\gamma C$  (Fig. S2). The tuning parameters were estimated based on  
 276 model behaviour in several tuning runs (see Section 2.3).

#### 277 **2.2.4 Sediment Entrained in Sea Ice and Subsequent Melt**

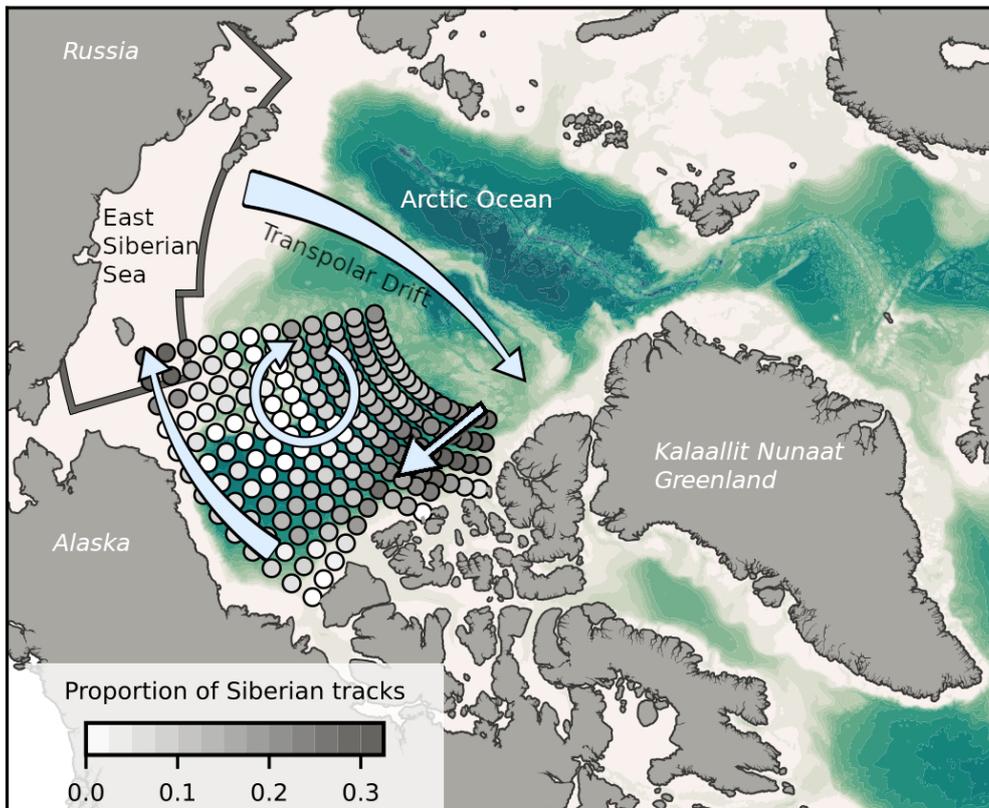
278 Sediment entrained in sea ice has been identified as an important source of reac-  
 279 tive trace metals such as aluminum and iron in the ocean, and thus may also be impor-  
 280 tant for Mn (Measures, 1999). In order to parameterize this contribution, we couple the  
 281 Mn contained in sediments in sea ice and the sea ice melt rate,  $I_{melt}$ :

$$S_{sed\ ice} = \frac{\alpha_0 \cdot f_{Mn\ sed}}{m \cdot \Delta z_{surface}} \cdot S_p \cdot I_{melt} \quad (8)$$

282 where  $S_p$  is the sediment content in sea ice at each grid point. The sediment content is  
 283 spatially variable, and depends on the amount of sediment that was incorporated dur-  
 284 ing ice formation on the shelves and on sea ice transport.

285 Through particle tracking experiments with Ocean Parcels (Lange & Van Sebille,  
 286 2017), we estimated the contribution of sea ice formed over the Siberian shelves during  
 287 the stormy fall months (September-December) to the ice in the Canada Basin (Fig. 4).  
 288 We released parcels every month over the course of a year and traced them backwards  
 289 for three years (the average sea ice age in the Canada Basin and the northwestern CAA  
 290 based on satellite information). Almost 40% of the sea ice tracks in the northwestern CAA  
 291 and Canada Basin region originated from the Siberian shelves via the transpolar drift

292 during the fall months, when strong sediment resuspension events coincide with sea ice  
 293 formation. The results of the particle tracking experiments were interpolated and smoothed  
 294 to create a forcing field which incorporates the spatial variation in sediment content in  
 295 sea ice (Fig. S3). In addition, we assumed a low background value of shelf sediments in  
 296 sea ice in the Archipelago. We multiply this forcing field by a tuned constant,  $0.85 \text{ kg m}^{-3}$ ,  
 297 which reflects the sediment content of the ice if it were fully formed over the Siberian  
 298 shelf, i.e. the proportion of Siberian tracks was one.



**Figure 4.** Sediment rich sea ice, produced over the Siberian shelves (East Siberian Sea; region definition outlined in brown) in the fall, is transported across the Arctic Ocean via the transpolar drift. From there it is found predominantly along the outer edges of the Beaufort Gyre; the largest contribution occurs in the northeastern CAA. Ice motion patterns are indicated with light blue arrows. For locations in the Canada Basin, the proportion of parcels traced back to the Siberian shelves (region defined with the brown outline) in the fall months are shown.

299 **2.2.5 Uptake and Remineralization**

300 Phytoplankton take up dissolved Mn in the euphotic zone and it is subsequently  
 301 remineralized below the euphotic zone. We can quantify this contribution by pairing the  
 302 addition and removal of Mn to the uptake and remineralization of nitrate:

$$S_{bio} = R_{Mn:N} \cdot \Delta N \quad (9)$$

303 where  $R_{Mn:N}$  is the extended Redfield ratio for Mn to nitrogen based on observations  
 304 in the North Atlantic (23,000 N : 1.6 Mn; Kuss and Kremling (1999)), and  $\Delta N$  is the  
 305 month-to-month change in nitrate concentration during the summer months (April-August)  
 306 from 2002-2015 derived from the Canadian Ocean Ecosystem Model (CanOE; Hayashida  
 307 et al. (2019)). We assume that the month-to-month change in nitrate is zero during sea-  
 308 sons with low biological activity to avoid confusing the replenishment of nitrate via mix-  
 309 ing with remineralization at the surface. We did not tune the uptake and remineraliza-  
 310 tion.

311 **2.2.6 Reversible Scavenging and Sinking**

312 Dissolved Mn oxidizes forming larger aggregates and adsorbs to particle surfaces.  
 313 dMn is regenerated by the reduction of oxidised Mn and desorption from particles. Since  
 314 we do not directly model particle-bound Mn, but rather incorporate its effect on dMn  
 315 through dissolution from the source components, we calculate the reversible scavenging  
 316 based on the dissolved and oxidised Mn concentrations (Van Hulst et al., 2017):

$$R_{scav} = -k_p \cdot [dMn] + k_d \cdot [oMn] \quad (10)$$

317 where  $k_p$  is the adsorption and oxidation rate, and  $k_d$  is the desorption and reduction  
 318 rate (see Text S1 for the full derivation). The  $R_{scav}$  term appears with opposing signs  
 319 in the dMn and oMn equations (Eqn. 1 and 2). We estimate the rate constant  $k_p$  from  
 320 observations of dissolved and particulate Mn in the Canadian Arctic (Li, 2017; Colombo  
 321 et al., 2020). As this estimate is based on field data, the rate intrinsically incorporates  
 322 the impact of abiotic and microbially enhanced oxidation. Assuming steady state, the  
 323 ratio of the scavenging rates is equal to the ratio of dissolved to particulate Mn concen-  
 324 trations. This assumption reduces the available observations to those far away from sources  
 325 and sinks, i.e. deep stations in Baffin Bay and the Canada Basin (Fig. S4). The ratio  
 326 of scavenging rates,  $k_p/k_d$ , is estimated as  $1.47 \pm 0.25$  and with a  $k_d$  of  $4.7 \cdot 10^{-7} \text{ s}^{-1}$

327 (Bruland et al., 1994),  $k_p$  is estimated as  $7.0 \cdot 10^{-7} \text{ s}^{-1}$  (Fig. S5). The reduction rate,  
 328  $k_d$ , increases from the base rate up to  $2.7 \cdot 10^{-5} \text{ s}^{-1}$  in the euphotic zone (photo-enhanced  
 329 reduction; Sunda and Huntsman (1994)), proportional to the solar flux that penetrates  
 330 into the ocean at the surface (from ANHA12). We estimate the euphotic zone depth as  
 331 70 m in the Canada Basin with a gradual transition to 50 m in the CAA based on es-  
 332 timates by Bhatia et al. (2021) and Laney et al. (2017) (see Fig. S6). The scavenging  
 333 rates in the model do not depend on the dissolved oxygen concentration since Arctic wa-  
 334 ters are generally well oxygenated.

335 The oxidised Mn aggregates sink,  $R_{sink}$ , and are removed through burial as in Van  
 336 Hulten et al. (2017):

$$R_{sink} = s_{ox} \frac{\partial[oMn]}{\partial z} \quad (11)$$

337 where  $s_{ox}$  is the sinking rate. The sinking rate was based on the estimate by Roy-Barman  
 338 (2009) of  $0.4 \text{ m d}^{-1}$  in the interior of the Arctic Ocean and then increased to  $0.6 \text{ m d}^{-1}$   
 339 based on an evaluation of modelled background oMn concentrations in the Canada Basin  
 340 far away from sources and sinks.

### 341 2.3 Tuning

342 Of the parameters in our model (Table 1), we tuned the oMn sinking rate, sedi-  
 343 ment resuspension rate, sediment solubility parameter, and the sediment content in sea  
 344 ice (in that order). Below, we describe our choice of criteria and approaches for tuning  
 345 these parameters, and compare the parameter values with observations.

346 The sinking rate sets the background oMn (and through reduction, dMn) concen-  
 347 trations in regions far away from sources such as deep parts of the Canada Basin. We  
 348 initialized the sinking rate in our model as  $0.4 \text{ m d}^{-1}$  based on a sinking rate derived  
 349 by Roy-Barman (2009) from modelled and measured  $^{230}\text{Th}$  profiles in the interior of the  
 350 Arctic Ocean. With a sinking rate of  $0.4 \text{ m d}^{-1}$ , the deep oMn concentrations in the Canada  
 351 Basin in the model were overestimated. An increased sinking rate of  $0.6 \text{ m d}^{-1}$  gave rea-  
 352 sonable background oMn concentrations. The global model of Mn uses a sinking rate of  
 353  $1 \text{ m d}^{-1}$  up to  $10 \text{ m d}^{-1}$  to account for loss near hydrothermal vents (Van Hulten et al.,  
 354 2017).

355 Our sediment resuspension parameterization incorporates two tuned parameters:  
 356 the tidal erosion rate constant,  $C$ , and solubility parameter,  $\gamma$ . The tidal erosion rate

357 controls the background (below about 100 m) and near-bottom dMn concentrations in  
358 shelf areas, so in our domain predominantly the CAA. With observed dMn profiles in  
359 the CAA, we assessed the tidal erosion constant that best represented dMn in the lower  
360 water column with multiple test model runs. The solubility parameter limits the sed-  
361 iment resuspension rate in shelf regions with high tidal speeds, and the most appropri-  
362 ate value was estimated mainly based on comparing modelled dMn with observations  
363 at stations CAA6 and CAA9 (characterized by strong tidal speeds). The resultant sed-  
364 iment resuspension rates in our model range from 0 to 2808 g m<sup>-2</sup> yr<sup>-1</sup> (average is 95  
365 g m<sup>-2</sup> yr<sup>-1</sup>). Particulate material collected in sediment traps over the Beaufort Shelf  
366 from spring 1987 to 1988 contained total dry weight particle fluxes ranging from 20 to  
367 140 g m<sup>-2</sup> yr<sup>-1</sup> (O’Brien et al., 2006). The largest particle fluxes occurred during the  
368 summer and fall. Our average tuned sediment resuspension rate falls within this range.

369 We tuned the sediment content in sea ice last, as it is the most important param-  
370 eter in our study. This parameter affects the surface dMn concentrations in regions where  
371 sea ice contains a significant proportion of non-local sediments (Fig. 4), so primarily the  
372 Canada Basin. We assessed the representation of surface dMn concentrations at stations  
373 in the Canada Basin after a few years of spin up using several values of the sediment con-  
374 tent in ice parameter. The chosen sediment content in sea ice in the Canada Basin in  
375 our model ranges from 0 to 267 g m<sup>-3</sup> (average is 64 g m<sup>-3</sup>). In observations, the sed-  
376 iment load ranges by several orders of magnitude depending on the location sampled,  
377 the type of ice, and is highly variable year-to-year (see Table S1 for a non-comprehen-  
378 sive list of observed sediment content). In the Beaufort Sea, the observed sediment con-  
379 tent in ice cores ranged from 31 to 593 g m<sup>-3</sup> with an average of 157 g m<sup>-3</sup> (Reimnitz  
380 et al., 1993). Our tuned ice sediment content is smaller, but of a similar order of mag-  
381 nitude.

## 382 2.4 Experimental Design

383 Three numerical experiments were performed with the Mn model, running from  
384 2002 to 2019: the reference and “clean” sea ice cases, and a sensitivity experiment for  
385 the rivers. An additional experiment was performed from 2002 to 2015 to assess the mag-  
386 nitude of the impact of biological uptake and remineralization. The reference run includes  
387 all model components except uptake and remineralization, and uses a lower bound es-  
388 timate of the river contributions (no particle-bound Mn,  $\beta = 0$  in Eqn. 3). The clean

389 sea ice case is the same as the reference run, except that the sea ice does not contain sed-  
390 iment (i.e.  $S_{sed\ ice} = 0$ ). In order to bound the riverine influence, we perform a sen-  
391 sitivity experiment with a distinctly upper bound riverine estimate ( $\beta = 1$  in Eqn. 3),  
392 compared to the lower bound estimate from the reference run. The treatment of river-  
393 ine Mn introduces uncertainties in the model due to the complex estuarine cycling and  
394 the influence of particulate matter on dissolved Mn concentrations. In the “upper bound”  
395 river experiment, we include the contribution from riverine sediments on the Mn con-  
396 centrations in addition to the dissolved Mn.

397 Each experiment is spun up by repeating the year 2002 three times, before start-  
398 ing the full run. The run is considered spun up when the year-to-year change in Mn pro-  
399 files is minimal (Fig. S7). Analysis was performed using Python 3 (<https://anaconda.com>)  
400 within Jupyter Notebooks with the NumPy, Pandas, SciPy, Matplotlib, Seaborn, scikit-  
401 learn, and cmocean packages (Pedregosa et al., 2011; Hunter, 2007; Kluyver et al., 2016;  
402 Oliphant, 2006; The Pandas development team, 2020; Thyng et al., 2016; Virtanen et  
403 al., 2020; Waskom & the Seaborn development team, 2020).

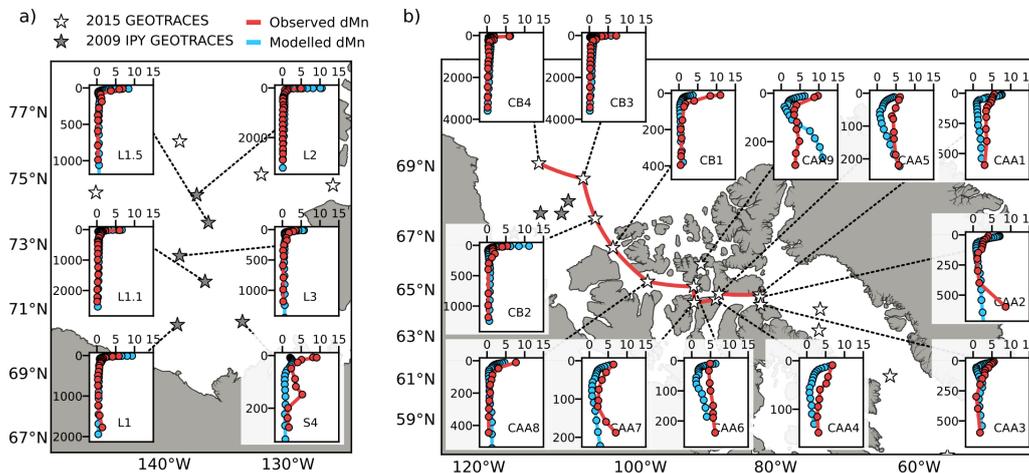
### 404 **3 Results**

405 Mn profiles throughout our domain are typical for a scavenged type element: con-  
406 centrations are higher near sources with a low and homogeneous background (Fig. 5).  
407 The background concentrations are controlled by scavenging, sinking, advection and mix-  
408 ing, and the resultant redistribution of materials throughout the water column, while the  
409 surface Mn maximum is a result of the contributions from river runoff, sea ice melt, dust  
410 deposition, photoreduction, and sediment that is resuspended directly into the polar mixed  
411 layer. Sediment resuspension leads to near-bottom increases in some regions.

#### 412 **3.1 Model Evaluation**

413 We evaluate the Mn model by comparing simulated dissolved Mn concentrations  
414 in August-September 2009 and 2015 from the reference experiment with measurements  
415 collected by the IPY and Canadian GEOTRACES cruises during those time periods (Sim  
416 (2018); Colombo et al. (2020); Fig. 5 and 6). We do not focus on particulate Mn as it  
417 is only incorporated into the model to improve the scavenging of dMn. Nevertheless, mod-  
418 elled oMn displays the observed variability in the upper 100 m in the CAA in 2015 well

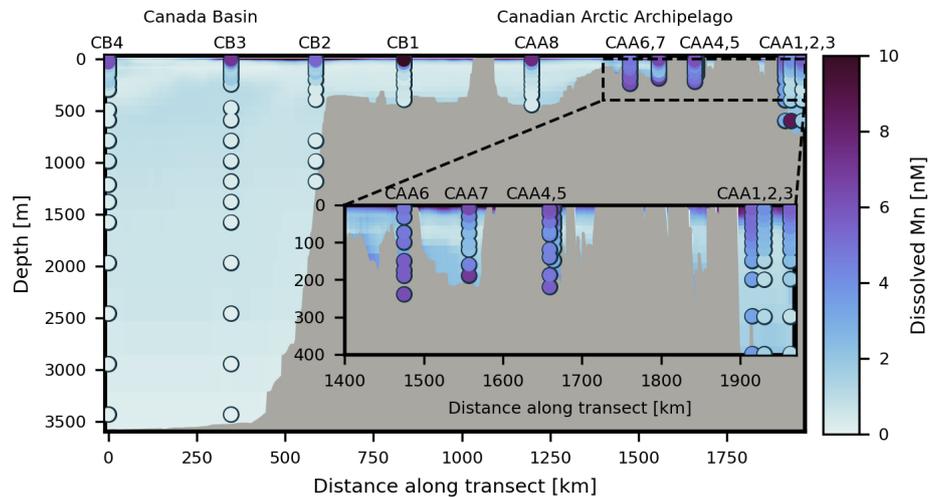
419 (Fig. S8). Overall, our intention is not to replicate the observations, but to incorporate  
 420 all the processes that control Mn distributions and to capture observed spatial variation.  
 421 The observations were not used in initial conditions or boundary conditions to allow for  
 422 an independent evaluation.



**Figure 5.** Simulated dissolved Mn profiles (blue) from the reference run compared to observed concentrations (red) from the (a) 2009 IPY GEOTRACES cruise on the Beaufort Shelf and (b) the 2015 Canadian GEOTRACES cruises in the Canadian Arctic Archipelago. Profiles are labelled with station names and their locations are marked with gray (2009 stations) and white (2015 stations) stars. Simulated concentrations were averaged over the time periods of the cruise observations, i.e. August-September. Note that the profile depth (vertical) scales vary.

423 The model captures the regional variation of Mn concentrations along a transect  
 424 from the deeper Canada Basin into the shallow CAA (Fig. 6). Observed surface concen-  
 425 trations range from 5-10 nM in the Canada Basin and on the Beaufort Shelf, up to 10-  
 426 11 nM at CB1, CAA8, and CAA9, and around 5 nM in the rest of the CAA (Fig. 5 and  
 427 6). The representation of the Canada Basin and the Beaufort shelf surface is variable  
 428 and dependent on the specific patterns of sea ice melt. Surface concentrations are over-  
 429 estimated at L2, L1, and CB2 and underestimated at S4 on the Beaufort shelf, and at  
 430 stations CB4, CB1, and CAA8 in the western CAA which receive outflow from the Canada  
 431 Basin. Within the CAA, surface concentrations are overestimated at stations CAA1 and  
 432 CAA2 in Lancaster Sound where waters from Baffin Bay recirculate, while on the south  
 433 side of the Channel at CAA3, the model captures the surface concentrations. Background

434 concentrations in the model and observations are low (0-2 nM) in the Canada Basin (0-  
 435 900 km along the transect in Fig. 6) and increase (to 1-4 nM) as the waters travel through  
 436 the shelf areas of the CAA.



**Figure 6.** A transect of Mn concentrations from the Canada Basin through Parry Channel in the Canadian Arctic Archipelago to Baffin Bay (path is shown in red in Fig. 5b). The background shading corresponds to simulated Mn averaged over the sampling time period (August-September, 2015) and the circles indicate observed Mn concentrations from the 2015 GEOTRACES cruises. The inset expands on the Parry Channel region east of Barrow Sill.

437 Within Parry Channel, background concentrations west of Barrow Sill are around  
 438 1-2 nM, similar to the Canada Basin, while in the eastern CAA they increase to 3-5 nM  
 439 with near-bottom maxima (appear as a slight bend in the modelled Mn profiles in Fig. 5  
 440 and Fig. 6). Background concentrations in shallow regions are set by the sediment re-  
 441 suspension rate which increases concentrations up to where the surface stratification lim-  
 442 its vertical mixing, while within the polar mixed layer concentrations are set by surface  
 443 sources. At depths of 40-100 m in the CAA, just below the polar mixed layer, the model  
 444 underestimates Mn. Within this depth range, Mn is remineralized, acting as a source  
 445 that is not considered in the reference experiment. In the biological experiment, we es-  
 446 timate that remineralization accounts for up to 0.3 nM (Fig. S15 and Text S3). At 100-  
 447 200 m depth in the Canada Basin and on the Beaufort shelf, observed Mn concentra-  
 448 tions are slightly higher than the background concentrations. This increase is associated

449 with the winter Bering Sea Water and is not captured by the model, as it was not rep-  
450 resented in the model’s western boundary condition.

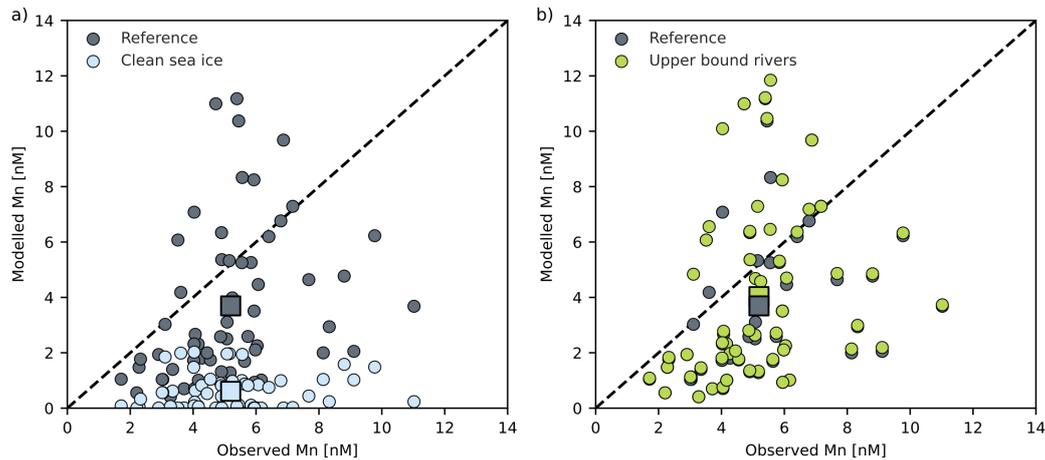
451 The net effect of sediment resuspension is well-represented in the background con-  
452 centrations, however there are a couple of unusual modelled near-bottom Mn profiles (Fig. 5).  
453 At station CAA9 in Penny Strait, the Mn model overestimates background and bottom  
454 concentrations by 5 nM. At this station, strong mixing results in constant, “vertical” ob-  
455 served Mn profiles (Hughes et al., 2018). Sediment resuspension, based on tidal stress,  
456 dominates as a source of Mn to this region. However, this version of the physical model  
457 does not incorporate tides. Hence, we add Mn at the bottom, proportional to the strength  
458 of tidal stress, without redistributing it due to tidal mixing. At stations CAA2 and CAA7,  
459 on the south side of Parry Channel, observed Mn concentrations increase up to 10 nM  
460 near the ocean bottom. These peaks in the observations are attributed to sediment re-  
461 suspension (Colombo et al., 2020), although the specific mechanism for the strong peak  
462 is unclear. The model does not reproduce these local extreme increases, which likely vary  
463 on much smaller spatial scales than our parameterizations can resolve. An increase in  
464 Mn over the 40 m above the bottom is reproduced by the model at stations CAA2, CAA4,  
465 CAA5 and CAA7.

466 While the model is limited in its representation of regions with strongly variable  
467 resuspension rates, it performs well within a range of environments: from deep regions  
468 in the Canada Basin to shallow areas in the CAA. The model is configured to ask ques-  
469 tions about the drivers of Mn variability; it is important to keep in mind that our pa-  
470 rameterizations are limited by the spatial and temporal resolution of available informa-  
471 tion, so small scale variations are unlikely to be captured by the model.

### 472 **3.2 Importance of Sediment in Sea Ice**

473 In order to evaluate the importance of sea ice and rivers on the representation of  
474 Mn in the upper water column (above 50 m), we compare the results of the “clean” sea  
475 ice and upper bound river experiments with the reference experiment (Fig. 7). For all  
476 experiments, the representation of surface concentrations has a broad spread. The “clean”  
477 sea ice experiment underestimates concentrations in the upper water column by several  
478 nM (Fig. 7a) and its mean underestimates concentrations by 4 nM. The mean of the ref-  
479 erence run, with sediment in sea ice, falls within 1 nM of observed concentrations. The

480 upper bound river experiment slightly increases the surface concentrations relative to  
 481 the reference experiment, particularly in the eastern CAA (Fig. 7b). Estimates for sta-  
 482 tions in the Canada Basin are unaffected by the addition of particulate matter in rivers.



**Figure 7.** Nearest-depth modelled Mn concentrations compared with observations for depths shallower than 50 m for all stations (2009 and 2015). Square markers indicate the averages of the experiments and observations. (a) The modelled Mn concentrations at the evaluation stations most closely resemble the observations in the reference experiment with sediment in sea ice compared to the “clean” sea ice experiment. Both of these experiments use the lower bound river estimate. (b) The lower and upper bound river experiments, which include sediment within the sea ice, indicate that additional contribution from riverine particulate matter has a relatively small impact.

483 We expect substantial vertical gradients in concentrations in the surface layer in  
 484 the Arctic Ocean as a result of the strong stratification. The shallowest observations are  
 485 collected at around 10 m below the surface, while the shallowest model estimate is at  
 486 0.5 m depth, so it is difficult to assess the uppermost modelled concentrations. However,  
 487 the Mn-salinity relationship in the model is similar to the observations for the experi-  
 488 ment with sediment in sea ice (Fig. S9). In the “clean” sea ice experiment, the model  
 489 significantly underestimates the low-salinity Mn endmember.

### 3.3 Contributions from External Sources of Mn

To assess the relative contributions of each of the external Mn sources, we calculated the annual contribution and flux from these model components in the reference experiment. We are most interested in the surface layer, so our estimate is for the upper 55 m of the water column. An estimate of the full water column differs by including the effects of resuspension in regions deeper than 55 m, thus increasing the importance of resuspension (Table S2). Estimates from the upper bound river experiment, which does not account for any removal of particulate or dissolved Mn in estuaries, are indicated in brackets. We did not include the contributions from (photo)reduction and remineralization as sources of dMn in these calculations since they are part of the internal cycling of Mn. In order to identify regional differences, we separated the domain into the Canada Basin and the Canadian Arctic Archipelago (details in Fig. S10) and subdivided the CAA into west and east along 100°W. Overall, the Canada Basin is more isolated and receives a lower annual contribution of Mn than the CAA: 238 (254) versus 370 (530)  $\mu\text{mol m}^{-2} \text{yr}^{-1}$  (Table 2).

In our model, the dominant source of Mn in the Canada Basin is the release of sediment by sea ice melt (Table 2); it accounts for 93 (87)% of the average yearly addition of Mn. The amount of melt fluctuates interannually, similar to sea ice area changes observed with satellite data. Nevertheless, from 2002 to 2019, sea ice melt is consistently the largest contributor of Mn in our model in the Canada Basin. Sediment resuspension contributes about 4.7 (4.4)% in the Canada Basin, mainly over the Beaufort shelf, and river discharge, predominantly from the Mackenzie River, contributes 2.2 (8.5)%. Atmospheric dust deposited onto the ocean surface, or released during sea ice melt, is not a significant source of Mn anywhere in the domain.

In the CAA, sediment resuspension contributes 58 (40)% of the annual external addition of Mn to the water column (Table 2). Sediment released by sea ice accounts for 37 (26)% of Mn; a combination of relatively “clean” sea ice with high melt rates. The river contributions cover a broader range from 5.0-34% in the CAA, compared to 2.2-8.5% in the Canada Basin. Since the total annual Mn addition is greater in the CAA, rivers contribute significantly more dMn to the CAA. Although the Canada Basin receives runoff from the Mackenzie River, the CAA has many rivers of a range of sizes that drain into it, including glacial rivers with high characteristic Mn concentrations.

**Table 2.** The spatial average annual dissolved Mn contributed by external model source components to the upper 55 m of the water column ( $\mu\text{mol m}^{-2} \text{yr}^{-1}$ ) in the reference experiment, averaged over the years 2002-2019, separated by region (Fig. S10). Sediment release by sea ice is the only component that varies significantly year-to-year. Estimates from the upper bound river experiment are indicated in brackets.

Component contribution	Canada Basin		Canadian Arctic Archipelago	
	$\mu\text{mol m}^{-2} \text{yr}^{-1}$	%	$\mu\text{mol m}^{-2} \text{yr}^{-1}$	%
River discharge	5.3 (22)	2.2 (8.5)	19 (178)	5.0 (34)
Sediment resuspension	11	4.7 (4.4)	213	58 (40)
Sediment from sea ice	221	93 (87)	138	37 (26)
Dust released by sea ice	0.2	0.1	0.3	0.1
Direct dust deposition	0.0	0.0	0.0	0.0
Total	238 (254)	100	370 (530)	100

**Table 3.** Same as Table 2, but with the Canadian Arctic Archipelago (CAA) subdivided into western and eastern halves along  $100^\circ\text{W}$  (near Barrow Sill).

Component contribution	Western CAA		Eastern CAA	
	$\mu\text{mol m}^{-2} \text{yr}^{-1}$	%	$\mu\text{mol m}^{-2} \text{yr}^{-1}$	%
River discharge	6.5 (28)	2.2 (8.7)	27 (289)	6.5 (42)
Sediment resuspension	155	52 (49)	256	61 (37)
Sediment from sea ice	136	46 (43)	140	33 (20)
Dust released by sea ice	0.3	0.1	0.3	0.1
Direct dust deposition	0.0	0.0	0.0	0.0
Total	297 (318)	100	424 (686)	100

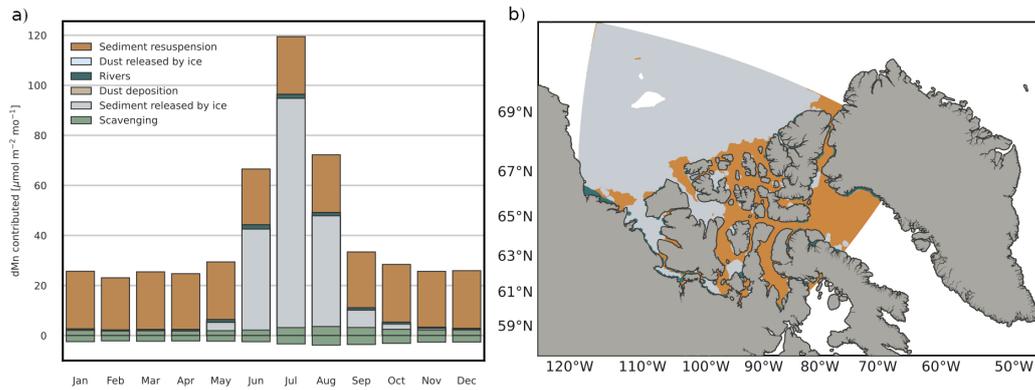
522 Within the CAA, there is a significant difference in dynamical regime west and east  
 523 of the approximately 120 m deep Barrow Sill (Table 3; Hughes et al. (2017); Colombo  
 524 et al. (2020, 2021); Q. Wang et al. (2012)). The overall contribution of Mn to the wa-  
 525 ter column in the eastern CAA is  $424 (686) \mu\text{mol m}^{-2} \text{yr}^{-1}$ , compared to  $297 (318) \mu\text{mol m}^{-2} \text{yr}^{-1}$   
 526 in the west. The main contributor to this difference is the 1.6 times stronger sediment  
 527 resuspension in the eastern CAA. In addition, rivers contribute more strongly to the east-

528 ern CAA relative to the western CAA, 6.5 (42)% versus 2.2 (8.7)%, with a broader range  
 529 in the estimate of their role in the eastern CAA. The eastern CAA receives contributions  
 530 from the high Mn content glacial rivers that drain Greenland, Ellesmere Island, and Baf-  
 531 fin Island.

532 Throughout our domain, Mn concentrations are highest in the summer months as  
 533 a result of seasonally fluctuating components (Fig. 8a). Sea ice melt is largest in July,  
 534 while the river runoff peak occurs during the freshet in May-June. Due to the large sup-  
 535 ply of dissolved Mn in the summer months and the increased solar flux, (photo)reduction  
 536 and oxidation are stronger from July through September. For the month of September,  
 537 we identified which component on average controls Mn for each horizontal grid cell over  
 538 the full time series (Fig. 8b). Note that this figure shows where the model adds the con-  
 539 tribution from a component; where the Mn ends up depends on the advection and dif-  
 540 fusion of the tracer as well.

541 Within the Canada Basin and portions of the western CAA (the Amundsen Gulf  
 542 and western Parry Channel), sea ice melt controls the simulated Mn concentrations (Fig. 8b).  
 543 In the interior of the Beaufort Gyre region, far away from sources and with relatively  
 544 “clean” sea ice, none of the components contribute significantly. Over the Beaufort Shelf,  
 545 the Mackenzie River is a regionally important source of Mn; generally river runoff is a  
 546 significant source at river mouths. In the shallower shelf regions, such as the Beaufort  
 547 Shelf and the CAA, sediment resuspension is prevalent.

548 The magnitudes of annual Mn fluxes from sources in this Arctic Model (AM; Ta-  
 549 ble 2) are comparable to those in the first global model of Mn by Van Hulst et al. (2017)  
 550 (VH). In the global model, dust contributes 0-2  $\mu\text{mol m}^{-2} \text{yr}^{-1}$  in the Arctic Ocean, whereas  
 551 in AM it ranged from 0-0.3  $\mu\text{mol m}^{-2} \text{yr}^{-1}$  (combining direct dust deposition from the  
 552 atmosphere and indirect release from ice). AM riverine fluxes were 5.3 (22)  $\mu\text{mol m}^{-2} \text{yr}^{-1}$   
 553 in the Canada Basin and 19 (178)  $\mu\text{mol m}^{-2} \text{yr}^{-1}$  in the CAA; higher than the VH es-  
 554 timate of 0-2  $\mu\text{mol m}^{-2} \text{yr}^{-1}$ . This range likely reflects a combination of the high Mn  
 555 content of rivers in the Arctic (Colombo et al., 2019) and alternate treatment of rivers;  
 556 VH assumes a relation between Fe and Mn content, while AM uses observations specific  
 557 to the Arctic rivers and their catchment basins. In the global model, the flux of Mn from  
 558 bottom sediments in the Arctic Ocean was 5-75  $\mu\text{mol m}^{-2} \text{yr}^{-1}$ ; AM has 11-213  $\mu\text{mol m}^{-2} \text{yr}^{-1}$ .  
 559 The difference in the upper limit of the range likely reflects the distinctive processes con-



**Figure 8.** Sediment released by sea ice dominates Mn contributions in the Canada Basin and peaks in July, while sediment resuspension is prevalent over shelf areas including the Canadian Arctic Archipelago. (a) Climatology of the seasonal cycle of Mn contributions for the full water column. The oxidation (removal) and reduction (addition) of Mn through scavenging are calculated as the average through the water column. Sediment resuspension is added at the bottom grid cell, while all other sources act directly on the ocean surface. The contributions from dust deposition and release from ice are too small to appear. (b) Most important Mn contributors to the water column in September based on climatology. At each grid cell, the color represents the most important model forcing component. The importance of scavenging is based on the average combined effect of reduction and oxidation throughout the water column. Places within the model domain where the net contributions are smaller than  $0.5 \mu\text{mol m}^{-2} \text{mo}^{-1}$  are white (i.e. in the Canada Basin).

560 sidered by the models: the global model considers sediment diffusion for the flux from  
561 sediments, whereas AM considers sediment resuspension because it is more important  
562 in the CAA (Colombo et al., 2020). It is also challenging to resolve the large continen-  
563 tal shelf regions in the Canadian Arctic in a global model. Lastly, on a global scale, hy-  
564 drothermal input of Mn at spreading ridges is important (Van Hulten et al., 2017), how-  
565 ever the spreading ridges in the Arctic are far away from the AM domain, so that con-  
566 tribution is not included.

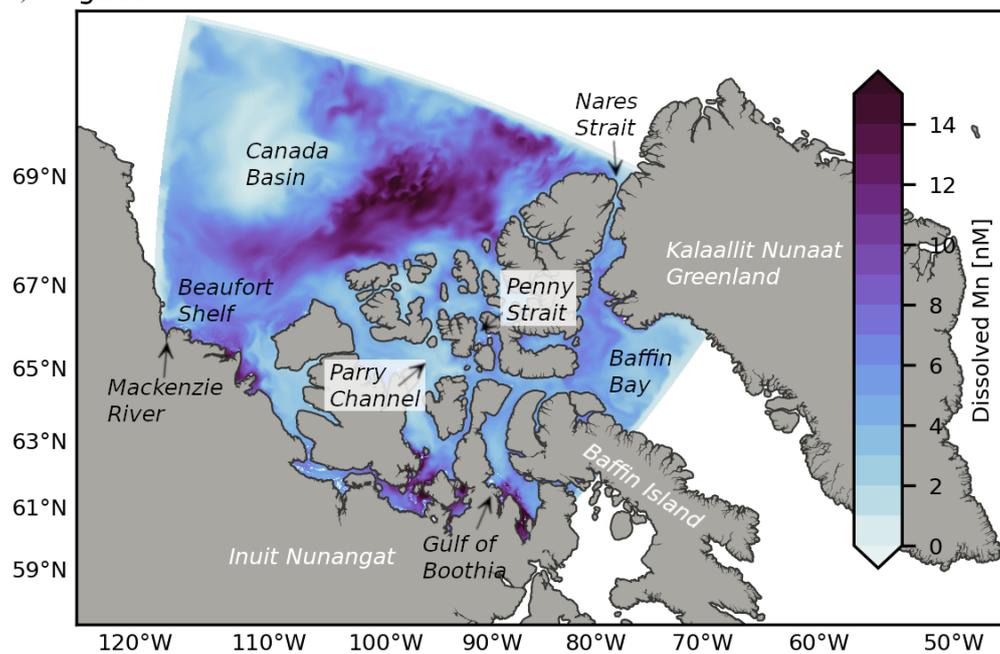
### 567 **3.4 Simulated Surficial Mn During the Summer and the Polar Night**

568 The most significant seasonal and interannual changes in Mn concentrations oc-  
569 cur in the polar mixed layer, defined here as the upper 35 m of the water column. For  
570 the following characterizations of the simulated concentrations, we will focus on this layer.  
571 The upper few meters of the ocean have a strong gradient in Mn concentrations (sim-  
572 ulated profile in Fig. S11). It is not possible to measure this layer with conventional meth-  
573 ods from a large ship. As such, we exclude the surface 3 m in the results presented here  
574 (see Fig. S12 for the surface Mn field) to allow for more direct comparison with exist-  
575 ing observations.

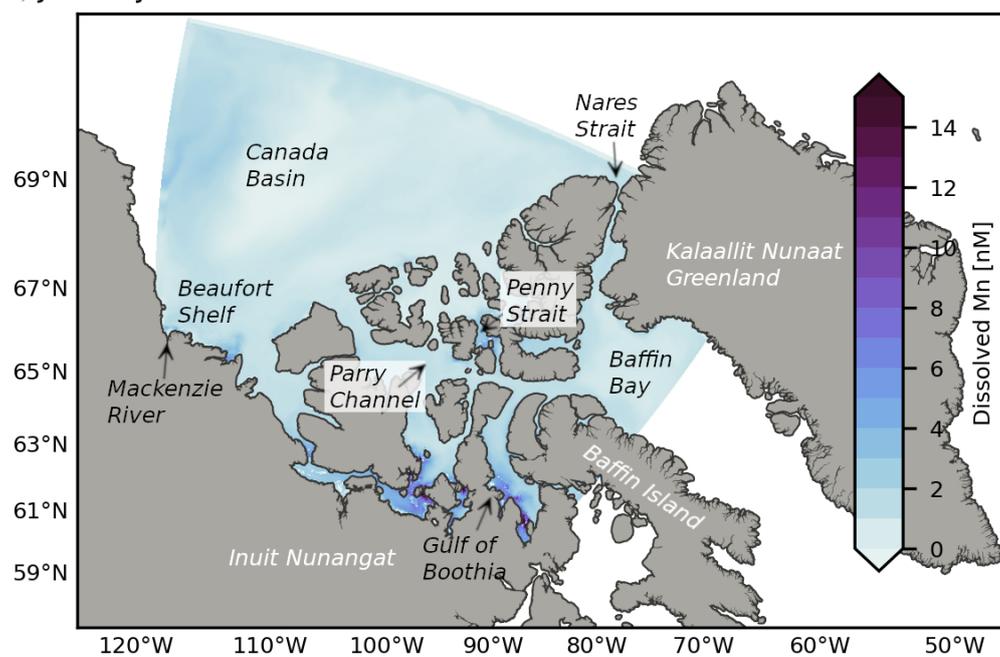
576 During summer months, the surface Mn concentrations in the Canada Basin mir-  
577 ror the areas of strong sea ice melt (Fig. 9a). The highest Mn values are found along the  
578 outer edges of the Beaufort Gyre (up to 14 nM). Although rivers contribute only a few  
579 percent annually to Mn in the Canada Basin (Table 2), over the continental shelf, plumes  
580 of higher Mn concentrations extend along coastlines in the summer, starting during the  
581 spring freshet (Fig. 9a and Fig. 8b). The plume from the Mackenzie River, the largest  
582 river in our domain, extends eastward along the shelf in August, 2015. Glacial drainage  
583 is apparent in surface Mn concentrations in a number of coastal regions (Fig. 9a). Along  
584 the coast of Greenland, high concentration Mn runoff drains the ice sheet and a num-  
585 ber of plumes are visible extending from Nares Strait. In the northern CAA, higher sur-  
586 face concentrations result from a combination of sea ice melt and glacial runoff (Fig. 8b).

587 Mn concentrations exhibit spatial variability within the CAA (Fig. 9). In west-central  
588 CAA, concentrations are low (2-6 nM) and homogeneous. Southern regions, including  
589 the Gulf of Boothia, have some of the highest concentrations (8-14+ nM) and flow into  
590 the Parry Channel east of Barrow Sill. In this section of central and eastern Parry Chan-

## a) August



## b) January



**Figure 9.** Simulated monthly Mn concentrations in the Polar Mixed Layer, excluding the surface three meters to allow for direct comparison with observations (surface fields in Fig. S12). (a) August, 2015. In the summer, sea ice melt and sediment resuspension dominate the Mn concentrations in the Canada Basin and the Canadian Arctic Archipelago, while freshwater sources such as the Mackenzie River and Greenland meltwater are important regionally. (b) January, 2015. During the Polar night, simulated Mn concentrations are homogeneous and low, however sediment resuspension continues to drive higher concentrations in south-central CAA.

591 nel (and Penny Strait), intermediate concentrations (4-8 nM) are present. In Lancaster  
592 Sound, the outflow from Parry Channel follows the southern half of the channel while  
593 waters from Baffin Bay (5-8 nM) recirculate along the northern half of Lancaster Sound.  
594 Baffin Bay is characterized by lower interior surface concentrations and higher bands as-  
595 sociated with Nares Strait and Lancaster Sound outflow.

596 During the Polar Night, fewer sources contribute Mn (Fig. 8a) and there is less spa-  
597 tial contrast in surface concentrations (Fig. 9b). Surface concentrations range from 1-  
598 5 nM (excluding the Gulf of Boothia region), while in the summer they ranged up to 14 nM.  
599 Scavenging has removed the relic of summer surface source signatures. Regions where  
600 Mn is most impacted by sediment resuspension (Fig. 8b), such as the Gulf of Boothia,  
601 still have high concentrations in the winter as this component does not vary seasonally.

## 602 **4 Discussion**

603 In the Arctic Ocean, maximum Mn concentrations occur near the surface in the  
604 polar mixed layer. These high concentrations are commonly attributed to freshwater sources  
605 such as river discharge and sea ice melt (Campbell & Yeats, 1982; Yeats & Westerlund,  
606 1991; Middag et al., 2011b; Cid et al., 2012; Kondo et al., 2016; Colombo et al., 2020).  
607 However, the relative contributions from rivers and sea ice to this low-salinity maximum  
608 are not easily distinguished. In this paper, we present a regional model of Mn in the Cana-  
609 dian Arctic which incorporates river input and sediment release by sea ice, as well as at-  
610 mospheric inputs, sediment resuspension, uptake and remineralization, scavenging, and  
611 sinking. With these components, the model captures the spatial variability and magni-  
612 tude of observed concentrations and we are able to use the model to assess the controls  
613 on Mn in the Arctic. With results from three Mn model experiments (reference, “clean”  
614 sea ice, and upper bound river), we identified the dominance of non-local sediment re-  
615 leased by sea ice in the Canada Basin, while rivers had a more regional importance. These  
616 findings suggest that future changes to sea ice transport across the Arctic Ocean may  
617 have a significant impact on the supply of Mn and other micronutrients to the Canada  
618 Basin and downstream to the CAA. Within the CAA, the dynamical differences between  
619 the western and eastern CAA translated into distinctive mean Mn concentrations and  
620 component contribution patterns with more influence from rivers and sediment resus-  
621 pension.

#### 4.1 Ice-rafted Sediments are the Predominant Source of Mn in the Canada Basin

With our model, we found that 93 (87)% of Mn in the Canada Basin (the main estimate is from the reference experiment with the upper bound river estimate in brackets) is supplied by sediment from sea ice and 37 (26)% in the CAA (Table 2). Sediments released by sea ice melt dominate the Mn concentrations in the polar mixed layer during the summer months (Fig. 9a), while in the winter, sea ice blocks the direct surface input of Mn and a lower, more homogeneous distribution results (Fig. 9b). Sediment transport and release by sea ice is the main source of Mn (and likely other similar nutrients) within the Canada Basin, and plays a role within the CAA as well. The sea ice in the interior of the Canada Basin originates from the Siberian shelf regions and traverses the Arctic Ocean via the transpolar drift. It spends several years in transit, during which it undergoes freeze-thaw cycles and loses some sediment. In our parameterization, the highest Mn concentrations (and relatively younger ice) are found along the outer edges of the Beaufort Gyre in the Canada Basin, while older ice transported to the interior of the Gyre by convergence has lower Mn concentrations (Fig. 9a). Sea ice formed over the Beaufort Shelf is transported towards Siberia and does not directly impact the Mn concentrations in the Canada Basin.

Mn sources from the land-ocean interface, such as rivers and sediments, were more important in the CAA than in the Canada Basin, and dynamical differences between the western and eastern CAA translated into distinctive Mn concentrations and component contribution patterns. This separation in dynamics is bounded by the  $\approx 120$  m deep Barrow Sill and has been noted in several studies (Hughes et al., 2017; Colombo et al., 2020). In the western CAA, surface concentrations range from 2-6 nM (Fig. 9) and Mn component contributions share characteristics with the Canada Basin: similar overall river contributions, significant influence from sediments in sea ice, and weaker contributions from sediment resuspension (Table 3). In contrast, in the eastern CAA, Mn concentrations are higher (4-8 nM; Fig. 9) and dominated by sediment resuspension associated with strong tidal speeds and river discharge play a more important role. The estimate of the component contributions is most sensitive in the eastern CAA: the importance of rivers ranges from 6.5% to 42% depending on the treatment of particulate matter. Rivers are prevalent in the eastern CAA and many of these drain glaciated regions associated with high suspended particulate matter and dissolved Mn. As a result, rivers have the poten-

655 tial to play an important role in the eastern CAA. However, the available information  
656 for river input and estuarine removal, limits our ability to constrain the most likely river  
657 contribution. Based on the surface concentration comparisons (Fig. 7), the upper bound  
658 river experiment alters the mean representation slightly; it is inconclusive on the most  
659 realistic representation. The uncertainties associated with these estimates highlight the  
660 need for studies looking at the estuarine cycling in the CAA.

661 Besides sea ice melt, Pacific water inflow from the Bering Strait and river runoff  
662 (Eurasian runoff and North American runoff) contribute freshwater to the Arctic Ocean  
663 (Proshutinsky et al., 2019; Krishfield et al., 2014) and could contribute Mn to the sur-  
664 face maximum. The central Canada Basin contains significant amounts of meteoric wa-  
665 ter and sea ice melt (Guay et al., 2009) which feed its freshening (Yamamoto-Kawai et  
666 al., 2009). Several studies have looked into the composition of this water. Fichot et al.  
667 (2013) did not identify much river runoff in the central basin and Kelly et al. (2019) found  
668 that the freshwater contribution from Siberian rivers has decreased since 1997 as a re-  
669 sult of the mainly anticyclonic atmospheric circulation pattern over the Canada Basin.  
670 Similarly, model trajectories of floats released from Siberian rivers since 1985 do not gen-  
671 erally reach the Canada Basin by 2007 (Proshutinsky et al., 2019). In our reference and  
672 upper bound river simulations, rivers contribute only 2.2 (8.5)% to the total budget of  
673 Mn in the Canada Basin and 5.0 (34)% in the CAA (Table 2). However, freshwater sources  
674 such as the Mackenzie River on the Beaufort shelf and glacial melt off the coast of Green-  
675 land (Fig. 9a) can dominate areas nearby coastlines. The supply of relatively fresh Pa-  
676 cific Water from Bering Strait to the Canada Basin is also affected by the atmospheric  
677 circulation in the Canada Basin (Kelly et al., 2019) and floats released from Bering Strait  
678 since 2000 do not enter the central Canada Basin by 2012 (Proshutinsky et al., 2019).  
679 Thus, inputs outside of our domain that originate from Siberian runoff and Pacific wa-  
680 ter are unlikely to significantly contribute to the freshwater-associated surface Mn max-  
681 imum in the Canada Basin. It is important to note that our simulated profiles (Fig. 5)  
682 do not capture the subtle increase in Mn concentrations associated with the winter Bering  
683 Sea Water around 100-200 m depth in the Canada Basin and on the Beaufort Shelf. This  
684 limitation is likely because our western boundary condition does not fully capture the  
685 higher concentrations of Mn found in the Alaskan Coastal Current and in waters from  
686 the Chukchi Shelf.

687 In order to assess whether we overestimated the sediment content of sea ice, we per-  
688 formed an experiment with “clean” sea ice. In the “clean” ice experiment, the surface  
689 Mn concentrations are underestimated by  $4 \text{ nmol L}^{-1}$  relative to observations (Fig. 7a).  
690 If we assume that all of the missing Mn comes from sediment and that Mn added at the  
691 surface mixes down to the turbocline, we miss a source that supplies 13-213 grams of sed-  
692 iment per squared meter to the surface ocean across the Canada Basin (range based on  
693 model turbocline depths in 2015). The magnitude of this component is similar to the  
694 average sediment load measured in sea ice cores (Reimnitz et al., 1993; Stierle & Eicken,  
695 2002; Eicken et al., 2005). Rivers would be unable to contribute the total amount miss-  
696 ing since it must occur over a large area and since the upper bound river experiment shows  
697 that additional contributions from rivers do not significantly affect the Canada Basin or  
698 the overall surface representation (Fig. 7b, Table 2). In the “clean” sea ice experiment,  
699 the freshwater endmember of Mn is also underestimated (Fig. S9). The Mn-salinity re-  
700 lationship in the Canada Basin and the CAA is more accurately represented in the ex-  
701 periment with sediment contained in sea ice and the regional differences are also repro-  
702 duced.

703 Our results demonstrate that the long range transport of sediments by sea ice from  
704 the Siberian shelves is an important source of Mn in the Canada Basin and the Cana-  
705 dian Arctic Archipelago. These findings provide support for the sea ice trace metal trans-  
706 port mechanism proposed by Measures (1999). Measures (1999) found that the highest  
707 Al and Fe concentrations in the central Arctic Ocean coincided with areas with high con-  
708 centrations of ice-rafted sediments, instead of river input, and so they hypothesized that  
709 transport of ice rafted sediments and the subsequent seasonal melt supplies reactive el-  
710 ements to the surface Arctic Ocean. However, their data set did not allow the quantifi-  
711 cation of annual fluxes of material to the central Arctic Ocean and so they were unable  
712 to quantify the exact contribution of this component to the observed trace metal con-  
713 centrations.

#### 714 **4.2 Declining Long Range Sea Ice Transport Could Reduce the Canada** 715 **Basin and Canadian Arctic Archipelago Nutrient Supply**

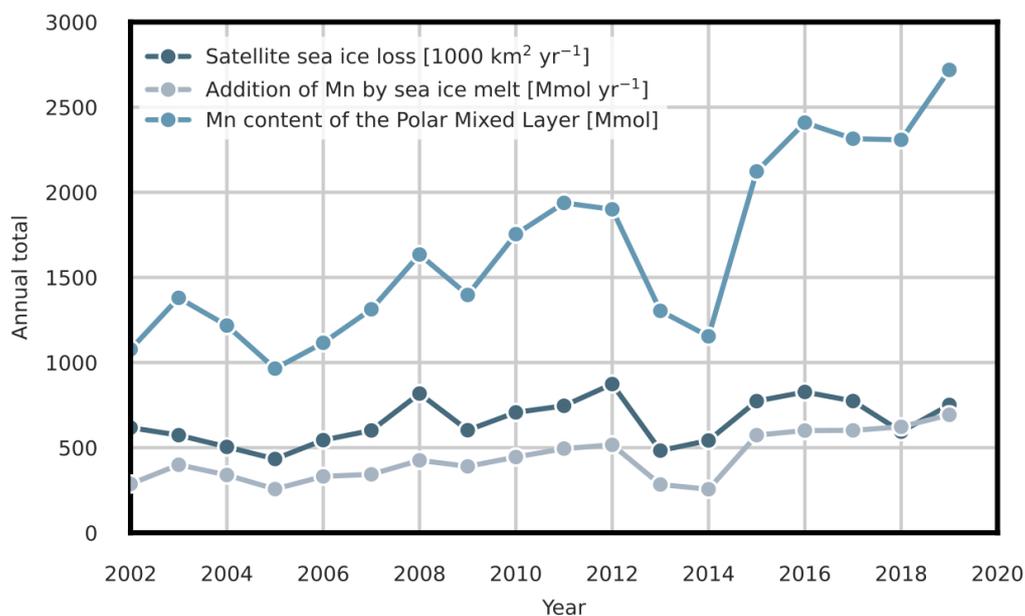
716 Based on the importance of non-local sediments transported by sea ice (particu-  
717 larly from the Siberian shelves), the distributions of trace metals, nutrients, and their  
718 biogeochemical cycles in the Arctic basins are likely to be significantly impacted by cli-

719 mate change associated reductions in sea ice. Rising oceanic and atmospheric temper-  
720 atures delay the freeze-up period and induce earlier melt of sea ice (Stroeve et al., 2012;  
721 Stroeve & Notz, 2018). In addition, in the relatively “quiet” dynamics of the Arctic Ocean,  
722 increased mixing may bring warmer Atlantic water (or Pacific Water; Kodaira et al. (2020))  
723 to the surface and further increase sea ice melt (D’Asaro & Morison, 1992; Liang & Losch,  
724 2018). These factors may significantly reduce the amount of first-year ice that survives  
725 in the Kara Sea, East Siberian sea, and western Laptev Sea (Krumpfen et al., 2019).

726 Studies of the transpolar ice drift indicate an increase in drift speed associated with  
727 a thinning ice cover and as a result, an increase in exchange of ice-rafted material be-  
728 tween regions (Spren et al., 2011; Kwok et al., 2013; Newton et al., 2017; Kipp et al.,  
729 2018). However, in recent years, summer ice extents have been small enough in the marginal  
730 ice zones, that most of the ice exported from shelves melts before it enters the transpo-  
731 lar drift (Krumpfen et al., 2019). These findings suggest a reduction in the transport of  
732 matter towards the central Arctic Ocean and Fram Strait by the transpolar ice drift.

733 In our study, we saw a steady increase in the Mn content of the Canada Basin po-  
734 lar mixed layer from 2002-2019 (Fig. 10), and the primary source of this Mn is sea ice  
735 melt (correlation R-squared of 0.96). Note that our experiments do not account for in-  
736 terannual changes in sea ice supply regions. The addition of Mn by melt in our model  
737 mirrors satellite observations of sea ice loss in the Beaufort Sea (Fig. 10; correlation R-  
738 squared of 0.54). Whereas in the short term, there may be an increase in nutrients sup-  
739 plied by sea ice into the Canada Basin through increased sea ice exchange and melt vol-  
740 ume, in the long run, we expect a decrease in supply of sediment rich sea ice from the  
741 Siberian shelves via the transpolar drift and a subsequent decline in the surface max-  
742 imum of Mn in the Canada Basin. Confounding this is the likely increase in transport  
743 of riverine and shelf-derived trace elements in the ocean by the transpolar drift as a re-  
744 sult of an intensification of the Arctic hydrological cycle and permafrost degradation (Charette  
745 et al., 2020).

746 A reduction in micronutrient supply to the Canada Basin may also have an impact  
747 downstream in Baffin Bay. With our experiments, we calculated the transport of Mn through  
748 Parry Channel and the contribution of sediment released by sea ice melt to this trans-  
749 port (see Text S2 for details). About 87% of the net Mn transported into Parry Chan-  
750 nel from the Canada Basin is contributed by sediments from sea ice (Fig. S13 and S14).



**Figure 10.** Interannual variations in sea ice melt contribute strongly to Mn supply to the Canada Basin. Conversely, surface Mn concentration changes in the Canada Basin are an indicator of the volume of sediments released by sea ice melt. Sea ice loss is calculated from regional monthly sea ice area changes in the Beaufort Sea measured by the Defense Meteorological Satellite Program series of passive microwave remote sensing instruments (Fetterer et al., 2017). The regional Mn model presented in this study is used to calculate Mn added by sea ice melt and the total Mn content of the Canada Basin.

751 Sea ice contributes around 34% of net Mn transported from Parry Channel into Baffin  
752 Bay. The reduction in the contribution of these components does not indicate loss in the  
753 CAA; it reflects the additional contributions from other sources (mainly sediment resus-  
754 pension) in the CAA. The sea ice contribution in the water column is significant down-  
755 stream, however, it is important to note that the sea ice transport in the CAA in the ocean-  
756 ice model is stronger than observed due to the lack of a land-fast ice parameterization  
757 (Grivault et al., 2018). As a result, we may overestimate the sea ice transport and thus  
758 melt in Parry Channel, particularly for the outflow from Parry Channel into Baffin Bay.  
759 There are also further factors contributing Mn within the CAA which confound this find-  
760 ing. The acceleration of the hydrological cycle and permafrost thaw may increase the  
761 contributions of riverine Mn to the CAA; our experiments do not take these changes into  
762 account. On the other hand, sea ice melt is associated with an increase in stratification  
763 which may reduce the depth up to which resuspended sediment can mix, reducing the  
764 Mn supplied into the upper water column (and productive areas) by sediment resuspen-  
765 sion in the CAA. However, reduced sea ice cover is also associated with increased wind-  
766 driven mixing.

767 Our findings for Mn in the Arctic have implications for nutrients which share sim-  
768 ilar sources. In the Arctic Ocean, iron (Fe) behaves similarly to Mn, although Fe is less  
769 soluble than Mn and oxidizes more rapidly (Landing and Bruland (1987); Colombo et  
770 al. (2020); for a comprehensive discussion, read Jensen et al. (2020)). Fe is an essential  
771 micronutrient and in some regions of the ocean, such as the Southern Ocean, parts of  
772 the North Atlantic, and the Pacific Northwest, it limits primary productivity (Martin  
773 & Gordon, 1988; Hawkings et al., 2014; Tagliabue et al., 2017). Generally, iron is not  
774 growth limiting in the Arctic (S. Wang et al., 2014), but there is evidence that Fe is lim-  
775 ited in some specific regions: on the outer shelf and shelf break in the Bering Sea (Aguilar-  
776 Islas et al., 2008), as well as in the Barents Sea and Nansen Basin (Rijkenberg et al., 2018).  
777 Past studies have indicated that sea ice contributes to the flux of Fe into the ocean (Measures,  
778 1999; Lannuzel et al., 2007; Aguilar-Islas et al., 2008; Kanna et al., 2020). Based on the  
779 expected changes to the Mn cycle and supply with sea ice melt over the next decades,  
780 the supply of Fe to the Canada Basin may be reduced as well. Meanwhile, the increase  
781 in simulated Mn content in the Canada Basin from 2002-2019 due to sea ice melt may  
782 also have supplied micronutrients such as Fe and have driven some of the observed in-  
783 creased Arctic Ocean primary production (Lewis et al., 2020). Changes to Fe availabil-

784 ity impact the community composition and the timing of the spring phytoplankton bloom  
785 (Aguilar-Islas et al., 2008), which in turn has consequences for biological productivity,  
786 Arctic ecosystems, and the carbon cycle.

### 787 **4.3 Limitations of Results**

#### 788 *4.3.1 Mn model evaluation*

789 The upper 100 m of the water column are most important to the key findings of  
790 this study. In this zone, the Mn representation is impacted by local sources, photo-enhanced  
791 reduction, and the physical model's salinity representation and associated mixing. Be-  
792 low, we discuss differences between the model and observations and identify the impacts  
793 on our findings.

794 In the CAA, the model underestimates Mn in the subsurface (upper 50 m) result-  
795 ing in a strong vertical gradient of Mn, particularly in the central sills region. The phys-  
796 ical model represents salinity well within the CAA, however the upper 20 m are slightly  
797 too fresh, possibly because of an overestimate in the freshwater transport due to too-mobile  
798 sea ice without a land-fast sea ice parameterization (Grivault et al., 2018). Despite this,  
799 the Mn-salinity relationship matches observations closely in the reference experiment (Fig. S9a).  
800 Increased photo-enhanced reduction could increase subsurface Mn, however, trials with  
801 a non-linear coupling between light penetration and sea ice concentration did not sig-  
802 nificantly affect the subsurface Mn concentrations. Further, while Mn oxides (oMn) are  
803 only modelled for their impact on dMn, oMn concentrations are fairly well-represented  
804 within the upper water column in the CAA (Fig. S8). Remineralization of Mn taken up  
805 by phytoplankton may also counteract some of the subsurface underestimation. How-  
806 ever, we estimated that uptake and remineralization altered dissolved Mn profiles by only  
807 up to 0.3 nM (Fig. S15 and Text S3). Hence, we suggest that the subsurface Mn under-  
808 estimation is most likely caused by improper distribution of materials in the upper wa-  
809 ter column from weaker mixing. Replicating the effect of stronger mixing by redistribut-  
810 ing the Mn, the average Mn concentrations are underestimated by 1 nM in the upper  
811 50 m, while in the subsurface alone they are underestimated by 3 nM. A similar argu-  
812 ment can be made for the near-bottom overestimation of Mn at CAA9; a region known  
813 to have strong tidal mixing. If we redistribute the Mn throughout the water column, the

814 modelled concentration is overestimated by 2 nM, compared to 5 nM for the lower wa-  
815 ter column alone.

816 In the Canada Basin, the physical model captures the depth of isohalines reason-  
817 ably well, however the amount of freshwater in the upper water column is underestimated  
818 (Hu et al., 2019). This underestimation may be due to a lower freshwater state in the  
819 initial conditions derived from the GLORYS2v3 product, but can more likely be attributed  
820 to overestimated sea ice concentration and thickness (and so, underestimated melt in the  
821 model). Despite this, the model represents the overall circulation and characteristics of  
822 the Canada Basin. For Mn, this shortcoming complicates the evaluation of the Mn-salinity  
823 relationship in the Canada Basin (Fig. S9) and instead, we focused our evaluation on Mn  
824 with depth (Fig. 5). The underestimation of sea ice melt does not change the actual com-  
825 ponent contributions estimated by the Mn model: the net effect of the sea ice compo-  
826 nent is a combination of ice melt and sediment content, so an increase in melt would be  
827 counterbalanced by a decrease in sediment richness. The exact spatial variability and  
828 content of sediment in sea ice of the forcing field is a rough first order estimate, never-  
829 theless it is able to provide us with an estimate of the magnitude of the sediment in sea  
830 ice component.

#### 831 **4.3.2 Parameterizations**

832 The findings in this study are limited by the parameterizations for scavenging, sed-  
833 iment in sea ice, sediment resuspension, and river runoff. Overall, the model is best con-  
834 strained for summer months, the southern CAA, and the Canada Basin due to the avail-  
835 ability of observations. Scavenging rates are important throughout the water column and  
836 are most likely to affect our results in coastal regions. We assumed steady state to es-  
837 timate the adsorption and desorption rates from observations; this assumption is least  
838 likely to hold in coastal regions and near the surface where scavenging rates are both im-  
839 portant and variable. For the sediment released by sea ice, we did not account for vari-  
840 ations in transport of sediment (and its origin) across the Arctic Ocean over the course  
841 of the time series. Sea ice drift patterns vary interannually, and so could the source re-  
842 gions for sediment transported to the Canada Basin by sea ice. The sediment content  
843 would more accurately be represented as a time dependent variable. The total Mn con-  
844 tent in the Canada Basin would increase (decrease) with a higher (lower) sediment con-  
845 tent in sea ice, while sediment in sea ice would be more (less) important overall. How-

846 ever, observed sediment sea ice loads range several orders of magnitude by location sam-  
847 pled and properties of the ice, and these fluctuations make it challenging to quantify an-  
848 nual changes in overall sediment content and path travelled. Similarly, sediment resus-  
849 pension varies interannually and seasonally and may be better represented as a time de-  
850 pendent variable.

851 We do not take into account the contributions from breaking of internal waves, storm  
852 generated currents, and surface waves on sediment resuspension and coastal erosion. As  
853 a result, we likely underestimate sediment resuspension contributions in some areas, par-  
854 ticularly during the summer ice-free period. Our treatment of rivers was simplistic and  
855 did not account for the complexity of transformations that occur in the estuarine zone.  
856 Our results indicate a lower and upper bound of the river contributions, however we are  
857 unable to indicate what the actual contribution is. We also did not account for the pro-  
858 jected seasonal ranges in riverine Mn concentrations with discharge (Colombo et al., 2019);  
859 the river discharge varies seasonally, but we hold the characteristic Mn concentrations  
860 of the rivers constant. This approximation could underestimate the riverine contribu-  
861 tions during the spring freshet in coastal areas and is most likely to impact the north-  
862 ern CAA and Greenland coast, where glacial rivers are most important.

863 While the numbers presented here should be taken as an estimate of magnitude  
864 rather than as exact values, the key results are robust to the uncertainties described above.  
865 The only way we were able to close the Mn budget (particularly in the Canada Basin)  
866 was by incorporating the sediment in sea ice component. Similarly, the only way to rep-  
867 resent the higher concentrations of Mn found in the lower water column at some stations  
868 in the CAA was through the sediment resuspension term. While the Mn model presented  
869 here is limited in its representation of these processes, it provides a platform to ask ques-  
870 tions about the drivers of Mn variability and to perform larger scale estimates of the pro-  
871 cesses that contribute Mn to the Arctic Ocean. Improvements to the estimates of sed-  
872 iment content in sea ice from, for example, satellite products would strengthen future  
873 predictions, while the model accuracy would be improved by more comprehensive esti-  
874 mates of the scavenging and sediment resuspension rates. Observations of Mn along a  
875 transect from an estuary into the ocean would help constrain the riverine contributions.

## 5 Conclusions

New trace metal datasets collected in the Arctic Ocean as part of the Canadian GEO-TRACES program have provided an essential base for studying biogeochemical cycling in this unique region. Using in situ observations from Colombo et al. (2020), we developed the first model of Mn in the Canadian Arctic Archipelago and the Canada Basin. With three experiments from 2002-2019, we looked at (1) the drivers of Mn distributions in the CAA and the Canada Basin and (2) implications of future sea ice transport changes on the biogeochemical cycles of nutrients in the Arctic Ocean.

(1) While sediment transport by sea ice is identified as important in the Arctic Ocean (Measures, 1999; Eicken et al., 2005), this mechanism is commonly considered less significant for Mn than riverine input. However, without the contribution from sediment in sea ice to Mn, we were unable to accurately represent the Mn concentrations in the Canada Basin with our model. Sediments transported in sea ice by the transpolar drift account for up to 93% of the total annual Mn added in the Canada Basin and up to 37% in the CAA, driving Mn surface maxima. These results support the hypothesis that “ice-rafted sediment may be an important transport mechanism for supplying reactive trace elements,” proposed by Measures (1999). Rivers are certainly locally important, but contribute only 2.2 (8.5)% annually in the Canada Basin. Within the CAA, our estimates for river contributions ranged from 5.0% up to 34% in the upper bound river experiment. This broad range is the result of the limited information available regarding estuarine cycling in the Arctic. A clear divide is present in the CAA: west of Barrow Sill, the mean concentrations are lower and the behaviour of Mn is more similar to the Canada Basin, while in the eastern CAA, sediments resuspended by high tidal speeds, as well as many glacial rivers drive higher Mn concentrations.

(2) Sea ice transport via the transpolar drift is interrupted by Arctic warming (Krumen et al., 2019) and the decline in this long range transport could reduce the Canada Basin and the CAA nutrient supply. These changes not only impact the Arctic, but also sub-arctic seas, with up to 34% of the Mn transported from Parry Channel into Baffin Bay added by sea ice melt. Mn behaves similarly to Fe in the Arctic Ocean and both of these micronutrients support phytoplankton growth. The importance of sea ice for nutrient supply to the photic zone in the Canada Basin, as well as downstream, is concerning given the recent changes in the Arctic Ocean sea ice regime (reduced summer min-

908 imum ice extent, ice thinning, reduction in multi-year ice extent, and altered drift paths).  
909 There are many competing factors that will contribute to changes in the biogeochem-  
910 ical cycles; combined model-observation studies are highly valuable to understand the  
911 individual contribution of these factors.

## 912 **Acronyms**

913 **CAA** Canadian Arctic Archipelago

914 **NEMO** Nucleus for European Modelling of the Ocean

915 **ANHA12** Arctic and Northern Hemispheric Atlantic 1/12 degree

916 **LIM2** Louvain-la-Neuve version 2

917 **TOP** Tracers in the Ocean Paradigm

918 **TVD** Total Variance Dissipation scheme

919 **CESM** Community Earth System Model

920 **CAM-Chem** Community Atmosphere Model with Chemistry

## 921 **Acknowledgments**

922 We thank Marco van Hulst for openly sharing his model code and results, Jacqui-Lee  
923 Epstein for extracting the tidal speeds for the sediment resuspension parameterization,  
924 Nadja Steiner and Hakase Hayashida for sharing CanOE model results, and Genevieve  
925 Parton for helpful discussions regarding sediment resuspension. This work was funded  
926 by the Natural Sciences and Engineering Council (NSERC) Climate Change and Atmo-  
927 spheric Research Grant: GEOTRACES (RGPC 433848-12) and VITALS (RGPC 433898),  
928 an NSERC Discovery Grant (RGPIN-2016-03865) to SEA, and by the University of British  
929 Columbia through a four year fellowship to BR. Computing resources were provided by  
930 Compute Canada (RRG 2648 RAC 2019, RRG 2969 RAC 2020, RRG 1541 RAC 2021).  
931 The model configuration, code, results, and analysis code are archived on FRDR at  
932 <https://doi.org/10.20383/102.0388>. Analysis code is also available via Github at  
933 <https://github.com/brogalla/Mn-sea-ice-paper>.

## 934 **References**

935 Aguilar-Islas, A. M., Rember, R. D., Mordy, C. W., & Wu, J. (2008). Sea ice-  
936 derived dissolved iron and its potential influence on the spring algal bloom in

- 937 the Bering Sea. *Geophys. Res. Lett.*, *35*(24). doi: 10.1029/2008GL035736
- 938 Bacon, S., Marshall, A., Holliday, N. P., Aksenov, Y., & Dye, S. R. (2014). Seasonal  
939 variability of the East Greenland Coastal Current. *J. Geophys. Res.-Ocean.*,  
940 *119*(6), 3967–3987. doi: 10.1002/2013JC009279
- 941 Balzer, W. (1982). On the distribution of iron and manganese at the sediment/water  
942 interface: Thermodynamic versus kinetic control. *Geochem. Cosmochim. Acta*,  
943 *46*(7), 1153–1161. doi: 10.1016/0016-7037(82)90001-1
- 944 Bamber, J., Van Den Broeke, M., Ettema, J., Lenaerts, J., & Rignot, E. (2012).  
945 Recent large increases in freshwater fluxes from Greenland into the North  
946 Atlantic. *Geophys. Res. Lett.*, *39*(19). doi: 10.1029/2012GL052552
- 947 Bhatia, M. P., Waterman, S., Burgess, D. O., Williams, P. L., Bundy, R. M., Mel-  
948 lett, T., ... Bertrand, E. M. (2021). Glaciers and Nutrients in the Cana-  
949 dian Arctic Archipelago Marine System. *Global Biogeochem. Cycles*, *35*,  
950 e2021GB006976. doi: 10.1029/2021GB006976
- 951 Bouillon, S., Morales Maqueda, M. A., Legat, V., & Fichet, T. (2009). An elastic-  
952 viscous-plastic sea ice model formulated on Arakawa B and C grids. *Ocean*  
953 *Model.*, *27*(3-4), 174–184. doi: 10.1016/j.ocemod.2009.01.004
- 954 Brand, L. E., Sunda, W. G., & Guillard, R. R. L. (1983). Limitation of marine  
955 phytoplankton reproductive rates by zinc, manganese, and iron. *Limnol.*  
956 *Oceanogr.*, *28*(6), 1182–1198. doi: 10.4319/lo.1983.28.6.1182
- 957 Brown, K. A., Williams, W. J., Carmack, E. C., Fiske, G., François, R., McLen-  
958 nan, D., & Peucker-Ehrenbrink, B. (2020). Geochemistry of small Canadian  
959 Arctic rivers with diverse geological and hydrological settings. *J. Geophys.*  
960 *Res.-Biogeosciences*, *125*(1). doi: 10.1029/2019JG005414
- 961 Bruland, K. W., Donat, J. R., & Hutchins, D. A. (1991). Interactive influences  
962 of bioactive trace metals on biological production in oceanic waters. *Limnol.*  
963 *Oceanogr.*, *36*(8), 1555–1577. doi: 10.4319/lo.1991.36.8.1555
- 964 Bruland, K. W., Oriens, K. J., & Cowen, J. P. (1994). Reactive trace metals in  
965 the stratified central North Pacific. *Geochem. Cosmochim. Acta*, *58*(15), 3171–  
966 3182. doi: 10.1016/0016-7037(94)90044-2
- 967 Campbell, J. A., & Yeats, P. A. (1982). The distribution of manganese, iron, nickel,  
968 copper and cadmium in the waters of Baffin Bay and the Canadian Arctic  
969 Archipelago. *Oceanol. Acta*, *5*(2), 161–168.

- 970 Carrère, L., & Lyard, F. (2003). Modeling the barotropic response of the global  
971 ocean to atmospheric wind and pressure forcing-comparisons with observa-  
972 tions. *Geophys. Res. Lett.*, *30*(6). doi: 10.1029/2002GL016473
- 973 Charette, M. A., Kipp, L. E., Jensen, L. T., Dabrowski, J. S., Whitmore, L. M.,  
974 Fitzsimmons, J. N., ... others (2020). The Transpolar Drift as a source  
975 of riverine and shelf-derived trace elements to the central Arctic Ocean. *J.*  
976 *Geophys. Res.-Ocean.*, *125*(5). doi: 10.1029/2019jc015920
- 977 Charette, M. A., Lam, P. J., Lohan, M. C., Kwon, E. Y., Hatje, V., Jeandel, C., ...  
978 Garcia-Orellana, J. (2016). Coastal ocean and shelf-sea biogeochemical cycling  
979 of trace elements and isotopes: lessons learned from GEOTRACES. *Philos.*  
980 *Trans. Roy. Soc. A*, *374*(2081), 20160076. doi: 10.1098/rsta.2016.0076
- 981 Chelton, D. B., de Szoeke, R. A., Schlax, M. G., El Naggar, K., & Siwertz,  
982 N. (1998). Geographical variability of the first baroclinic Rossby ra-  
983 dius of deformation. *J. Phys. Oceanogr.*, *28*(3), 433–460. doi: 10.1175/  
984 1520-0485(1998)028%3C0433:GVOTFB%3E2.0.CO;2
- 985 Cid, A. P., Nakatsuka, S., & Sohrin, Y. (2012). Stoichiometry among bioactive trace  
986 metals in the Chukchi and Beaufort Seas. *J. Oceanogr.*, *68*(6), 985–1001. doi:  
987 10.1007/s10872-012-0150-8
- 988 Colombo, M., Brown, K. A., De Vera, J., Bergquist, B. A., & Orians, K. J. (2019).  
989 Trace metal geochemistry of remote rivers in the Canadian Arctic Archipelago.  
990 *Chem. Geol.*, *525*, 479–491. doi: 10.1016/j.chemgeo.2019.08.006
- 991 Colombo, M., Jackson, S. L., Cullen, J. T., & Orians, K. J. (2020). Dissolved  
992 iron and manganese in the Canadian Arctic Ocean: on the biogeochemical  
993 processes controlling their distributions. *Geochem. Cosmochim. Acta*, *277*,  
994 150–174. doi: 10.1016/j.gca.2020.03.012
- 995 Colombo, M., Rogalla, B., Li, J., Allen, S. E., Orians, K. J., & Maldonado, M. T.  
996 (2021). Canadian Arctic Archipelago Shelf-Ocean Interactions: A Major  
997 Iron Source to Pacific Derived Waters Transiting to the Atlantic. *Global Bio-*  
998 *geochem. Cycles*, *35*(10), e2021GB007058. doi: 10.1029/2021GB007058
- 999 Dai, A., Qian, T., Trenberth, K. E., & Milliman, J. D. (2009). Changes in continen-  
1000 tal freshwater discharge from 1948 to 2004. *J. Climate*, *22*(10), 2773–2792. doi:  
1001 10.1175/2008JCLI2592.1
- 1002 Damm, E., Bauch, D., Krumpfen, T., Rabe, B., Korhonen, M., Vinogradova, E.,

- 1003 & Uhlig, C. (2018). The Transpolar Drift conveys methane from the  
1004 Siberian Shelf to the central Arctic Ocean. *Sci. Rep.*, 8(1), 1–10. doi:  
1005 10.1038/s41598-018-22801-z
- 1006 Darby, D. A., Myers, W. B., Jakobsson, M., & Rigor, I. (2011). Modern dirty sea ice  
1007 characteristics and sources: The role of anchor ice. *J. Geophys. Res.-Ocean.*,  
1008 116(9). doi: 10.1029/2010JC006675
- 1009 D'Asaro, E. A., & Morison, J. H. (1992). Internal waves and mixing in the Arctic  
1010 Ocean. *Deep Sea Res. Pt. I*, 39(2), S459–S484. doi: 10.1016/S0198-0149(06)  
1011 80016-6
- 1012 Dethleff, D., & Kuhlmann, G. (2009). Entrainment of fine-grained surface deposits  
1013 into new ice in the southwestern Kara Sea, Siberian Arctic. *Cont. Shelf Res.*,  
1014 29(4), 691–701. doi: 10.1016/j.csr.2008.11.009
- 1015 Dethleff, D., & Kuhlmann, G. (2010). Fram Strait sea-ice sediment provinces based  
1016 on silt and clay compositions identify Siberian Kara and Laptev seas as main  
1017 source regions. *Polar Sci.*, 29(3). doi: 10.3402/polar.v29i3.6070
- 1018 Dethleff, D., Rachold, V., Tintelnot, M., & Antonow, M. (2000). Sea-ice transport  
1019 of riverine particles from the Laptev Sea to Fram Strait based on clay mineral  
1020 studies. *Intl. J. Earth Sci.*, 89(3), 496–502. doi: 10.1007/s005310000109
- 1021 Drinkwater, K. F., & Harding, G. C. (2001). Effects of the Hudson Strait outflow on  
1022 the biology of the Labrador Shelf. *Can. J. Fish. Aquat. Sci.*, 58(1), 171–184.  
1023 doi: 10.1139/f00-210
- 1024 Eicken, H., Gradinger, R., Gaylord, A., Mahoney, A., Rigor, I., & Melling, H.  
1025 (2005). Sediment transport by sea ice in the Chukchi and Beaufort Seas:  
1026 Increasing importance due to changing ice conditions? *Deep Sea Res. Pt. II*,  
1027 52, 3281–3302. doi: 10.1016/j.dsr2.2005.10.006
- 1028 Eicken, H., Kolatschek, J., Freitag, J., Lindemann, F., Kassens, H., & Dmitrenko,  
1029 I. (2000). A key source area and constraints on entrainment for basin-scale  
1030 sediment transport by Arctic sea ice. *Geophys. Res. Lett.*, 27(13), 1919–1922.  
1031 doi: 10.1029/1999GL011132
- 1032 Eicken, H., Reimnitz, E., Alexandrov, V., Martin, T., Kassens, H., & Viehoff,  
1033 T. (1997). Sea-ice processes in the Laptev Sea and their importance  
1034 for sediment export. *Cont. Shelf Res.*, 17(2), 205–233. doi: 10.1016/  
1035 S0278-4343(96)00024-6

- 1036 Epstein, J.-L. (2018). *The impact of internal tide mixing parameterizations in*  
1037 *an eddy-permitting model of the Arctic Ocean* (Master's thesis, University of  
1038 British Columbia). doi: 10.14288/1.0365809
- 1039 Evans, L. K., & Nishioka, J. (2018). Quantitative analysis of Fe, Mn and Cd from  
1040 sea ice and seawater in the Chukchi Sea, Arctic Ocean. *Polar Sci.*, *17*, 50–58.  
1041 doi: 10.1016/j.polar.2018.07.002
- 1042 Fetterer, F., Knowles, K., Meier, W. N., Savoie, M., & Windnagel, A. K. (2017).  
1043 Updated daily: Sea ice index, version 3. Boulder, Colorado USA. *NSIDC:*  
1044 *National Snow and Ice Data Center*. doi: 10.7265/N5K072F8
- 1045 Fichefet, T., & Maqueda, M. A. M. (1997). Sensitivity of a global sea ice model to  
1046 the treatment of ice thermodynamics and dynamics. *J. Geophys. Res.-Ocean.*,  
1047 *102*(C6), 12609–12646. doi: 10.1029/97JC00480
- 1048 Fichot, C. G., Kaiser, K., Hooker, S. B., Amon, R. M., Babin, M., Bélanger, S., ...  
1049 Benner, R. (2013). Pan-Arctic distributions of continental runoff in the Arctic  
1050 Ocean. *Sci. Rep.*, *3*(1), 1–6. doi: 10.1038/srep01053
- 1051 Fishwick, M. P., Ussher, S. J., Sedwick, P. N., Lohan, M. C., Worsfold, P. J., Buck,  
1052 K. N., & Church, T. M. (2018). Impact of surface ocean conditions and aerosol  
1053 provenance on the dissolution of aerosol manganese, cobalt, nickel and lead in  
1054 seawater. *Mar. Chem.*, *198*, 28–43. doi: 10.1016/J.MARCHEM.2017.11.003
- 1055 Gent, P. R., Willebrand, J., McDougall, T. J., & McWilliams, J. C. (1995). Param-  
1056 eterizing eddy-induced tracer transport in ocean circulation models. *J. Phys.*  
1057 *Oceangr.*, *25*(4), 463–474. doi: 10.1175/1520-0485(1995)025%3C0463:PEITTI%  
1058 3E2.0.CO;2
- 1059 Granskog, M. A., Kaartokallio, H., & Shirasawa, K. (2003). Nutrient status of Baltic  
1060 Sea ice: Evidence for control by snow-ice formation, ice permeability, and ice  
1061 algae. *J. Geophys. Res.-Ocean.*, *108*(C8). doi: 10.1029/2002jc001386
- 1062 Greene, C. H., & Pershing, A. J. (2007). Climate drives sea change. *Science*,  
1063 *315*(5815), 1084–1085. doi: 10.1126/science.1136495
- 1064 Grivault, N., Hu, X., & Myers, P. G. (2018). Impact of the surface stress on the  
1065 volume and freshwater transport through the Canadian Arctic Archipelago  
1066 from a high-resolution numerical simulation. *J. Geophys. Res.-Ocean.*, *123*(12),  
1067 9038–9060. doi: 10.1029/2018JC013984
- 1068 Guay, C. K. H., McLaughlin, F. A., & Yamamoto-Kawai, M. (2009). Differentiating

- 1069 fluvial components of upper Canada Basin waters on the basis of measure-  
1070 ments of dissolved barium combined with other physical and chemical tracers.  
1071 *J. Geophys. Res.-Ocean.*, *114*(C1). doi: 10.1029/2008JC005099
- 1072 Hawkings, J. R., Wadham, J. L., Tranter, M., Raiswell, R., Benning, L. G.,  
1073 Statham, P. J., ... Telling, J. (2014). Ice sheets as a significant source of  
1074 highly reactive nanoparticulate iron to the oceans. *Nat. Commun.*, *5*(1), 1–8.  
1075 doi: 10.1038/ncomms4929
- 1076 Hayashida, H., Christian, J. R., Holdsworth, A. M., Hu, X., Monahan, A. H.,  
1077 Mortenson, E., ... Steiner, N. S. (2019). CSIB v1 (Canadian Sea-ice Bio-  
1078 geochemistry): a sea-ice biogeochemical model for the NEMO community  
1079 ocean modelling framework. *Geosci. Model Dev.*, *12*(5), 1965–1990. doi:  
1080 10.5194/gmd-12-1965-2019
- 1081 Hölemann, J., Wegener, A., & Schirmacher, M. (1999). Dissolved and particulate  
1082 major and trace elements in newly formed ice from the Laptev Sea (Trans-  
1083 drift III, October 1995). In *Land-ocean systems in the Siberian Arctic* (pp.  
1084 101–111). Springer Berlin Heidelberg. doi: 10.1007/978-3-642-60134-7\_11
- 1085 Holland, M. M., Bailey, D. A., Briegleb, B. P., Light, B., & Hunke, E. (2012). Im-  
1086 proved sea ice shortwave radiation physics in CCSM4: The impact of melt  
1087 ponds and aerosols on Arctic sea ice. *J. Climate*, *25*(5), 1413–1430. doi:  
1088 10.1175/JCLI-D-11-00078.1
- 1089 Hu, X., Myers, P. G., & Lu, Y. (2019). Pacific water pathway in the Arctic Ocean  
1090 and Beaufort Gyre in two simulations with different horizontal resolutions. *J.*  
1091 *Geophys. Res.-Ocean.*, *124*(8), 6414–6432. doi: 10.1029/2019JC015111
- 1092 Hu, X., Sun, J., Chan, T. O., & Myers, P. G. (2018). Thermodynamic and dy-  
1093 namic ice thickness contributions in the Canadian Arctic Archipelago in  
1094 NEMO-LIM2 numerical simulations. *Cryosphere*, *12*, 1233–1247. doi:  
1095 10.5194/tc-12-1233-2018
- 1096 Hughes, K. G., Klymak, J. M., Hu, X., & Myers, P. G. (2017). Water mass  
1097 modification and mixing rates in a 1/12 simulation of the Canadian Arc-  
1098 tic Archipelago. *J. Geophys. Res.-Ocean.*, *122*, 803–820. doi: 10.1002/  
1099 2016JC012235
- 1100 Hughes, K. G., Klymak, J. M., Williams, W. J., & Melling, H. (2018). Tidally  
1101 modulated internal hydraulic flow and energetics in the central Canadian

- 1102 Arctic Archipelago. *J. Geophys. Res.-Ocean.*, *123*(8), 5210–5229. doi:  
1103 10.1029/2018JC013770
- 1104 Hunter, J. D. (2007). Matplotlib: A 2d graphics environment. *Comput. Sci. Eng.*,  
1105 *9*(3), 90–95.
- 1106 Jakobsson, M. (2002). Hypsometry and volume of the Arctic Ocean and  
1107 its constituent seas. *Geochem. Geophys. Geosystems*, *3*(5), 1–18. doi:  
1108 10.1029/2001GC000302
- 1109 Jensen, L. T., Morton, P., Twining, B. S., Heller, M. I., Hatta, M., Measures, C. I.,  
1110 ... Fitzsimmons, J. N. (2020). A comparison of marine Fe and Mn cycling:  
1111 U.S. GEOTRACES GN01 Western Arctic case study. *Geochem. Cosmochim.*  
1112 *Acta*. doi: 10.1016/j.gca.2020.08.006
- 1113 Johnson, K. S., Coale, K. H., Berelson, W. M., & Michael Gordon, R. (1996). On  
1114 the formation of the manganese maximum in the oxygen minimum. *Geochem.*  
1115 *Cosmochim. Acta*, *60*(8), 1291–1299. doi: 10.1016/0016-7037(96)00005-1
- 1116 Kanna, N., Lannuzel, D., van der Merwe, P., & Nishioka, J. (2020). Size fraction-  
1117 ation and bioavailability of iron released from melting sea ice in a subpolar  
1118 marginal sea. *Mar. Chem.*, *221*, 103774. doi: 10.1016/j.marchem.2020.103774
- 1119 Kay, J. E., Deser, C., Phillips, A., Mai, A., Hannay, C., Strand, G., ... Vertenstein,  
1120 M. (2015). The community earth system model (CESM) large ensemble  
1121 project : A community resource for studying climate change in the presence  
1122 of internal climate variability. *B. Am. Meteorol. Soc.*, *96*(8), 1333–1349. doi:  
1123 10.1175/BAMS-D-13-00255.1
- 1124 Kelly, S. J., Proshutinsky, A., Popova, E. K., Aksenov, Y. K., & Yool, A. (2019).  
1125 On the origin of water masses in the Beaufort Gyre. *J. Geophys. Res.-Ocean.*,  
1126 *124*(7), 4696–4709. doi: 10.1029/2019JC015022
- 1127 Kempema, E. W., Reimnitz, E., & Barnes, P. (1989). Sea ice sediment entrainment  
1128 and rafting in the Arctic. *J. Sediment. Petrol.*, *59*(2), 308–317. doi: 10.1306/  
1129 212F8F80-2B24-11D7-8648000102C1865D
- 1130 Kipp, L. E., Charette, M. A., Moore, W. S., Henderson, P. B., & Rigor, I. G.  
1131 (2018). Increased fluxes of shelf-derived materials to the central Arctic Ocean.  
1132 *Sci. Advances*, *4*(1). doi: 10.1126/sciadv.aao1302
- 1133 Klinkhammer, G. P., & Bender, M. L. (1980). The distribution of manganese in  
1134 the Pacific Ocean. *Earth Planet Sc. Lett.*, *46*(3), 361–384. doi: 10.1016/0012

- 1135 -821X(80)90051-5
- 1136 Kluyver, T., Ragan-Kelley, B., Pérez, F., Granger, B., Bussonnier, M., Frederic, J.,  
 1137 ... others (2016). *Jupyter notebooks – a publishing format for reproducible*  
 1138 *computational workflows* (F. Loizides & B. Schmidt, Eds.). IOS Press.
- 1139 Kodaira, T., Waseda, T., Nose, T., & Inoue, J. (2020). Record high Pacific Arctic  
 1140 seawater temperatures and delayed sea ice advance in response to episodic  
 1141 atmospheric blocking. *Sci. Rep.*, *10*(1), 1–12.
- 1142 Kondo, Y., Obata, H., Hioki, N., Ooki, A., Nishino, S., Kikuchi, T., & Kuma, K.  
 1143 (2016). Transport of trace metals (Mn, Fe, Ni, Zn and Cd) in the western  
 1144 Arctic Ocean (Chukchi Sea and Canada Basin) in late summer 2012. *Deep Sea*  
 1145 *Res. Pt. I*, *116*, 236–252. doi: 10.1016/J.DSR.2016.08.010
- 1146 Krachler, M., Zheng, J., Fisher, D., & Shotyk, W. (2005). Analytical procedures for  
 1147 improved trace element detection limits in polar ice from Arctic Canada using  
 1148 ICP-SMS. *Anal. Chim. Acta*, *530*(2), 291–298. doi: 10.1016/j.aca.2004.09.024
- 1149 Krishfield, R. A., Proshutinsky, A., Tateyama, K., Williams, W. J., Carmack, E. C.,  
 1150 McLaughlin, F. A., & Timmermans, M. L. (2014). Deterioration of peren-  
 1151 nial sea ice in the Beaufort Gyre from 2003 to 2012 and its impact on the  
 1152 oceanic freshwater cycle. *J. Geophys. Res.-Ocean.*, *119*(2), 1271–1305. doi:  
 1153 10.1002/2013JC008999
- 1154 Krumpen, T., Belter, H. J., Boetius, A., Damm, E., Haas, C., Hendricks, S., ...  
 1155 Stein, R. (2019). Arctic warming interrupts the Transpolar Drift and affects  
 1156 long-range transport of sea ice and ice-rafted matter. *Sci. Rep.*, *9*(1), 1–9. doi:  
 1157 10.1038/s41598-019-41456-y
- 1158 Kuss, J., & Kremling, K. (1999). Spatial variability of particle associated trace  
 1159 elements in near-surface waters of the North Atlantic (30 N/60 W to 60 N/2  
 1160 W), derived by large volume sampling. *Mar. Chem.*, *68*(1-2), 71–86. doi:  
 1161 10.1016/S0304-4203(99)00066-3
- 1162 Kwok, R., Spreen, G., & Pang, S. (2013). Arctic sea ice circulation and drift speed:  
 1163 Decadal trends and ocean currents. *J. Geophys. Res.-Ocean.*, *118*(5), 2408–  
 1164 2425. doi: 10.1002/jgrc.20191
- 1165 Landing, W. M., & Bruland, K. W. (1987). The contrasting biogeochemistry of iron  
 1166 and manganese in the Pacific Ocean. *Geochem. Cosmochim. Acta*, *51*(1), 29–  
 1167 43. doi: 10.1016/0016-7037(87)90004-4

- 1168 Laney, S. R., Krishfield, R. A., & Toole, J. M. (2017, 9). The euphotic zone under  
1169 Arctic Ocean sea ice: Vertical extents and seasonal trends. *Limnol. Oceanogr.*,  
1170 *62*, 1910–1934. doi: 10.1002/LNO.10543
- 1171 Lange, M., & Van Sebille, E. (2017). Parcels v0.9: Prototyping a Lagrangian ocean  
1172 analysis framework for the petascale age. *Geosci. Model Dev.*, *10*(11), 4175–  
1173 4186. doi: 10.5194/gmd-10-4175-2017
- 1174 Lannuzel, D., Schoemann, V., de Jong, J., & Tison, J.-L. (2007). Distribution and  
1175 biogeochemical behaviour of iron in the East Antarctic sea ice. *Mar. Chem.*,  
1176 *106*(1-2), 18–32. doi: 10.1016/J.MARCHEM.2006.06.010
- 1177 Lavelle, J. W., Cowen, J. P., & Massoth, G. J. (1992). A model for the deposition  
1178 of hydrothermal manganese near ridge crests. *J. Geophys. Res.*, *97*(C5), 7413.  
1179 doi: 10.1029/92JC00406
- 1180 Lévy, M., Estublier, A., & Madec, G. (2001). Choice of an advection scheme for bio-  
1181 geochemical models. *Geophys. Res. Lett.*, *28*(19), 3725–3728. doi: 10.1029/  
1182 2001GL012947
- 1183 Lewis, K. M., van Dijken, G. L., & Arrigo, K. R. (2020). Changes in phytoplankton  
1184 concentration now drive increased Arctic Ocean primary production. *Science*,  
1185 *369*(6500), 198–202. doi: 10.1126/science.aay8380
- 1186 Li, J. (2017). *Particulate trace metals and iron availability to phytoplankton in a*  
1187 *changing Arctic Ocean* (Master’s thesis, University of British Columbia). doi:  
1188 10.14288/1.0348666
- 1189 Liang, X., & Losch, M. (2018). On the effects of increased vertical mixing on the  
1190 Arctic Ocean and sea ice. *J. Geophys. Res.-Ocean.*, *123*(12). doi: 10.1029/  
1191 2018JC014303
- 1192 Macdonald, R. W., & Gobeil, C. (2012). Manganese sources and sinks in the Arc-  
1193 tic Ocean with reference to periodic enrichments in basin sediments. *Aquat.*  
1194 *Geochem.*, *18*(6), 565–591. doi: 10.1007/s10498-011-9149-9
- 1195 Madec, G. (2008). NEMO ocean engine. *Note du Pôle de modélisation, Insti-*  
1196 *tut Pierre-Simon Laplace*, *27*(1288-1619). Retrieved from [https://www.nemo-](https://www.nemo-ocean.eu/wp-content/uploads/NEMO_book.pdf)  
1197 [ocean.eu/wp-content/uploads/NEMO\\_book.pdf](https://www.nemo-ocean.eu/wp-content/uploads/NEMO_book.pdf)
- 1198 Martin, J. H., & Gordon, R. M. (1988). Northeast Pacific iron distributions in rela-  
1199 tion to phytoplankton productivity. *Deep Sea Res. Pt. I*, *35*(2), 177–196. doi:  
1200 10.1016/0198-0149(88)90035-0

- 1201 Masina, S., Storto, A., Ferry, N., Valdivieso, M., Haines, K., Balmaseda, M., ...  
 1202 Parent, L. (2017). An ensemble of eddy-permitting global ocean reanal-  
 1203 yses from the MyOcean project. *Clim. Dynam.*, *49*(3), 813–841. doi:  
 1204 10.1007/s00382-015-2728-5
- 1205 Measures, C. I. (1999). The role of entrained sediments in sea ice in the distribution  
 1206 of aluminium and iron in the surface waters of the Arctic Ocean. *Mar. Chem.*,  
 1207 *68*, 59–70. doi: 10.1016/S0304-4203(99)00065-1
- 1208 Michel, C., Ingram, R. G., & Harris, L. R. (2006, 10). Variability in oceanographic  
 1209 and ecological processes in the canadian arctic archipelago. *Prog. Oceanogr.*,  
 1210 *71*, 379–401. doi: 10.1016/J.POCEAN.2006.09.006
- 1211 Middag, R., de Baar, H. J. W., Laan, P., Cai, P. H., & van Ooijen, J. C. (2011a).  
 1212 Dissolved manganese in the Atlantic sector of the Southern Ocean. *Deep Sea*  
 1213 *Res. Pt. II*, *58*(25-26), 2661–2677. doi: 10.1016/J.DSR2.2010.10.043
- 1214 Middag, R., de Baar, H. J. W., Laan, P., & Klunder, M. B. (2011b). Fluvial and hy-  
 1215 drothermal input of manganese into the Arctic Ocean. *Geochem. Cosmochim.*  
 1216 *Acta*, *75*(9), 2393–2408. doi: 10.1016/J.GCA.2011.02.011
- 1217 Newton, R., Pfirman, S., Tremblay, B., & DeRepentigny, P. (2017). Increasing  
 1218 transnational sea-ice exchange in a changing Arctic Ocean. *Earths Future*,  
 1219 *5*(6), 633–647. doi: 10.1002/2016EF000500
- 1220 Nürnberg, D., Wollenburg, I., Dethleff, D., Eicken, H., Kassens, H., Letzig, T.,  
 1221 & Reimnitz, E. (1994). Sediments in Arctic sea ice: Implications for  
 1222 entrainment, transport and release. *Mar. Geol.*, *119*, 185–214. doi:  
 1223 10.1016/0025-3227(94)90181-3
- 1224 O’Brien, M. C., Macdonald, R. W., Melling, H., & Iseki, K. (2006). Particle  
 1225 fluxes and geochemistry on the Canadian Beaufort Shelf: Implications for  
 1226 sediment transport and deposition. *Cont. Shelf Res.*, *26*(1), 41–81. doi:  
 1227 10.1016/J.CSR.2005.09.007
- 1228 Oliphant, T. E. (2006). *A guide to numpy* (Vol. 1). Trelgol Publishing USA.
- 1229 Pedregosa, F., Varoquaux, G., Gramfort, A., Michel, V., Thirion, B., Grisel, O., ...  
 1230 Duchesnay, E. (2011). Scikit-learn: Machine learning in Python. *J. Mach.*, *12*,  
 1231 2825–2830.
- 1232 Peeken, I., Primpke, S., Beyer, B., Gütermann, J., Katlein, C., Krumpen, T.,  
 1233 ... Gerdtts, G. (2018). Arctic sea ice is an important temporal sink and

- 1234 means of transport for microplastic. *Nat. Commun.*, *9*(1), 1–12. doi:  
1235 10.1038/s41467-018-03825-5
- 1236 Pfirman, S. L., Eicken, H., Bauch, D., & Weeks, W. (1995). The potential transport  
1237 of pollutants by Arctic sea ice. *Sci. Total Environ.*, *159*(2-3), 129–146. doi: 10  
1238 .1016/0048-9697(95)04174-Y
- 1239 Proshutinsky, A., Krishfield, R., Toole, J. M., Timmermans, M. L., Williams, W.,  
1240 Zimmermann, S., . . . Zhao, J. (2019). Analysis of the Beaufort Gyre freshwa-  
1241 ter content in 2003–2018. *J. Geophys. Res.-Ocean.*, *124*(12), 9658–9689. doi:  
1242 10.1029/2019JC015281
- 1243 Reimnitz, E., McCormick, M., McDougall, K., & Brouwers, E. (1993). Sediment ex-  
1244 port by ice rafting from a coastal polynya. *Arct. Antarct. Alp. Res.*, *25*(2), 83–  
1245 98. doi: 10.1080/00040851.1993.12002988
- 1246 Rijkenberg, M. J. A., Slagter, H. A., van der Loeff, M., van Ooijen, J., & Gerringa,  
1247 L. J. A. (2018). Dissolved Fe in the deep and upper Arctic Ocean with a  
1248 focus on Fe limitation in the Nansen Basin. *Front. Mar. Sci.*, *5*, 88. doi:  
1249 10.3389/fmars.2018.00088
- 1250 Roy-Barman, M. (2009). Modelling the effect of boundary scavenging on Thorium  
1251 and Protactinium profiles in the ocean. *Biogeosciences*, *6*, 3091–3107. doi: 10  
1252 .5194/bg-6-3091-2009
- 1253 Sim, N. (2018). *Biogeochemical cycling of dissolved and particulate manganese*  
1254 *in the northeast Pacific and Canadian western Arctic* (Doctoral dissertation,  
1255 University of British Columbia). doi: 10.14288/1.0374222
- 1256 Smith, G. C., Roy, F., Mann, P., Dupont, F., Brasnett, B., Lemieux, J.-F., . . .  
1257 Bélair, S. (2014). A new atmospheric dataset for forcing ice-ocean models:  
1258 Evaluation of reforecasts using the Canadian global deterministic prediction  
1259 system. *Q. J. R. Meteorol. Soc.*, *140*(680), 881–894. doi: 10.1002/qj.2194
- 1260 Spreen, G., Kwok, R., & Menemenlis, D. (2011). Trends in Arctic sea ice drift and  
1261 role of wind forcing: 1992-2009. *Geophys. Res. Lett.*, *38*(19). doi: 10.1029/  
1262 2011GL048970
- 1263 Stierle, A. P., & Eicken, H. (2002). Sediment inclusions in Alaskan coastal  
1264 sea ice: Spatial distribution, interannual variability, and entrainment re-  
1265 quirements. *Arct. Antarct. Alp. Res.*, *34*(4), 465–476. doi: 10.1080/  
1266 15230430.2002.12003518

- 1267 Stroeve, J. C., & Notz, D. (2018). Changing state of Arctic sea ice across all sea-  
 1268 sons. *Environ. Res. Lett.*, *13*(10). doi: 10.1088/1748-9326/aade56
- 1269 Stroeve, J. C., Serreze, M. C., Holland, M. M., Kay, J. E., Malanik, J., & Barrett,  
 1270 A. P. (2012). The Arctic’s rapidly shrinking sea ice cover: A research synthe-  
 1271 sis. *Clim. Change*, *110*(3-4), 1005–1027. doi: 10.1007/s10584-011-0101-1
- 1272 Sunda, W. G., & Huntsman, S. A. (1994). Photoreduction of manganese oxides in  
 1273 seawater. *Mar. Chem.*, *46*(1-2), 133–152. doi: 10.1016/0304-4203(94)90051-5
- 1274 Tagliabue, A., Bowie, A. R., Boyd, P. W., Buck, K. N., Johnson, K. S., & Saito,  
 1275 M. A. (2017). The integral role of iron in ocean biogeochemistry. *Nature*,  
 1276 *543*(7643), 51–59. doi: 10.1038/nature21058
- 1277 The Pandas development team. (2020). Pandas-dev/pandas: Pandas. *Zenodo*, *21*, 1–  
 1278 9.
- 1279 Thyng, K. M., Greene, C. A., Hetland, R. D., Zimmerle, H. M., & DiMarco, S. F.  
 1280 (2016). True colors of oceanography: Guidelines for effective and accurate  
 1281 colormap selection. *Oceanogr.*, *29*(3), 9–13.
- 1282 Tilmes, S., Lamarque, J. F., Emmons, L. K., Kinnison, D. E., Marsh, D., Garcia,  
 1283 R. R., ... Blake, N. (2016). Representation of the Community Earth System  
 1284 Model (CESM1) CAM4-chem within the Chemistry-Climate Model Initiative  
 1285 (CCMI). *Geosci. Model Dev.*, *9*(5), 1853–1890. doi: 10.5194/gmd-9-1853-2016
- 1286 Tovar-Sánchez, A., Duarte, C. M., Alonso, J. C., Lacorte, S., Tauler, R., & Galbán-  
 1287 Malagón, C. (2010). Impacts of metals and nutrients released from melt-  
 1288 ing multiyear Arctic sea ice. *J. Geophys. Res.*, *115*(C7), C07003. doi:  
 1289 10.1029/2009JC005685
- 1290 Tucker, W. B., Gow, A. J., Meese, D. A., Bosworth, H. W., & Reimnitz, E. (1999).  
 1291 Physical characteristics of summer sea ice across the Arctic Ocean. *J. Geophys.*  
 1292 *Res.-Ocean.*, *104*(C1), 1489–1504. doi: 10.1029/98jc02607
- 1293 Van Hulst, M., Middag, R., Dutay, J.-C., De Baar, H., Roy-Barman, M., Gehlen,  
 1294 M., ... Sterl, A. (2017). Manganese in the west Atlantic Ocean in the context  
 1295 of the first global ocean circulation model of manganese. *Biogeosciences*, *14*,  
 1296 1123–1152. doi: 10.5194/bg-14-1123-2017
- 1297 Virtanen, P., Gommers, R., Oliphant, T. E., Haberland, M., Reddy, T., Courn-  
 1298 peau, D., ... others (2020). Scipy 1.0: fundamental algorithms for scientific  
 1299 computing in python. *Nature methods*, *17*(3), 261–272.

- 1300 Wang, Q., Myers, P. G., Hu, X., & Bush, A. B. G. (2012). Flow constraints on  
1301 pathways through the Canadian Arctic Archipelago. *Atmos.-Ocean*, *50*(3),  
1302 373–385. doi: 10.1080/07055900.2012.704348
- 1303 Wang, S., Bailey, D., Lindsay, K., Moore, J. K., & Holland, M. (2014). Impact of sea  
1304 ice on the marine iron cycle and phytoplankton productivity. *Biogeosciences*,  
1305 *11*(17), 4713–4731. doi: 10.5194/bg-11-4713-2014
- 1306 Waskom, M., & the Seaborn development team. (2020). *Seaborn*. Zenodo. doi: 10  
1307 .5281/zenodo.592845
- 1308 Wedepohl, H. K. (1995). The composition of the continental crust. *Geochem. Cos-*  
1309 *mochim. Acta*, *59*(7), 1217–1232. doi: 10.1016/0016-7037(95)00038-2
- 1310 Yamamoto-Kawai, M., McLaughlin, F. A., Carmack, E. C., Nishino, S., Shimada,  
1311 K., & Kurita, N. (2009). Surface freshening of the Canada Basin, 2003–2007:  
1312 River runoff versus sea ice meltwater. *J. Geophys. Res.*, *114*(C1), C00A05. doi:  
1313 10.1029/2008JC005000
- 1314 Yeats, P. A., & Westerlund, S. (1991). Trace metal distributions at an Arctic Ocean  
1315 ice island. *Mar. Chem.*, *33*(3), 261–277. doi: 10.1016/0304-4203(91)90071-4
- 1316 Zalesak, S. T. (1979). Fully multidimensional flux-corrected transport algorithms for  
1317 fluids. *J. Comput. Phys.*, *31*(3), 335–362. doi: 10.1016/0021-9991(79)90051-2